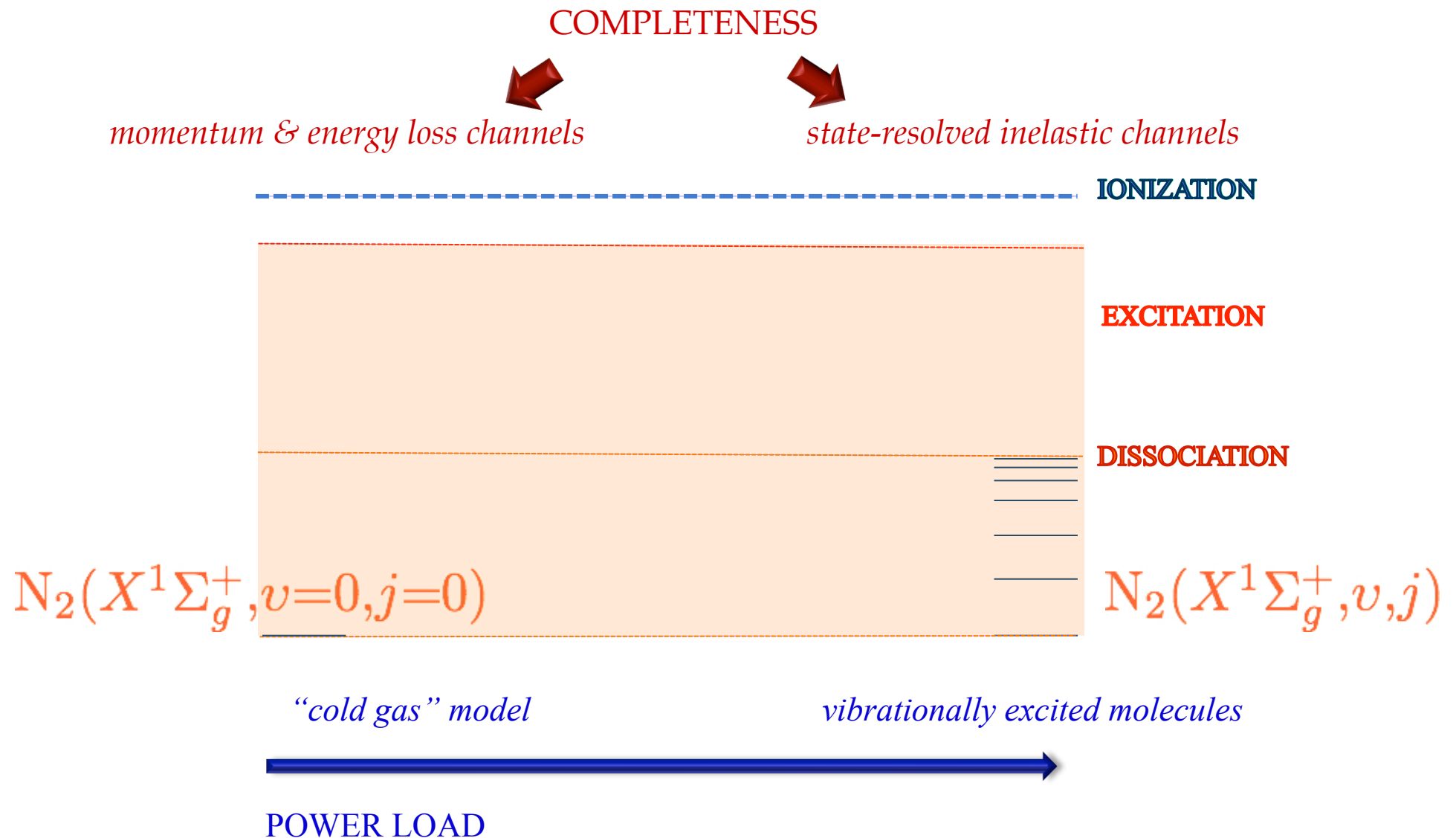


The state-of-the-art description of vibrational-electronic excitations

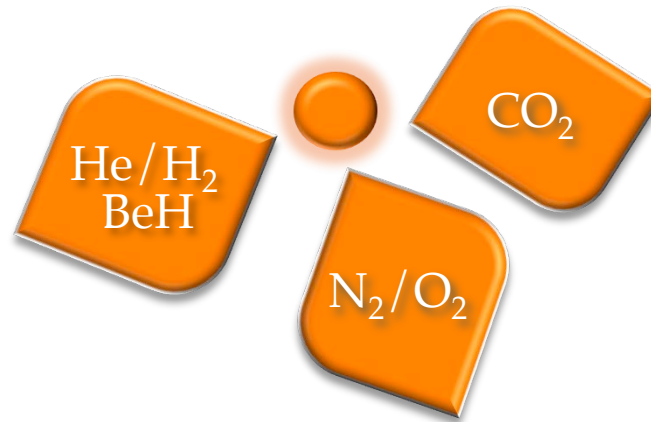
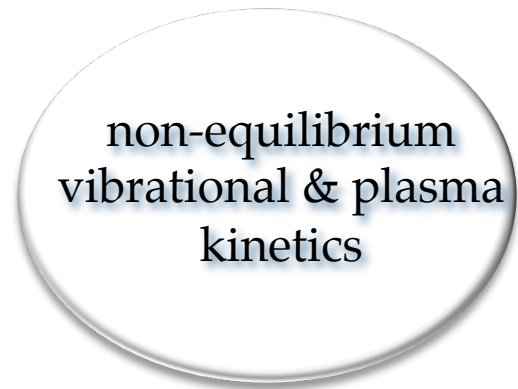
A. Laricchiuta

CNR IMIP Bari (Italy)

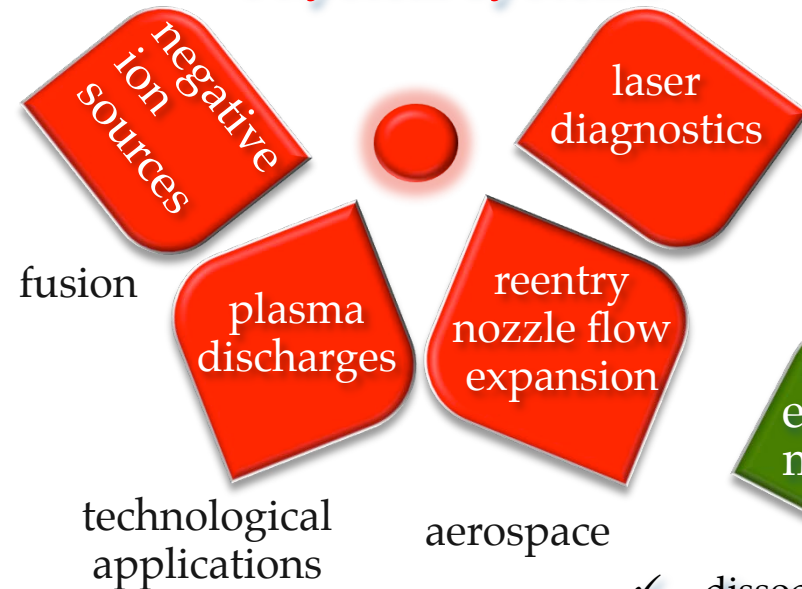


Chemical Species

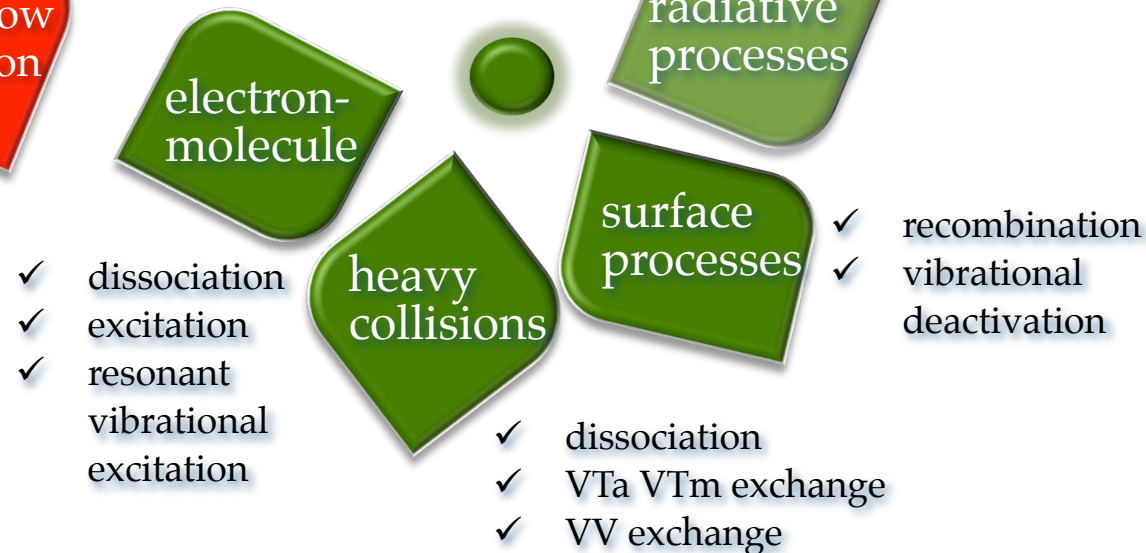
Theoretical approaches



Physical Systems



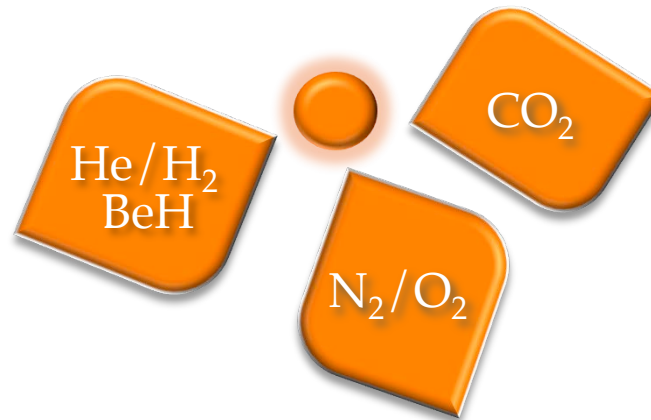
Elementary Processes



Chemical Species

Theoretical approaches

non-equilibrium
vibrational & plasma
kinetics



quantum

- ✓ R-matrix
- ✓ local resonance theory
- ✓ coupled-channels

Physical Systems

semiclassical

quasi-classical

classical & semi-empirical

- ✓ Gryzinskii
- ✓ *fBE* scaled approach
- ✓ similarity

- ✓ IPM
- ✓ semiclassical theory for surface interaction

Elementary Processes

electron-molecule

heavy collisions

radiative processes

surface processes

negative ion sources

fusion

plasma discharges

technological applications

laser diagnostics

reentry nozzle flow expansion

aerospace

ELEMENTARY PROCESS DATA

F Esposito, I Armenise, G Capitta, M Capitelli, **O–O₂ state-to-state vibrational relaxation and dissociation rates based on quasiclassical calculations**, Chemical Physics 351 (2008) 91-98

Panesi, M., Jaffe, R.L., Schwenke, D.W., Magin, T.E., **Rovibrational internal energy transfer and dissociation of N₂(¹Σ_g⁺)-N(⁴S_u) system in hypersonic flows**, J. Chem. Phys. 138 (2013) 044312

Kim, J.G., Kwon, O.J., Park, C., **Non-equilibrium rotation-vibration transitions and chemical reactions in H +H₂ and He+H₂**, AIP Conf Proc 1084 (2009) 813-818, RGD26

A Kurnosov, M Cacciatore, A Laganà, F Pirani, M Bartolomei, E Garcia, **The effect of the intermolecular potential formulation on the state-selected energy exchange rate coefficients in N₂–N₂ collisions**, J. comp. chem. 35 (2014), 722-736

A Lombardi, A Laganà, F Pirani, M Bartolomei, **Carbon dioxide dynamics in earth and planetary atmospheres: vibrational energy transfer in CO₂+ CO₂ collisions**, Virt&l-Comm 1 (2012), pp. 16-17

Cacciatore, M. , Rutigliano, M., **The semiclassical and quantum-classical approaches to elementary surface processes: Dissociative chemisorption and atom recombination on surfaces**, Physica Scripta 78 (2008) 058115

Capitelli, M., Laricchiuta, A., Celiberto, R., Kosarim, A.V., Smirnov, B.M., **Electron-molecule collision cross-sections for air kinetics**, AIP Conf. Proc. 762 (2005) 914-919, (RGD24)

V Laporta, R Celiberto, J Tennyson, **Resonant vibrational-excitation cross sections and rate constants for low-energy electron scattering by molecular oxygen**, Plasma Sources Science and Technology 22 (2013), 025001

DATABASEs



LXCAT DATABASE

<http://fr.lxcat.net/home/>



ALADDIN

Numerical database maintained by
the IAEA Nuclear Data Section A+M Data Unit

ALADDIN IAEA DATABASE

<https://www-amdis.iaea.org/ALADDIN/>

Phys4Entry Database

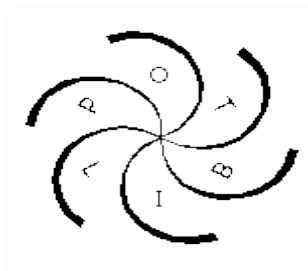
PHYS4ENTRY DATABASE

<http://users.ba.cnr.it/imip/cscpal38/phys4entry/database.html>



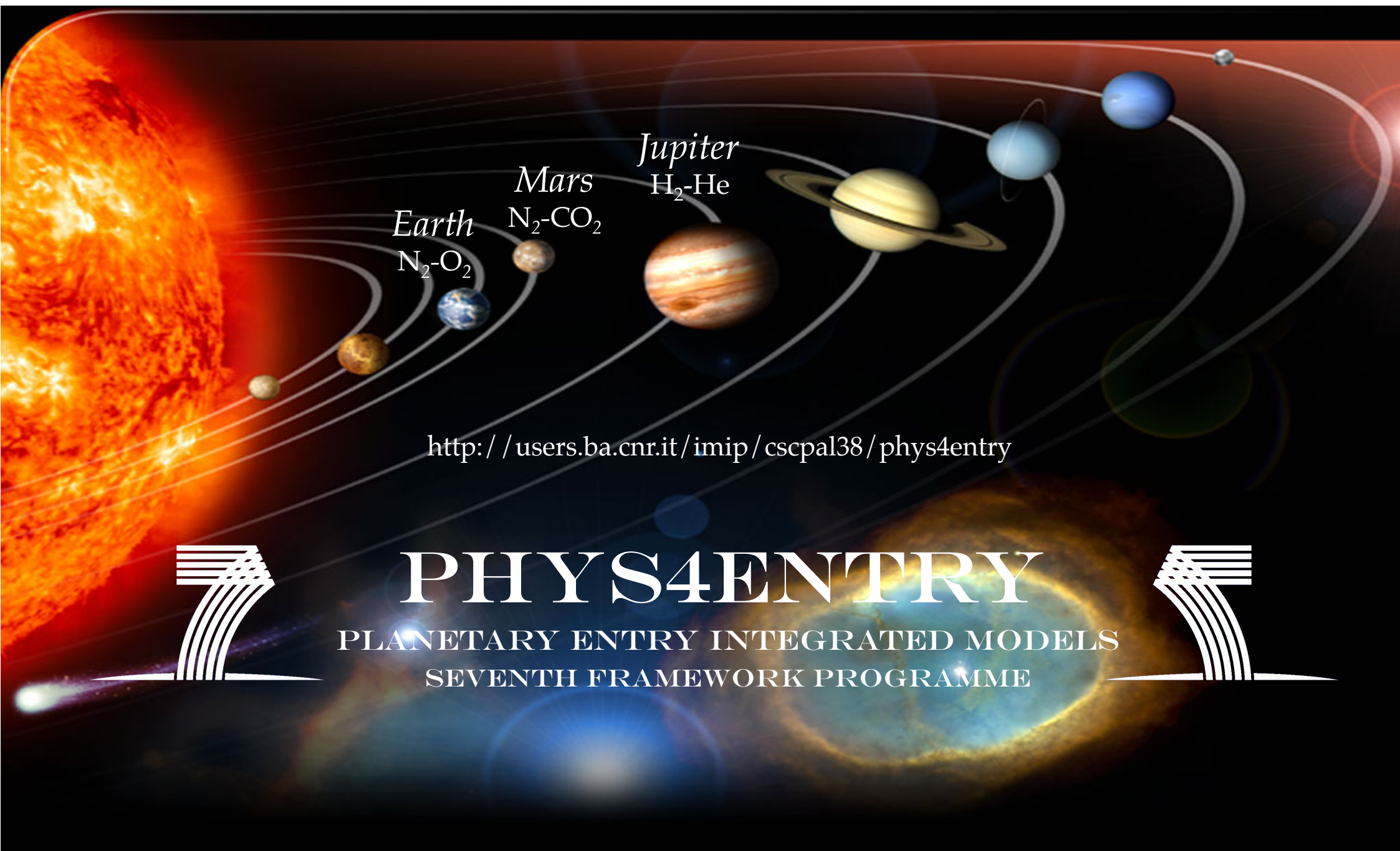
STELLAR DATABASE

<http://esther.ist.utl.pt/pages/stellar.html>



POTLIB DATABASE

<http://comp.chem.umn.edu/potlib>



Earth
 $\text{N}_2\text{-O}_2$

Mars
 $\text{N}_2\text{-CO}_2$

Jupiter
 $\text{H}_2\text{-He}$

<http://users.ba.cnr.it/imip/cscpal38/phys4entry>



PHYS4ENTRY

PLANETARY ENTRY INTEGRATED MODELS
SEVENTH FRAMEWORK PROGRAMME



The Phys4EntryDB

GRANT AGREEMENT n. 242311

THEME 9 SPACE

COLLABORATIVE PROJECT

PHYS4ENTRYDB

state-resolved dynamical information for elementary processes relevant to the state-to-state kinetic modeling of planetary-atmosphere entry conditions

Designed and implemented by CNR IMIP Bari and SER&Practices spin-off of the University of Bari.

THE PHYS4ENTRYDB TEAM

✓ *accuracy*

✓ *completeness*

✓ *reliability*

PHYS4ENTRY database design

MOLECULAR DYNAMICS
COMMUNITY
CONTRIBUTORS

data-insert procedure

- ✓ registration as new contributor
- ✓ file upload
- ✓ selection class and process
- ✓ wizard for the term symbols of chemical species
- ✓ declaration of units
- ✓ database population

DATABASE
INTERFACING
with predefined queries

ADMINISTRATOR

DB management

- ✓ db species-table extension
- ✓ authorizing procedure
- ✓ approval status for new record

KINETIC MODELLING
COMMUNITY
INQUIRERS

data availability

- ✓ registration required for download
- ✓ wizard for building of queries
- ✓ selection of output units
- ✓ download of data files (format)
- ✓ plotting of energy and internal dof dependencies

ENTER THE DB

HOME

PROJECT MANAGER

PARTNER NETWORK

ACTIVITIES

GANTT chart

PERT

MEETINGS

CONTACTS

Download AREA

DOCUMENTS

BUDGET

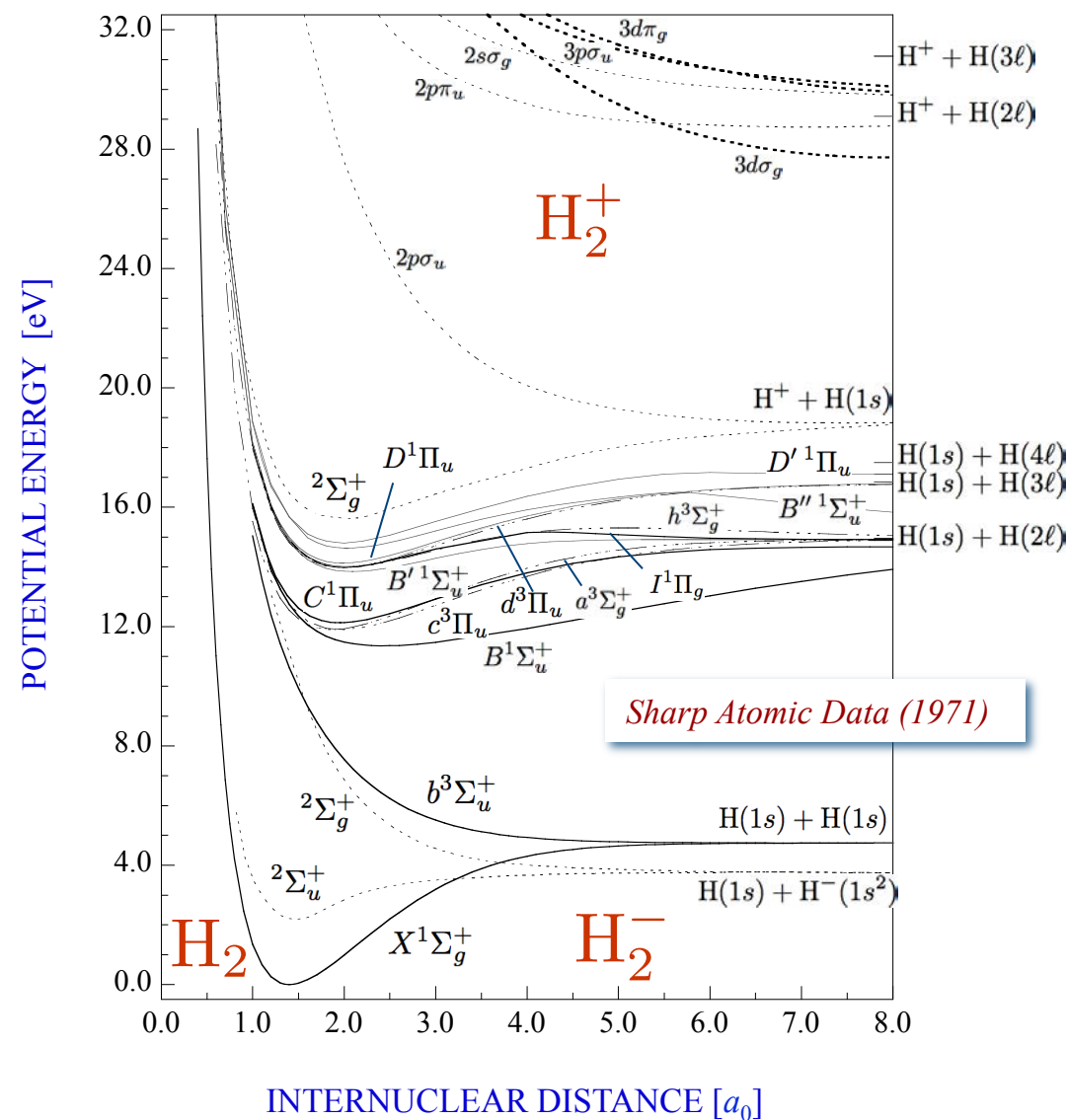
DATABASE

7FP

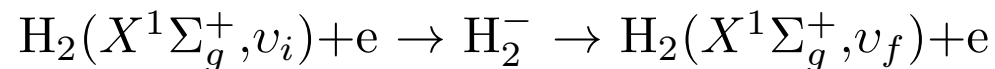
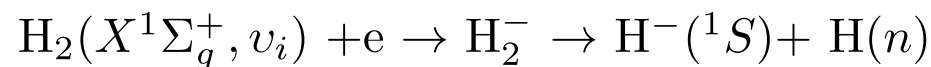
NEWS

contact the webmaster

ELECTRON IMPACT INDUCED PROCESSES: THE DATA SET FOR H₂



RESONANT PROCESSES



Celiberto et al, Chemical Physics (2012)
 Celiberto et al, Physical Review A (2013)
 Horačček et al, Physical Review A (2004) and (2006)

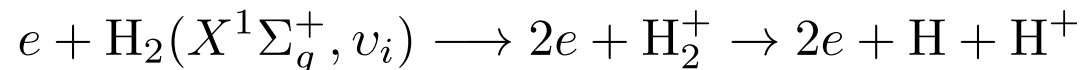
non-RESONANT VIBRONIC EXCITATION



Celiberto et al., AtomicData&NuclearDataTables (2001)
 Laricchiuta et al., Physical Review A (2004)

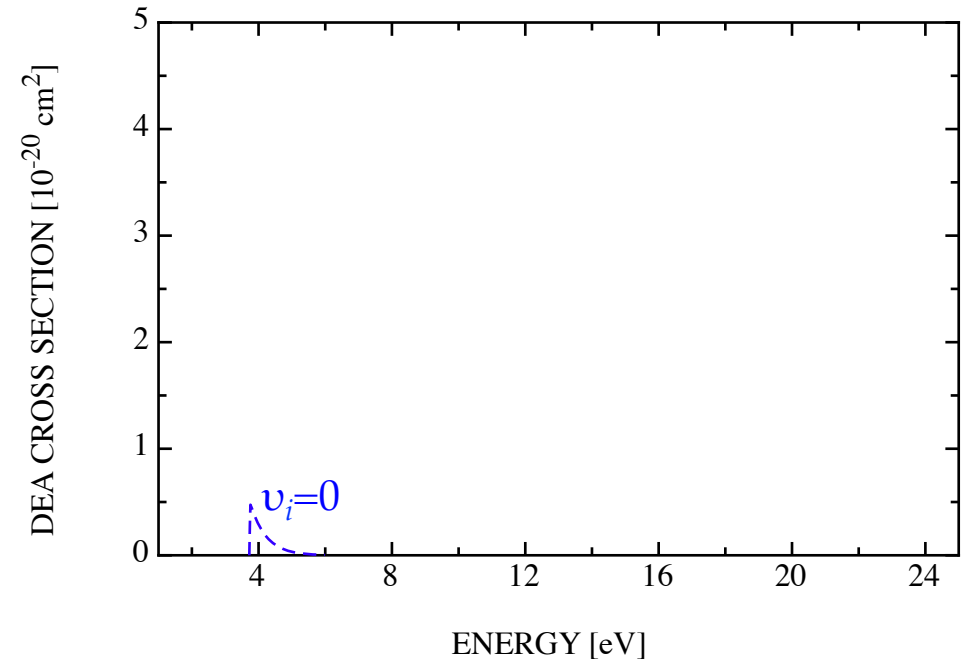
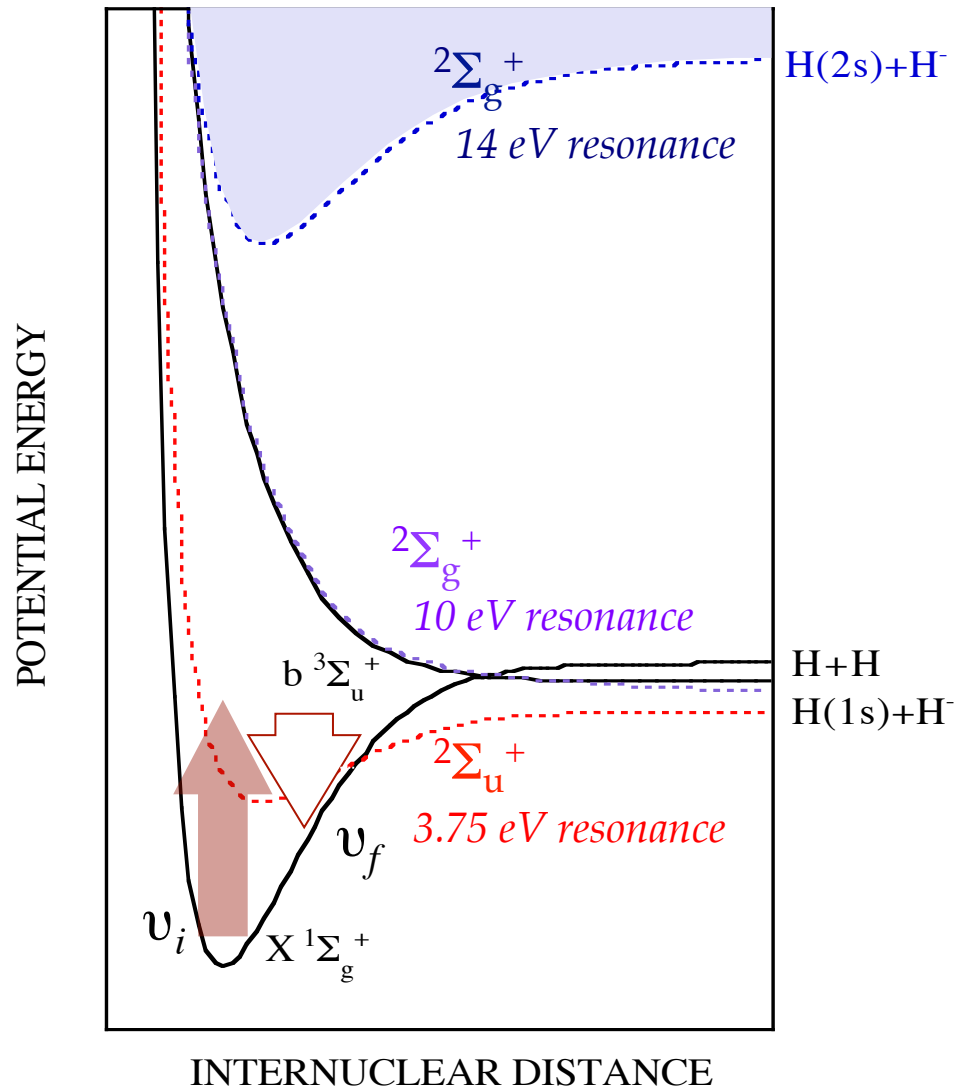
Liu et al J Phys B 45 (2012) 015201

DISSOCIATIVE & non-DISSOCIATIVE IONIZATION



Wunderlich Chemical Physics 390 (2011) 75–82

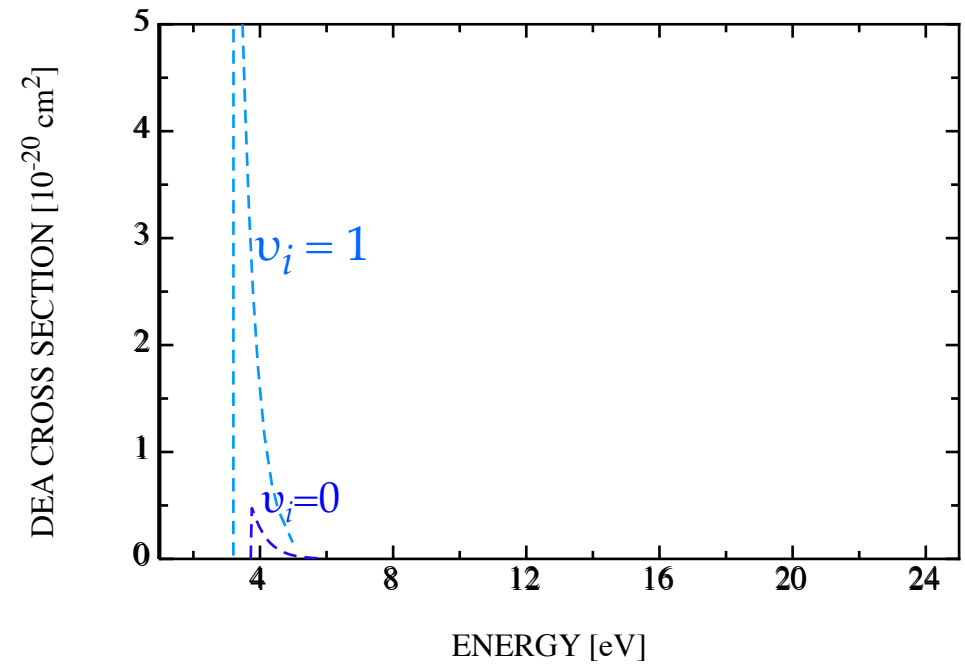
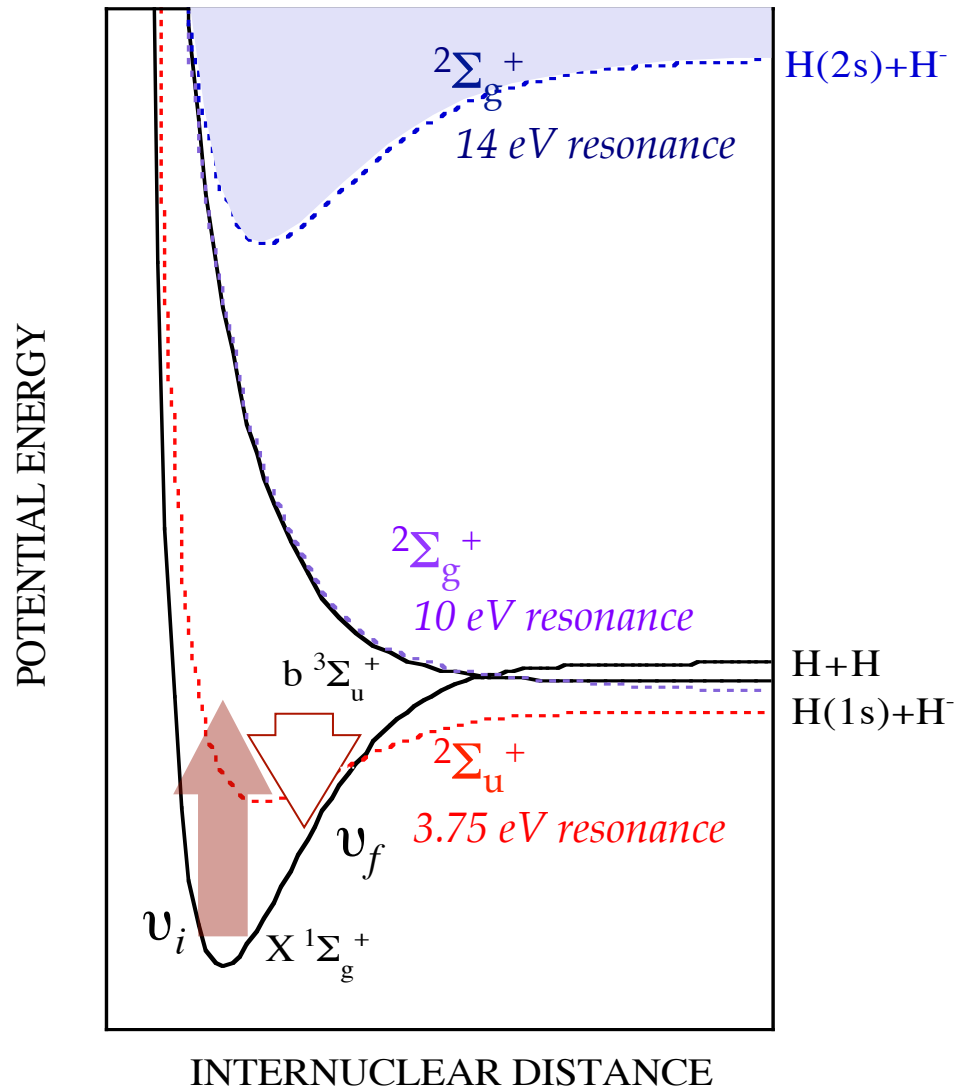
H₂ Dissociative Attachment



Horáček et al, Physical Review A (2004)

3.75 eV resonance

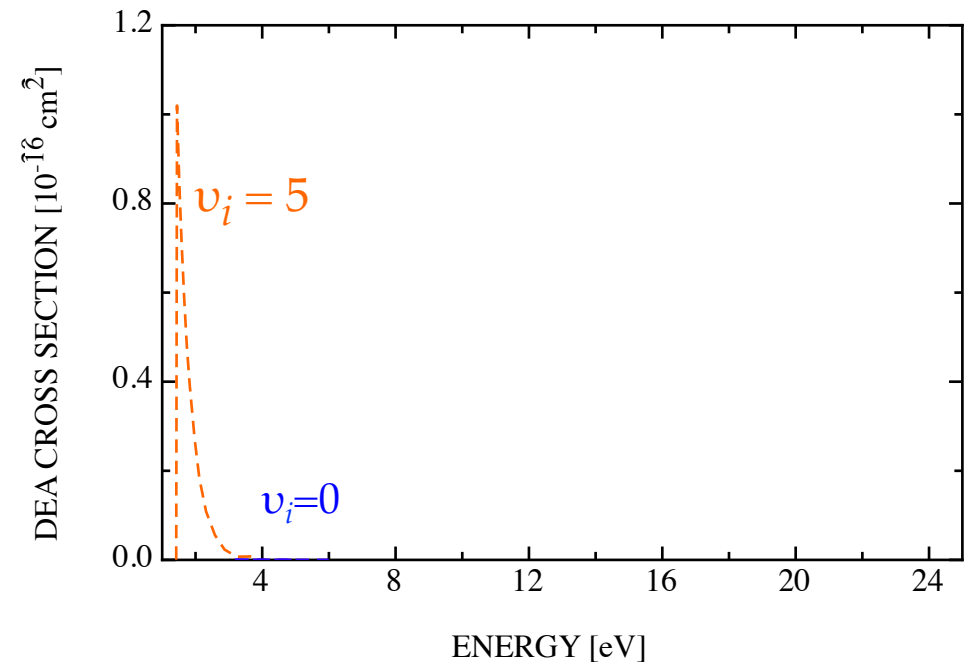
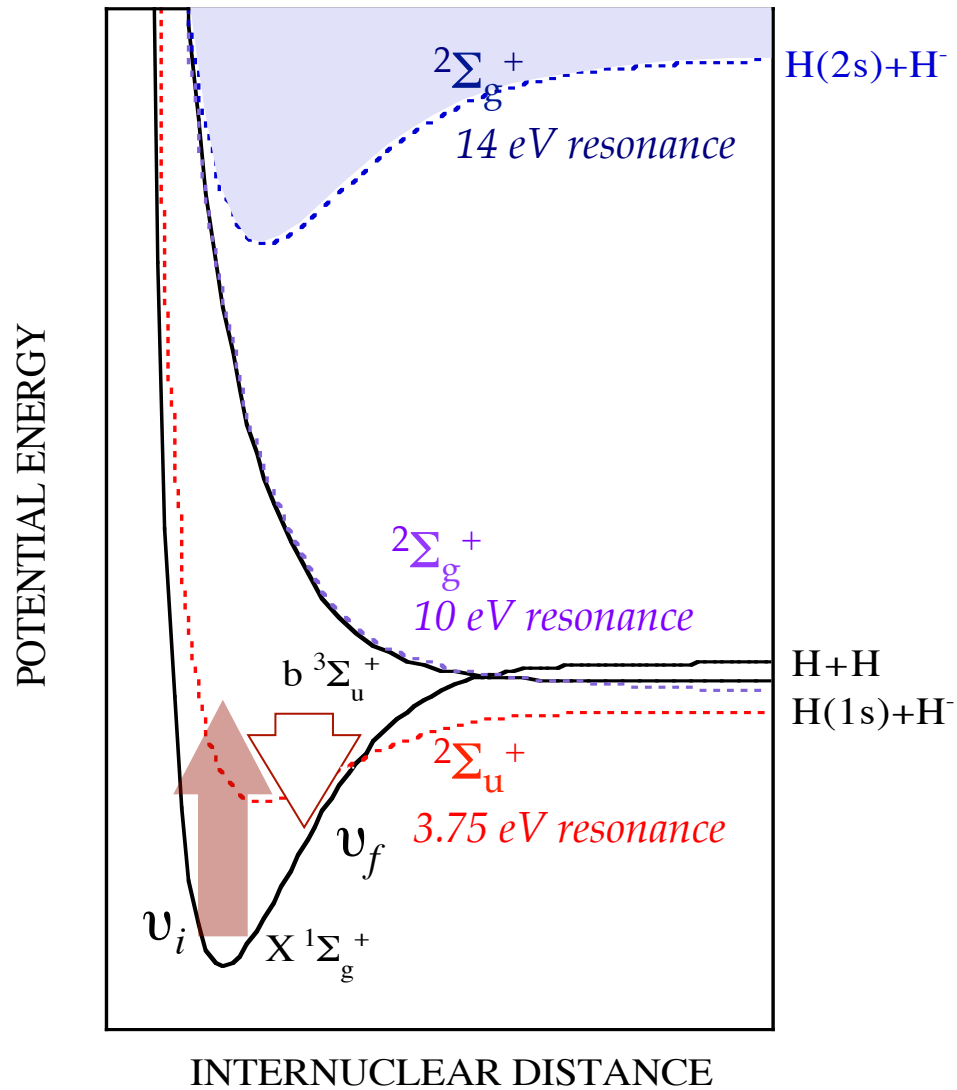
H₂ Dissociative Attachment



Horáček et al, Physical Review A (2004)

3.75 eV resonance

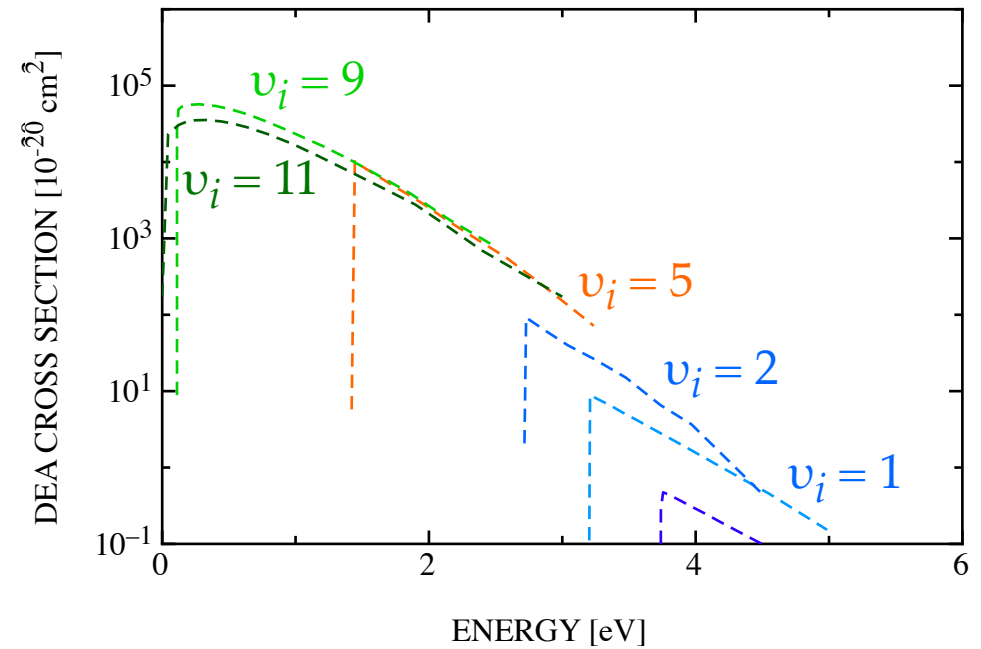
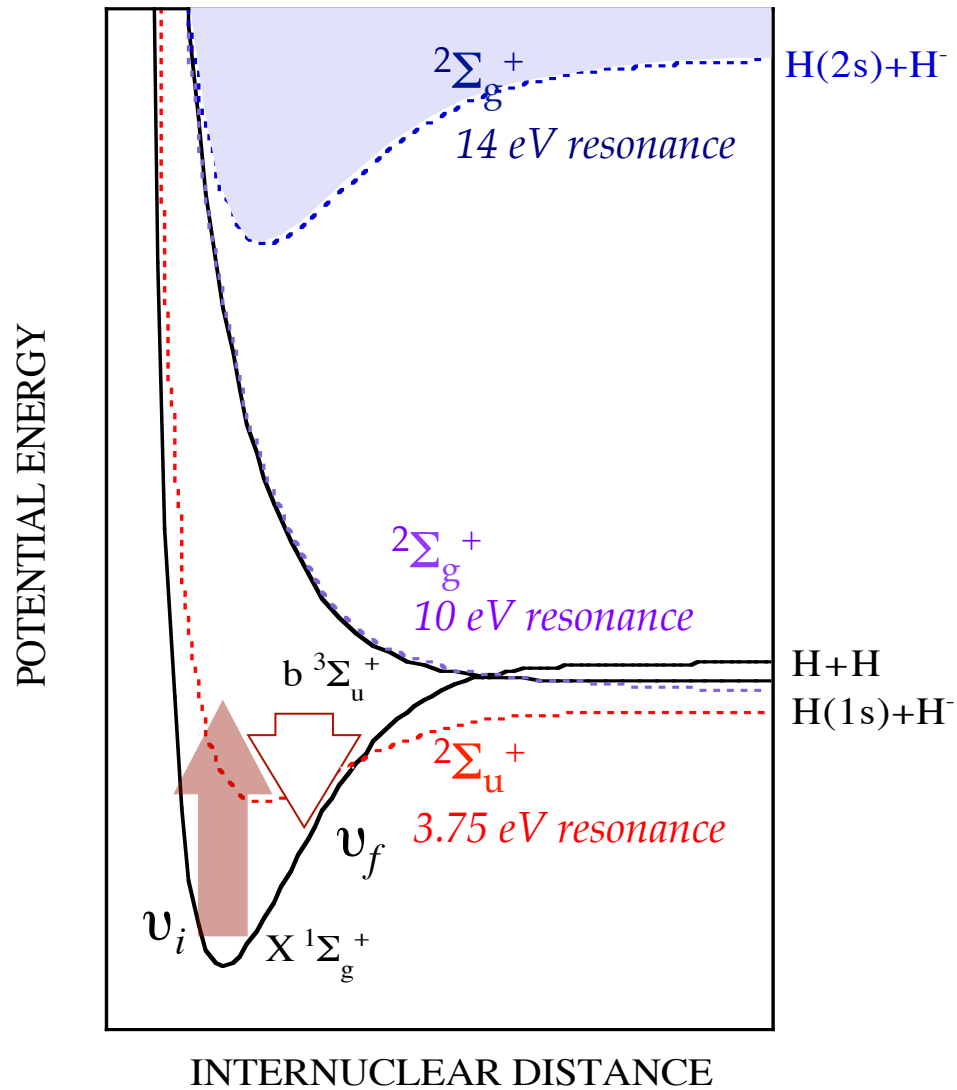
H₂ Dissociative Attachment



Horaček et al, Physical Review A (2004)

3.75 eV resonance

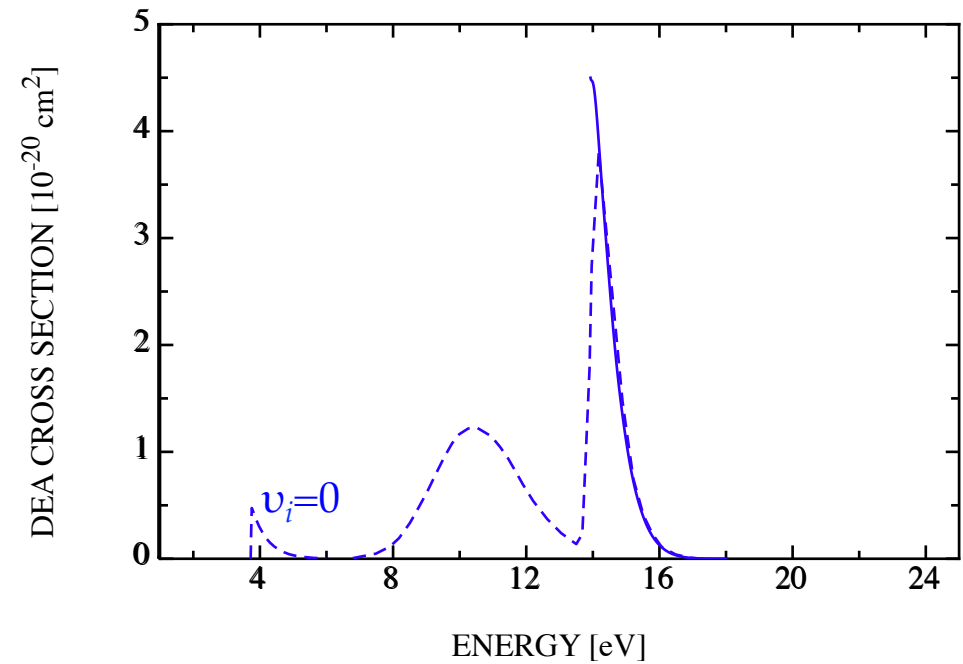
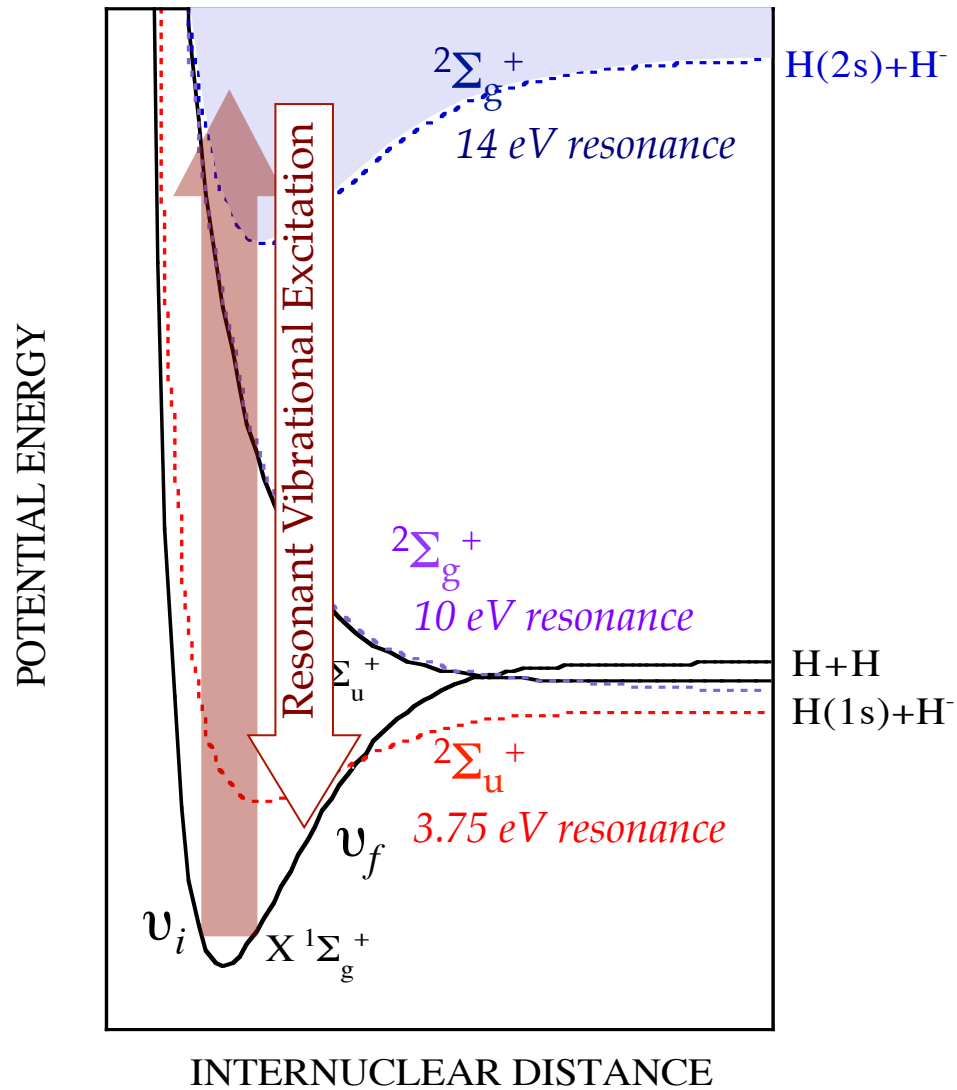
H₂ Dissociative Attachment



Horaček et al, Physical Review A (2004)

3.75 eV resonance

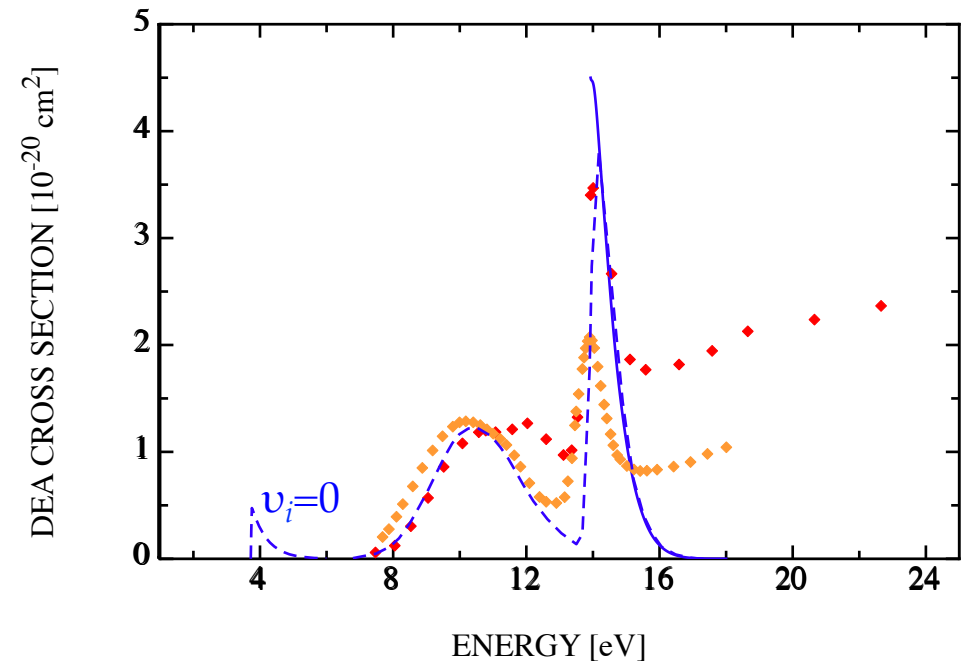
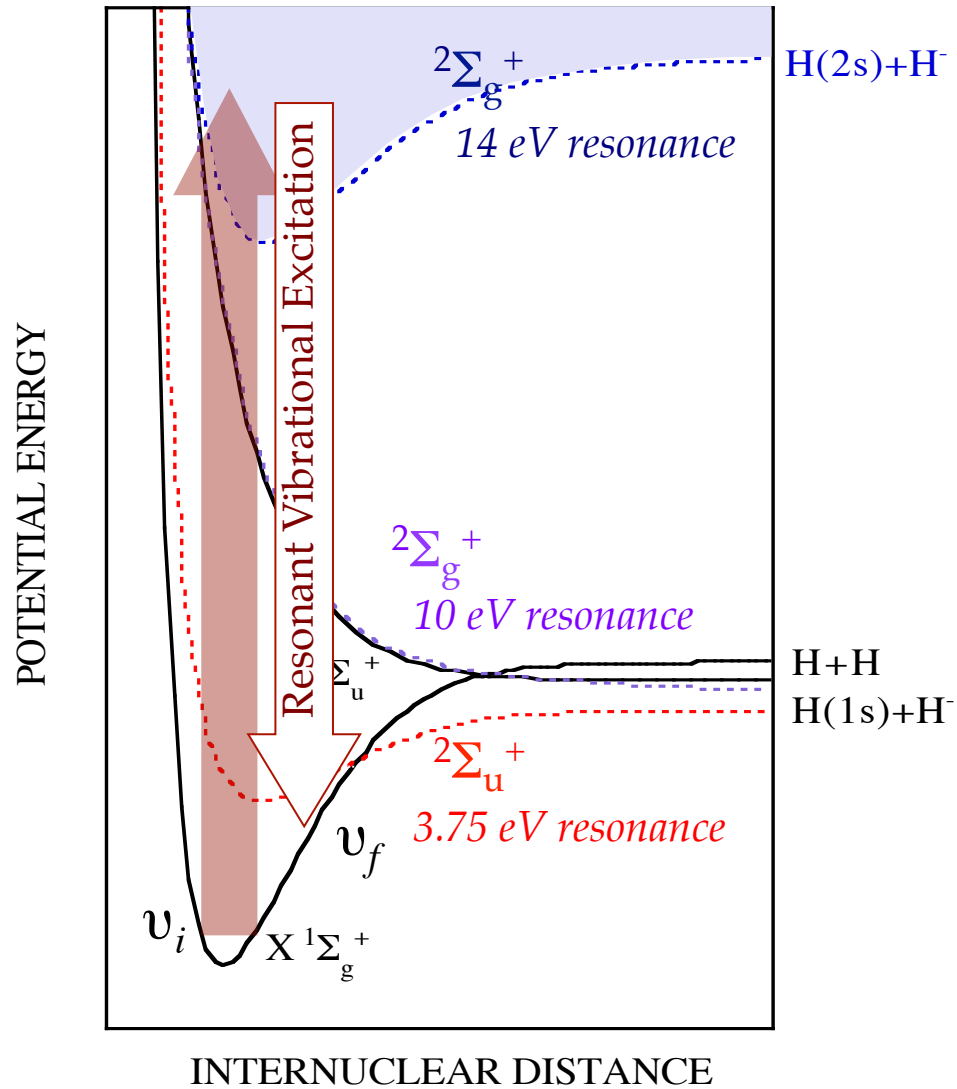
H₂ Dissociative Attachment



Horaček et al, Physical Review A (2004)

Celiberto et al, Chemical Physics (2012)

H₂ Dissociative Attachment



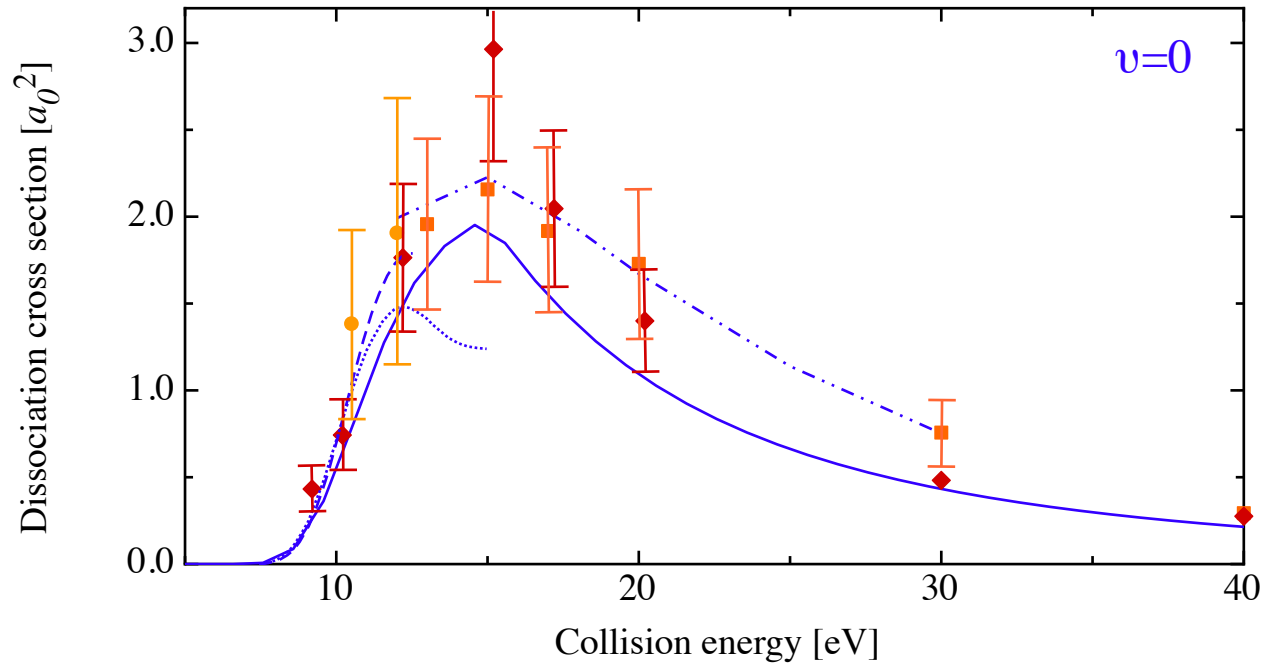
Horaček et al, Physical Review A (2004)

Celiberto et al, Chemical Physics (2012)

G. J. Schulz, Phys. Rev. 113 (1959) 816

D. Rapp et al, Phys. Rev. Lett. 14 (1965) 533

H₂ Direct Dissociation through *b* state



Khakoo & Segura, J Phys B (1994)
Khakoo et al., Phys Rev A (1987)

H. Nishimura, J. Phys. Soc. Jpn. 55 (1986) 3031

R-matrix

Stibbe & Tennyson,
The Astrophysical Journal (1999)
Trevisan&Tennyson,
Plasma Phys & Controlled Fusion (2002)

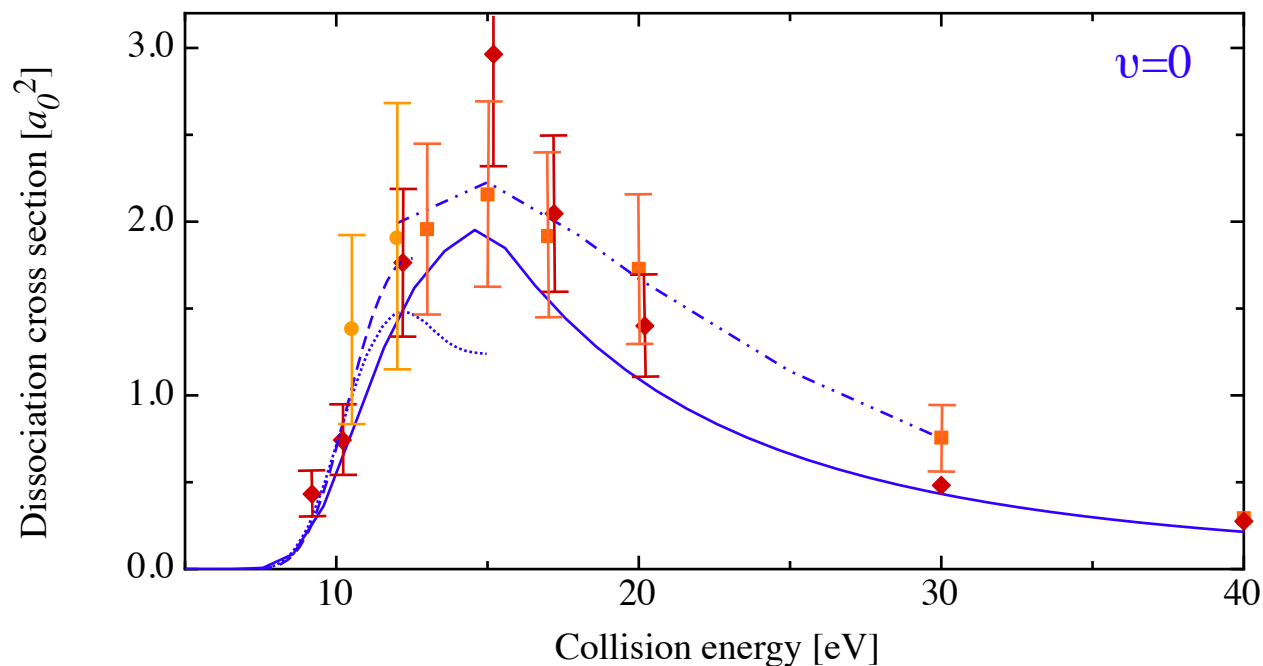
Gryzinskii approach

Celiberto et al.,
AtomicData&NuclearDataTables (2001)

Complex Khon method

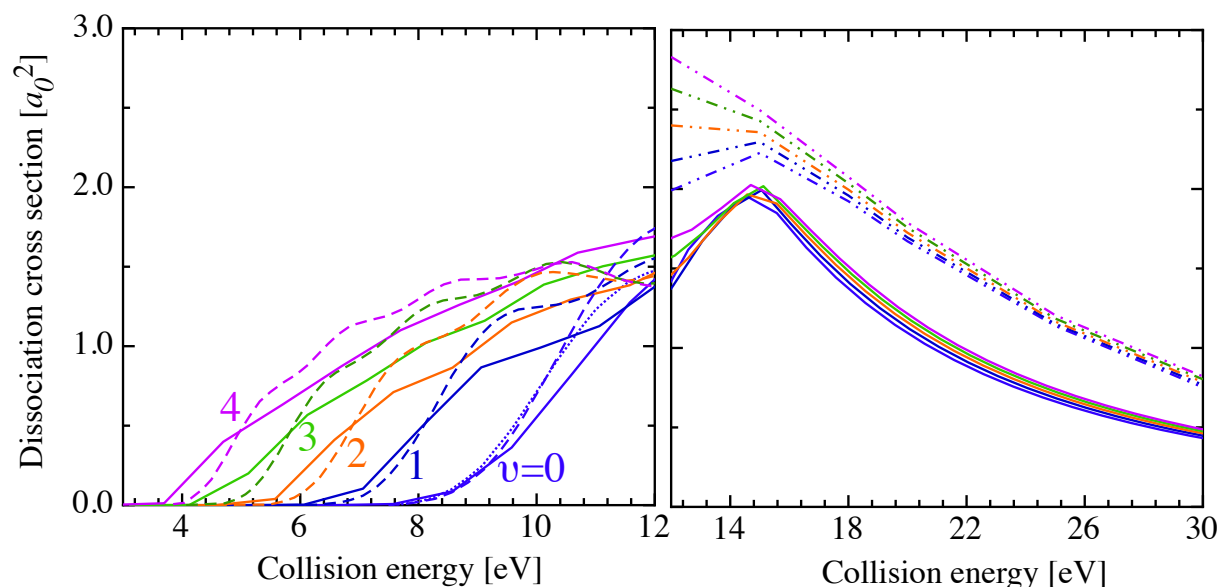
Rescigno & Schneider, J Phys B (1988)

H₂ Direct Dissociation through *b* state



Khakoo & Segura, J Phys B (1994)
Khakoo et al., Phys Rev A (1987)

H. Nishimura, J. Phys. Soc. Jpn. 55 (1986) 3031



R-matrix

Stibbe & Tennyson,
The Astrophysical Journal (1999)
Trevisan & Tennyson,
Plasma Phys & Controlled Fusion (2002)

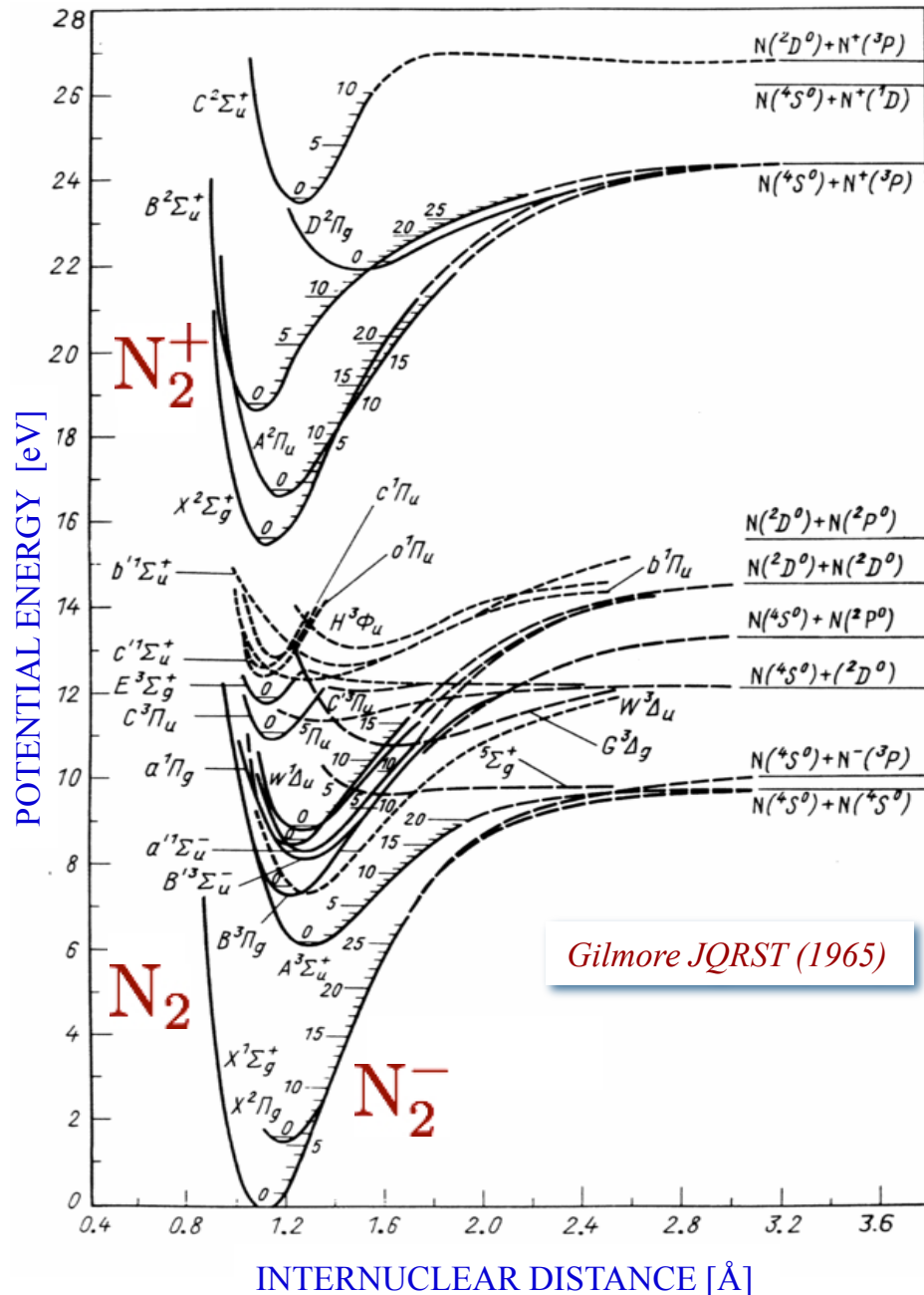
Gryzinskii approach

Celiberto et al.,
AtomicData&NuclearDataTables (2001)

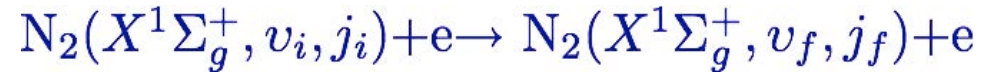
Complex Khon method

Rescigno & Schneider, J Phys B (1988)

ELECTRON IMPACT INDUCED PROCESSES: THE DATA SET FOR N₂

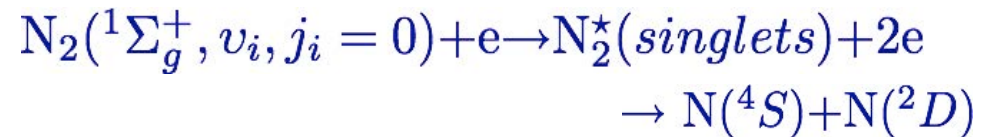


RESONANT VIBRATIONAL EXCITATION



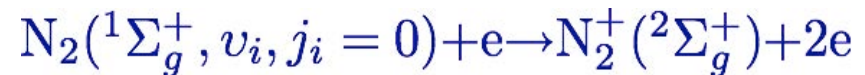
Laporta et al PlasmaSourcesScience&Tech (2012)

VIBRONIC EXCITATION



Capitelli&Celiberto in
"Novel Aspects of Electron-Molecule Collisions"
Ed KH Becker (1998)

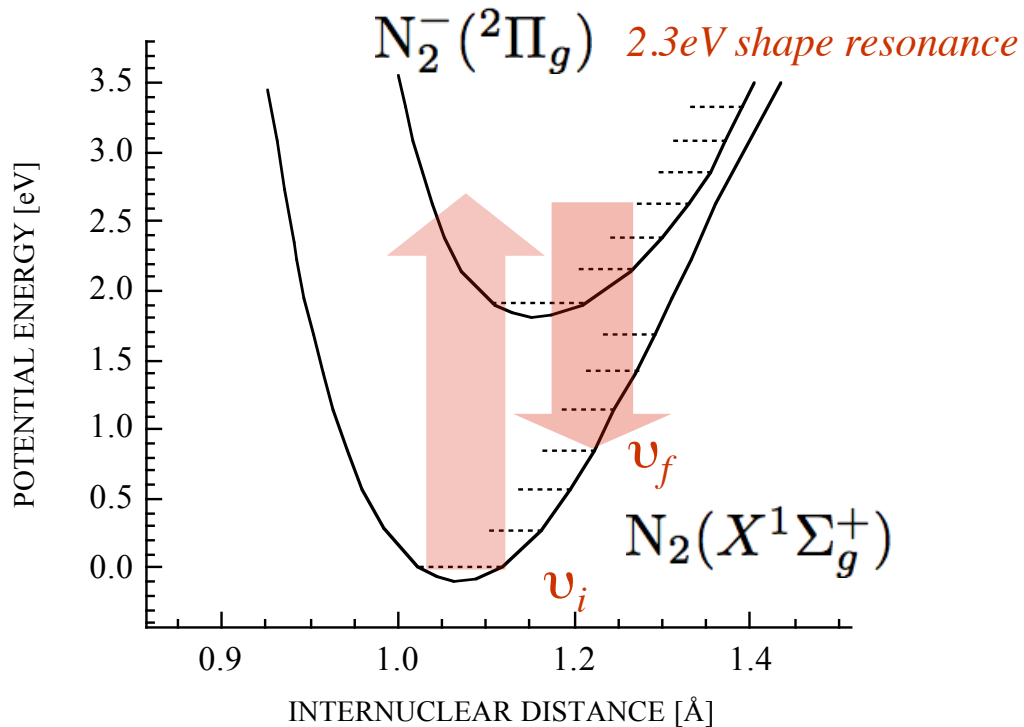
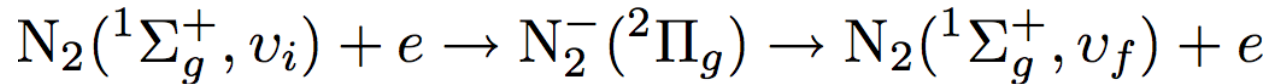
NON-DISSOCIATIVE IONIZATION



Kosarim et al ChemPhysLett (2005)
Laricchiuta et al. AIAA paper 2006-2898 (2006)

Electron-Impact induced Vibrational Pumping

resonant vibrational excitation (eV processes)



V. Laporta, R. Celiberto, J.M. Wadehra,
Plasma Sources Sci. Technol. 21 (2012) 055018

M. Capitelli et al., Plasma Chemistry &
Plasma Proc. 32 (2012) 427

*collection of theoretical and
experimental cross sections*

$$v_i = 0 \rightarrow v_f = 1-17$$

M.Allan, J.Phys.B (1985)

$$v_i = 0 \rightarrow v_f = 18-19$$

L.A.Morgan, J.Phys.B (1986)

$$v_i = 1-7 \rightarrow v_f = v_i + 1-8$$

N. Chandra & A.Temkin, NASA-TN (1976)

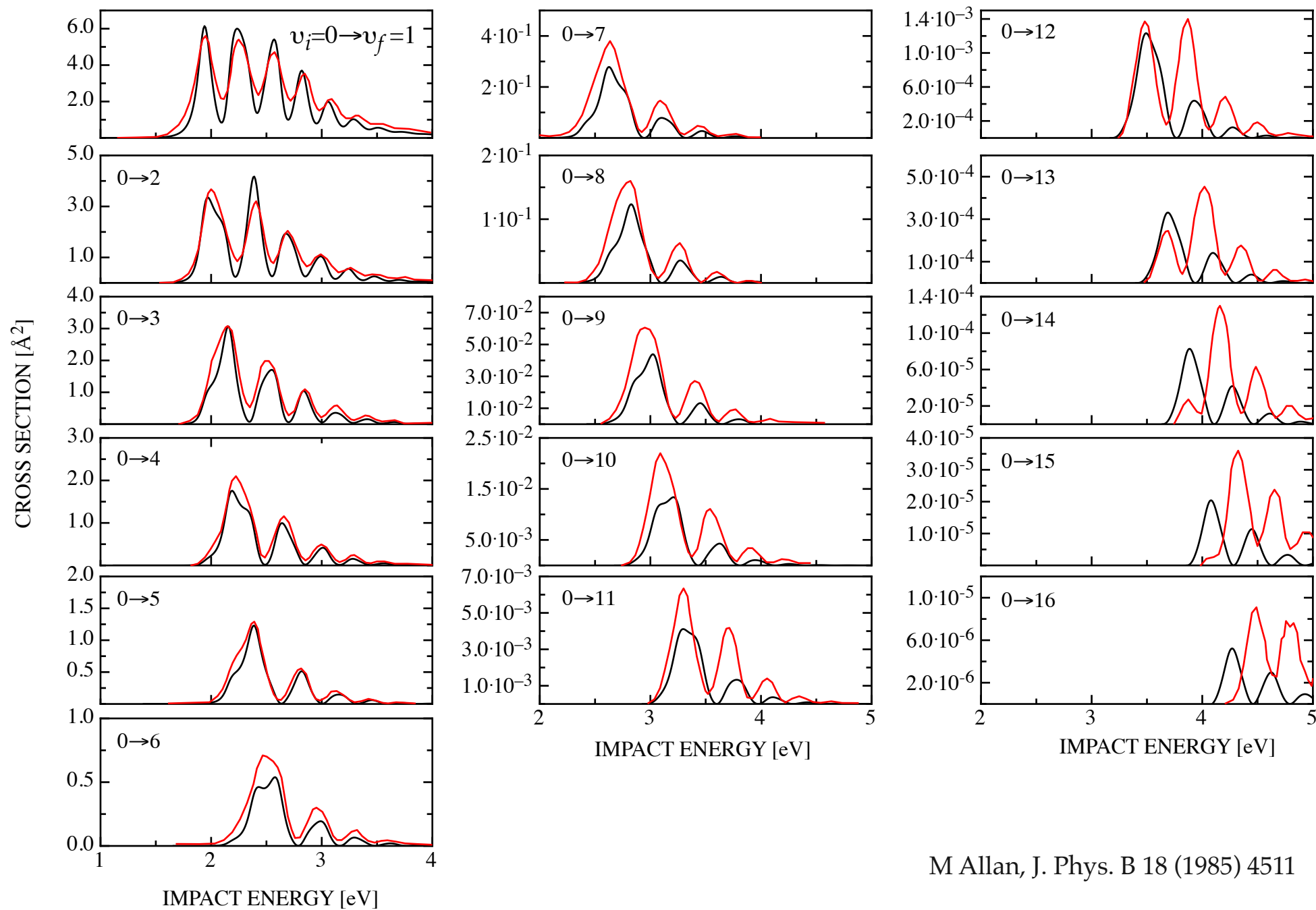
OLD DATASET



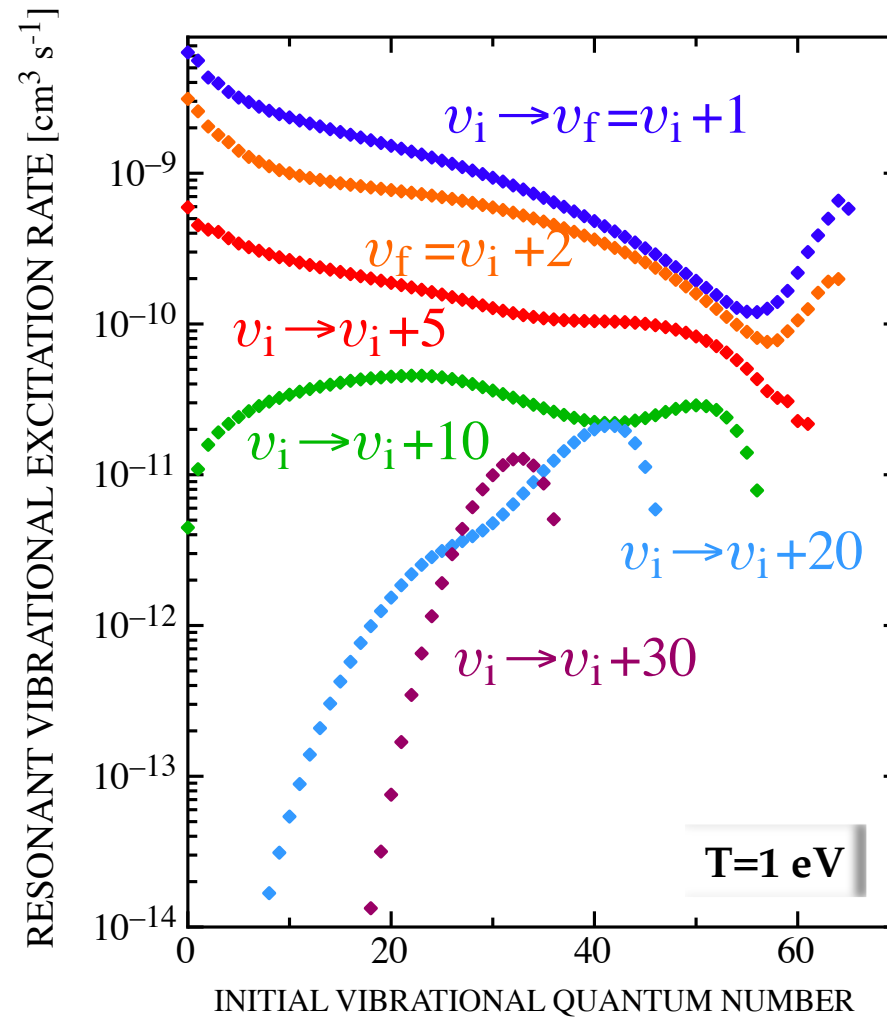
*local theory of resonance
complete set of cross sections for
transitions among all the levels
in the vibrational ladder*

NEW DATASET

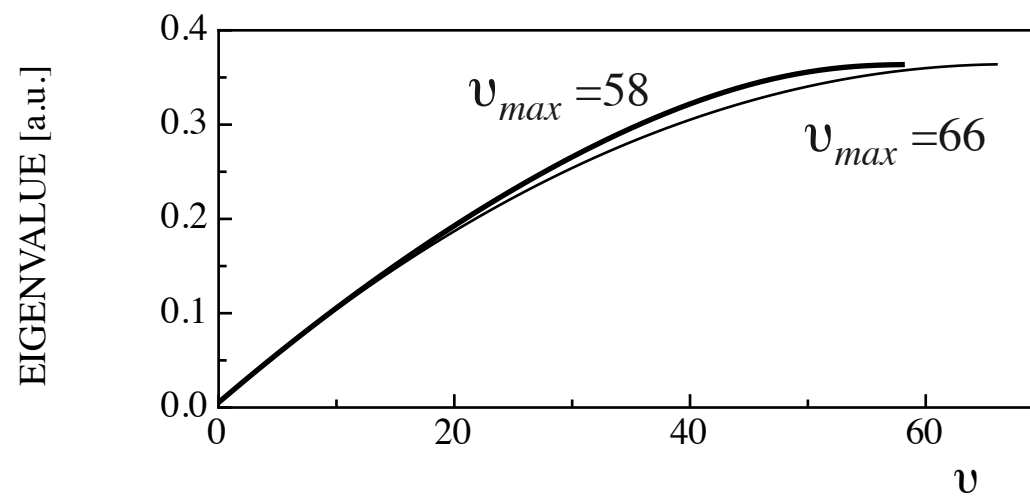
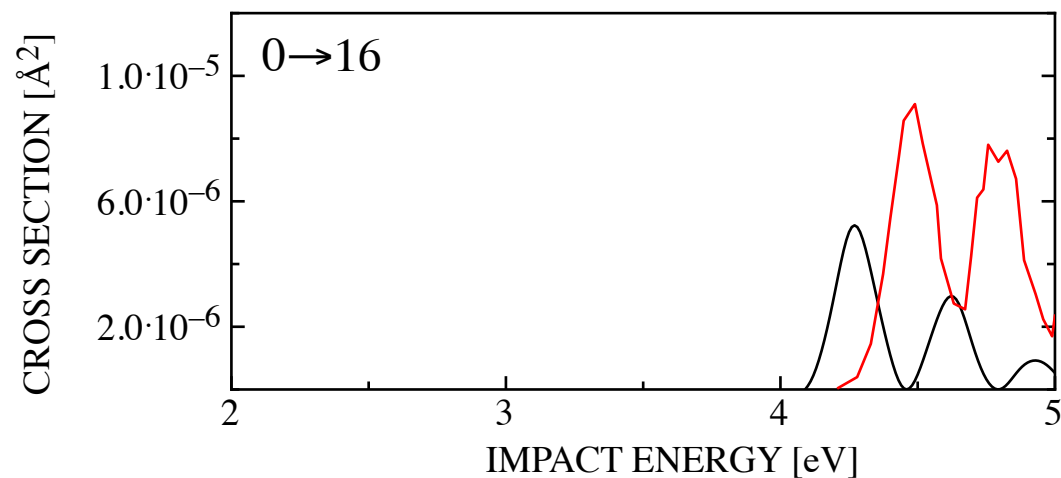
Electron-Impact induced Vibrational Pumping



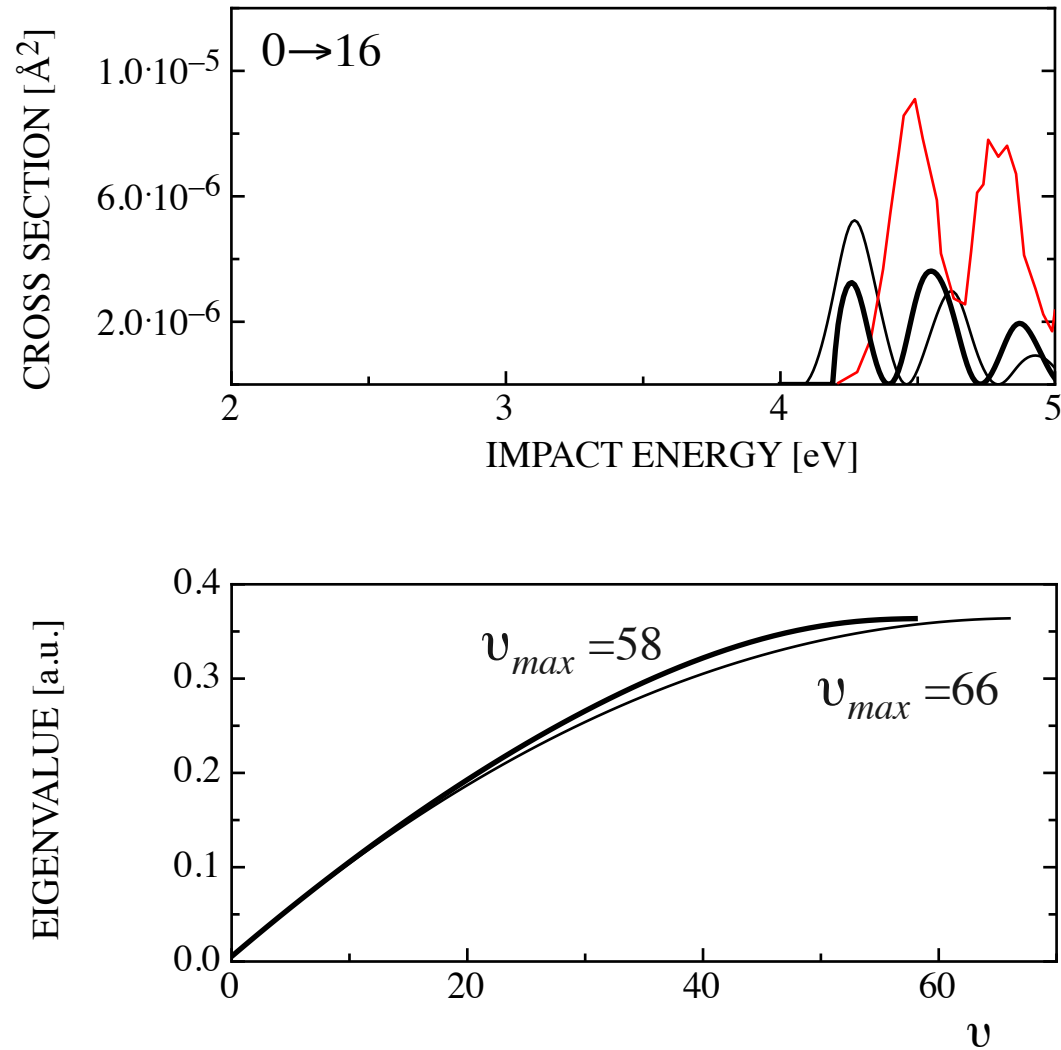
Electron-Impact induced Vibrational Pumping



Electron-Impact induced Vibrational Pumping

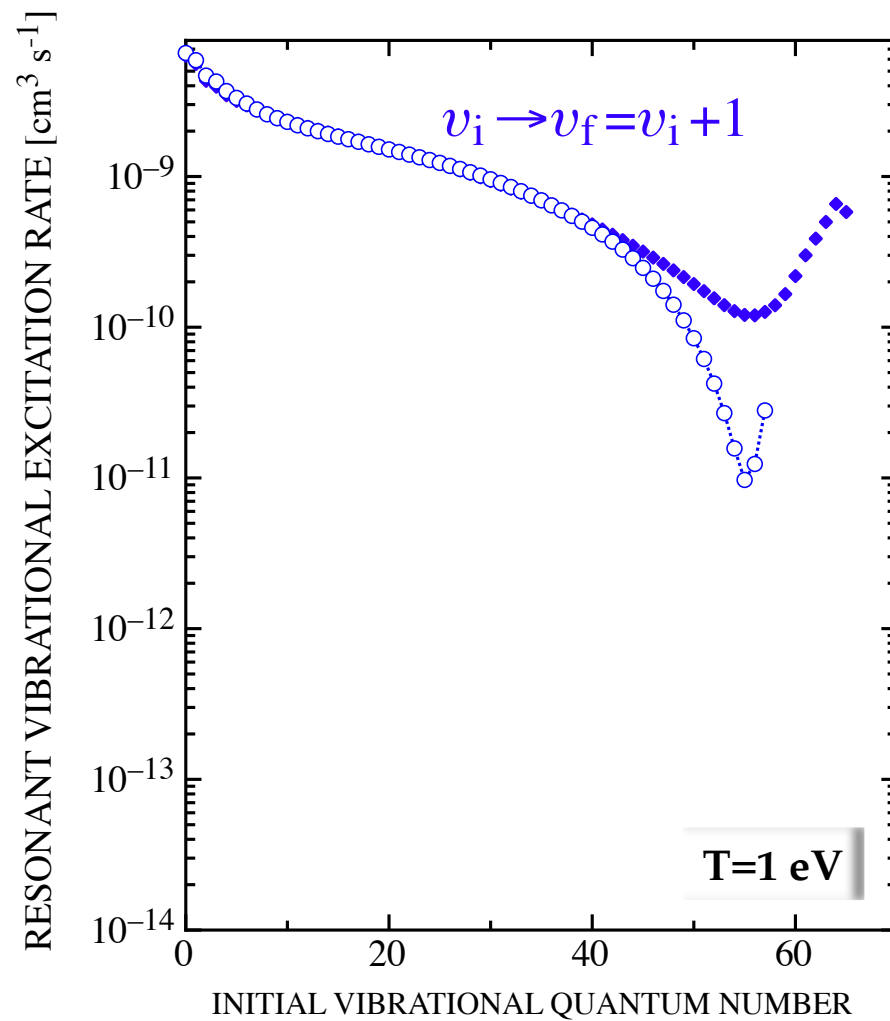


Electron-Impact induced Vibrational Pumping



Laporta et al. Plasma Sources Sci. Technol. (2014)

Electron-Impact induced Vibrational Pumping

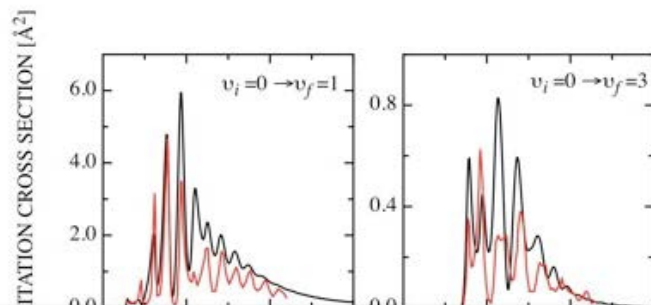


The Phys4EntryDB : Resonant Vibrational Excitation

Comparisons with literature

Cross sections

V. Laporta, R. Celiberto, J.M. Wadehra, PSST 21 (2012) 055018

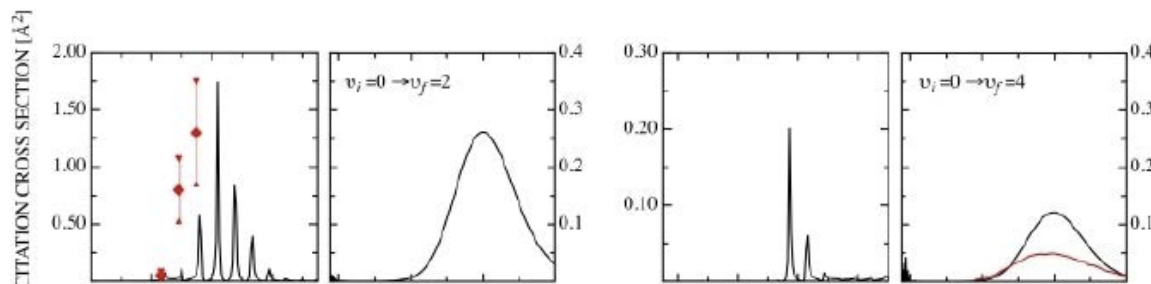


RVE e-NO

Comparisons with literature

Cross sections

V. Laporta, R. Celiberto, J. Tennyson, PSST 22 (2013) 025001

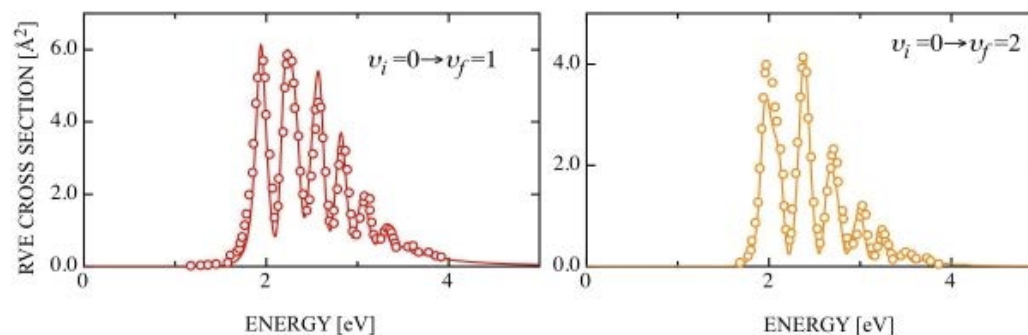


RVE e-O₂

Comparisons with literature

Cross sections

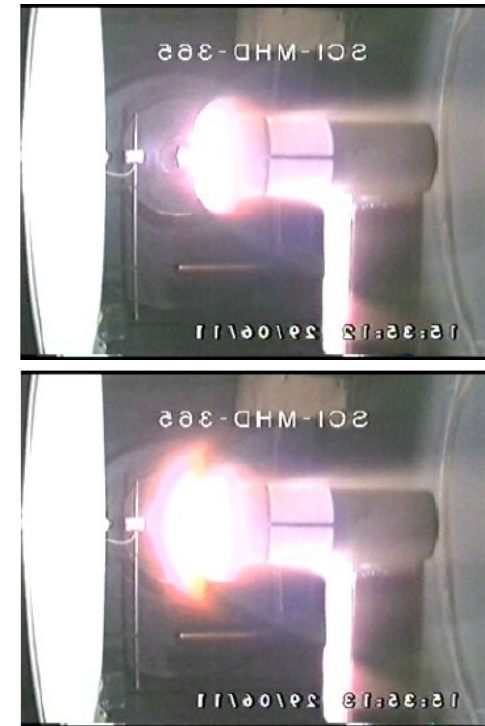
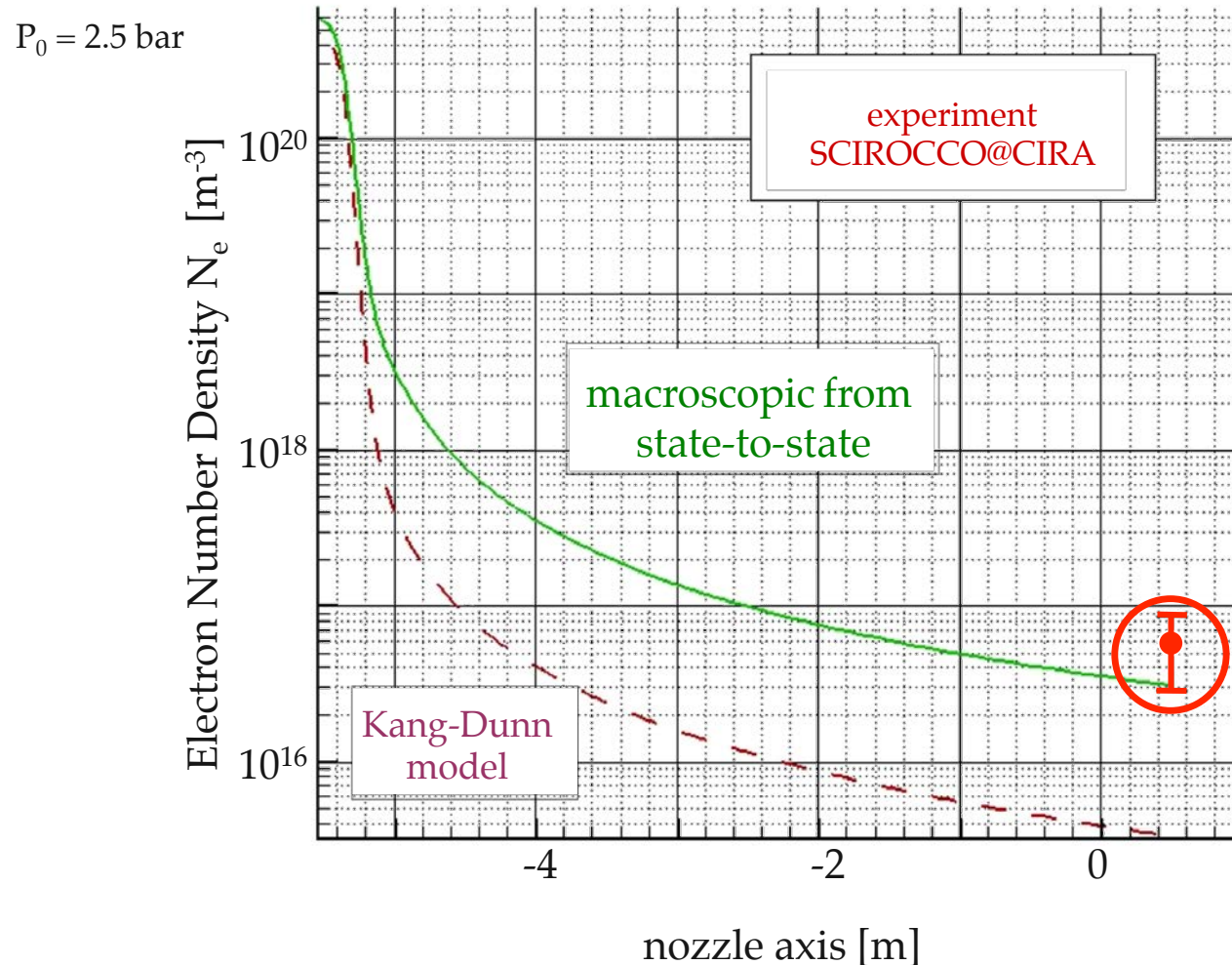
V. Laporta, R. Celiberto, J.M. Wadehra, PSST 21 (2012) 055018



RVE e-N₂

Resonant vibrational excitation cross sections as function of collision energy for 0→1, 0→2 excitations transitions. (open circles) unpublished experimental results of Wong as cited in [L. Dubé, A. Herzberg (1979)]

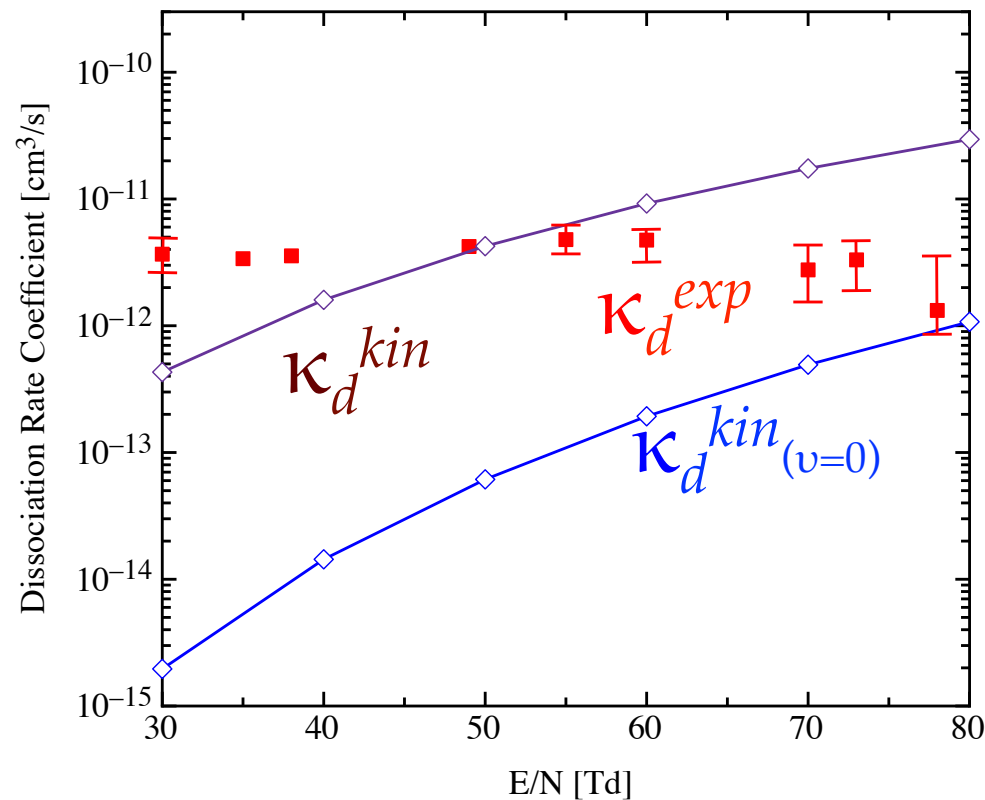
Experimental/theoretical investigation of MHD interaction in Air



A. Cristofolini et al.,
AIAA paper 2012-3179 (2012)

A. Schettino & D. D'Ambrosio,
The Open Plasma Physics Journal 7 (Special Issue) (2014)

N₂ continuous discharge: dissociation rate coefficient at stationary state



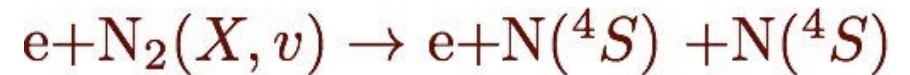
$T = 1000 \text{ K}$
 $p = 5.6 \text{ torr}$
 SS at 0.1s

κ_d^{exp}

experimental global dissociation rate

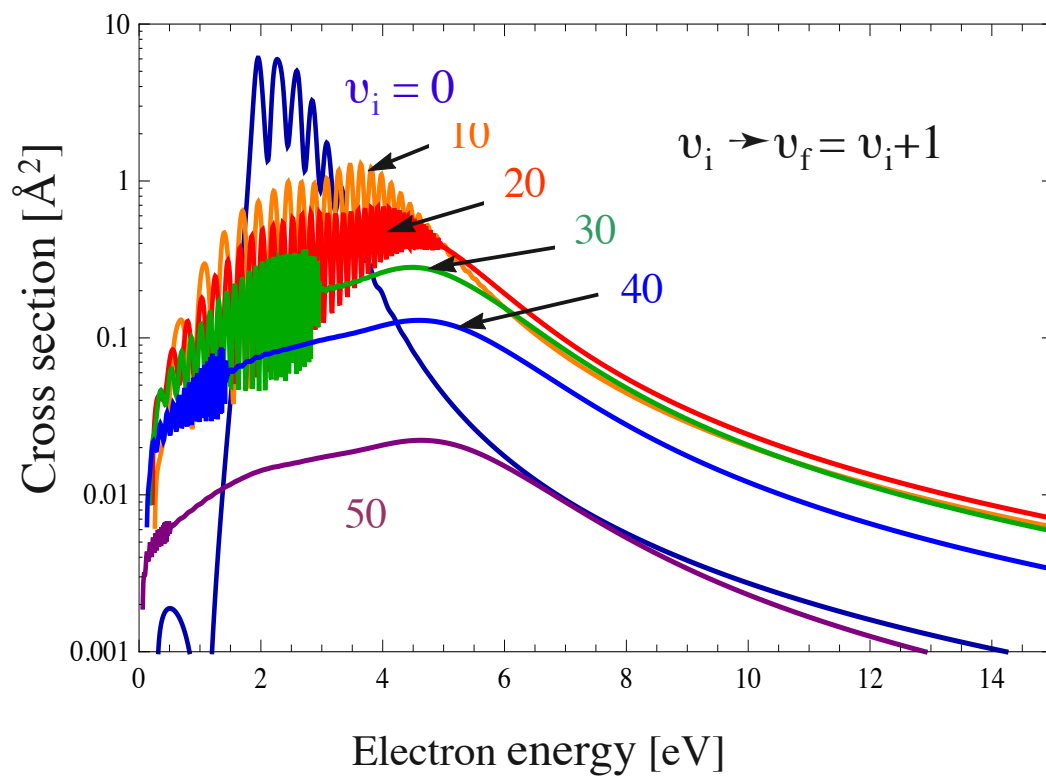
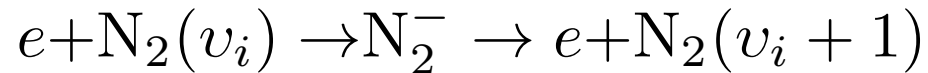
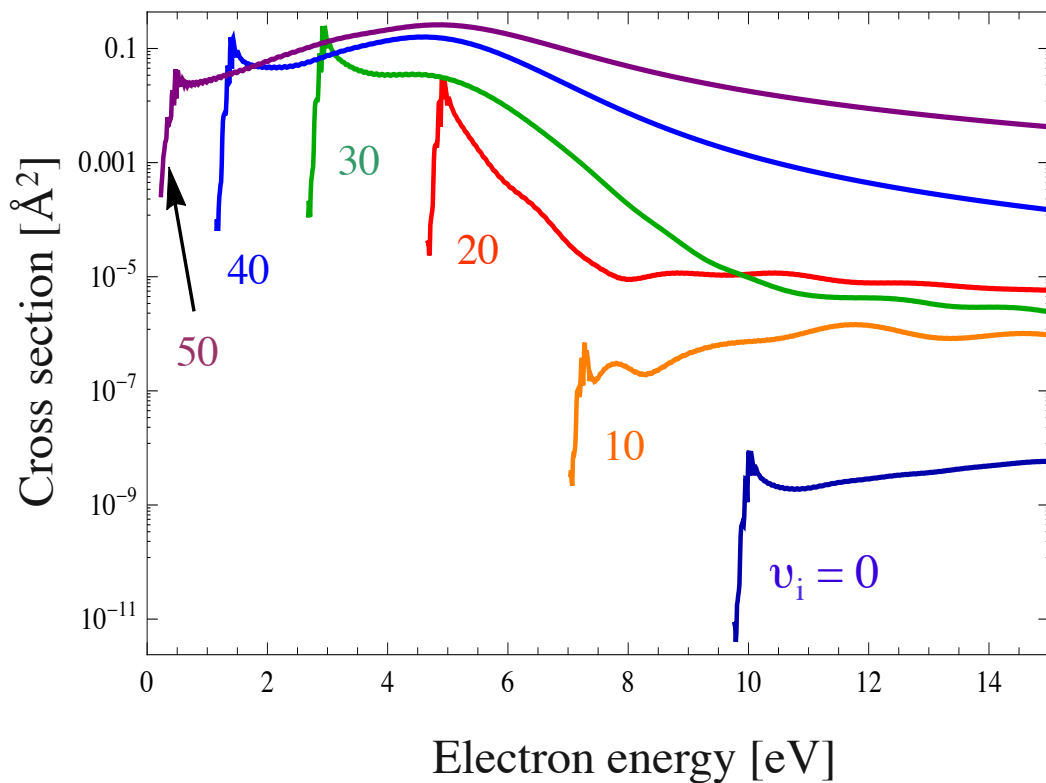
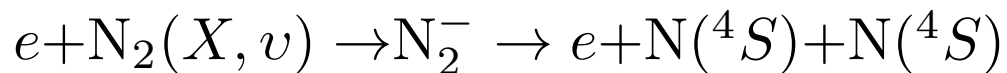
L. S. Polak et al. in Proceedings 12th ICPIG 1975

$$\kappa_d^{kin} = \sum_{v=0}^{v_{\max}} k_d(v) \frac{N_v}{N_{\text{tot}}}$$



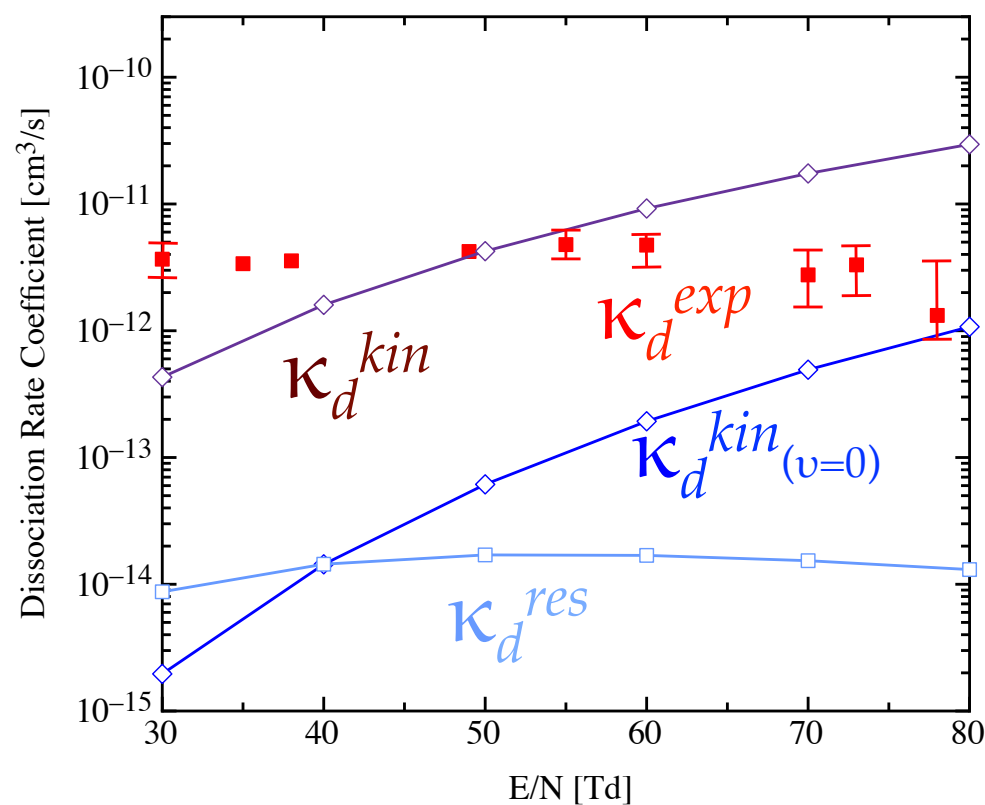
direct dissociation (predissociation)

resonant dissociation *vs* mono-quantum excitation in e -N₂ collisions



Laporta et al. Plasma Sources Sci. Technol. (2014)

N_2 continuous discharge: dissociation rate coefficient at stationary state



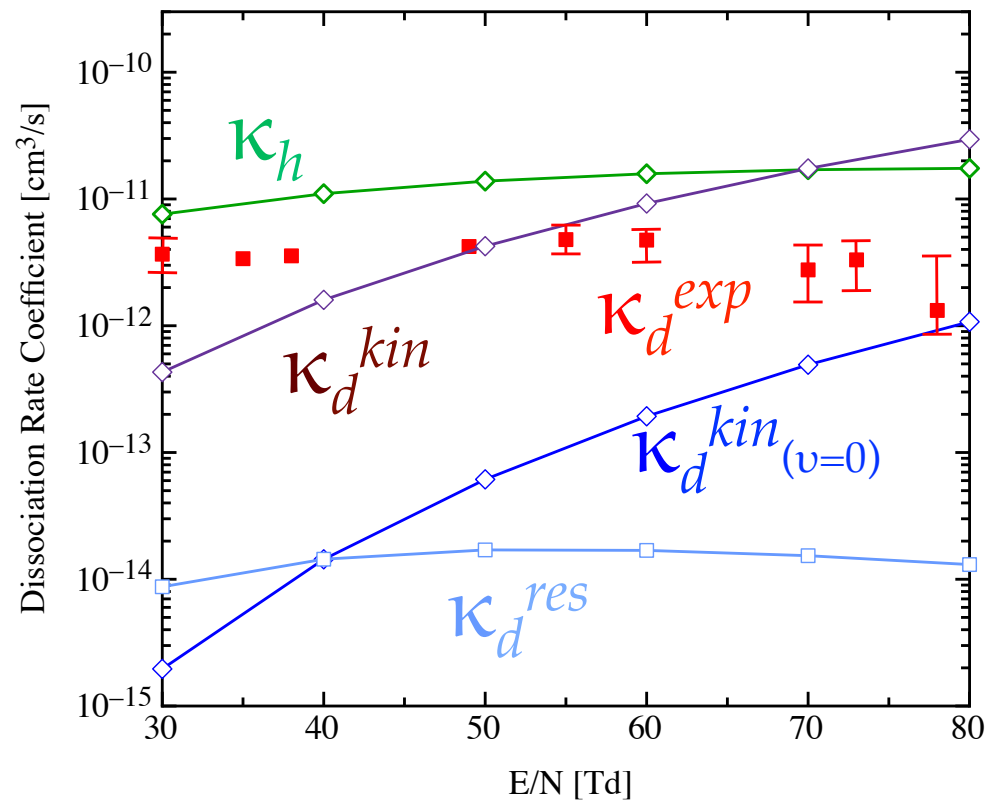
κ_d^{res}



resonant dissociation

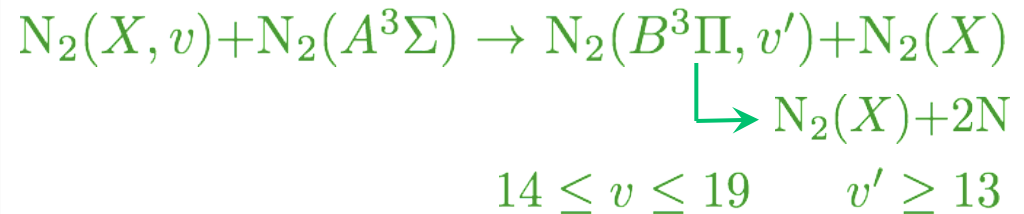
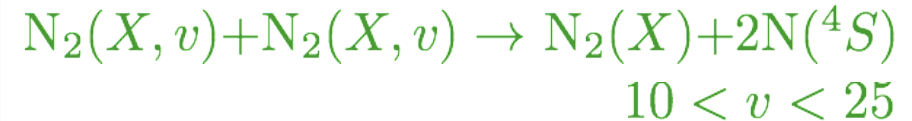
M. Capitelli et al. Chemical Physics (2014)

N_2 continuous discharge: dissociation rate coefficient at stationary state



κ_h

heavy collisions involving electronically and vibrationally excited molecules



V. Guerra, E. Galiaskarov, J. Loureiro,
Chem. Phys. Lett. 371 (2003) 576

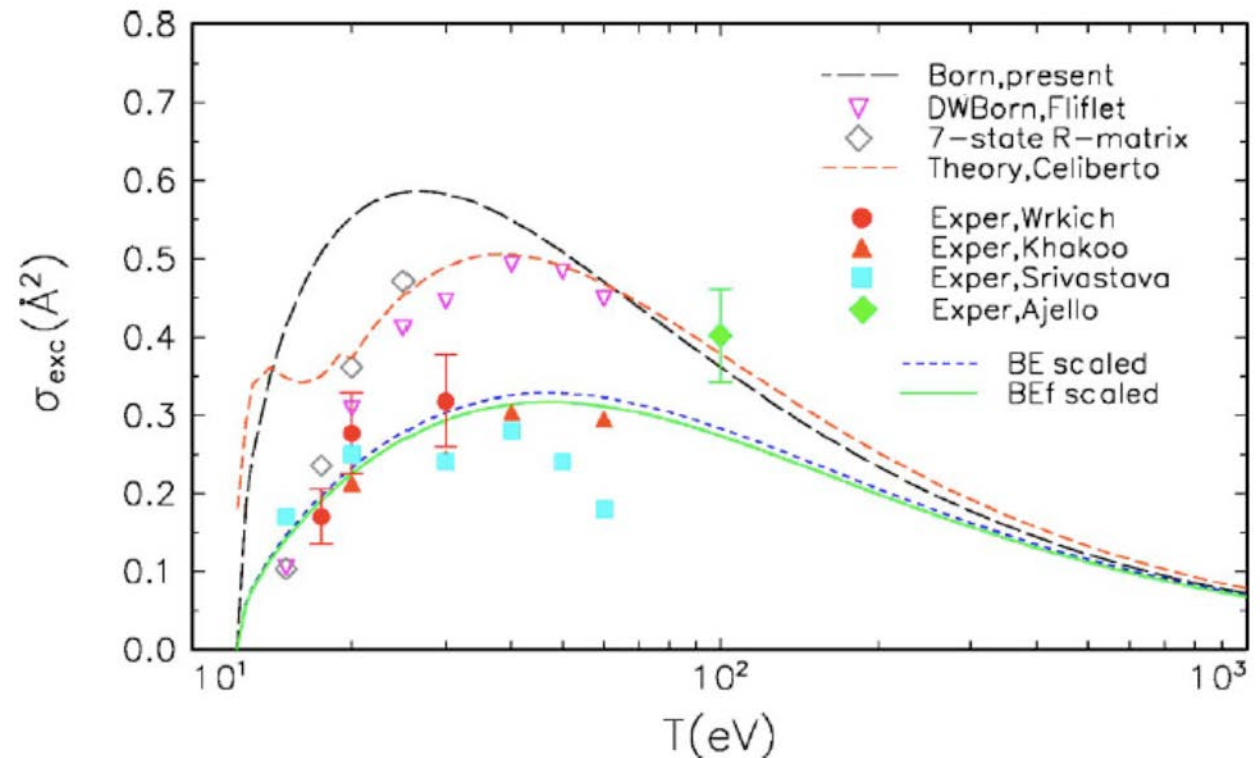
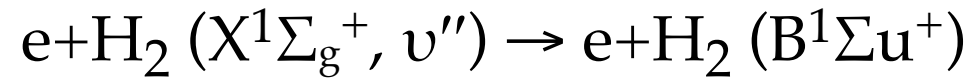
M. Capitelli et al. Chemical Physics (2014)

Simplified Theoretical Approaches

- ✓ *Gryzinskii approach*
- ✓ *Bef-scaled approach*
- ✓ *Similarity approach*

Simplified Theoretical Approaches

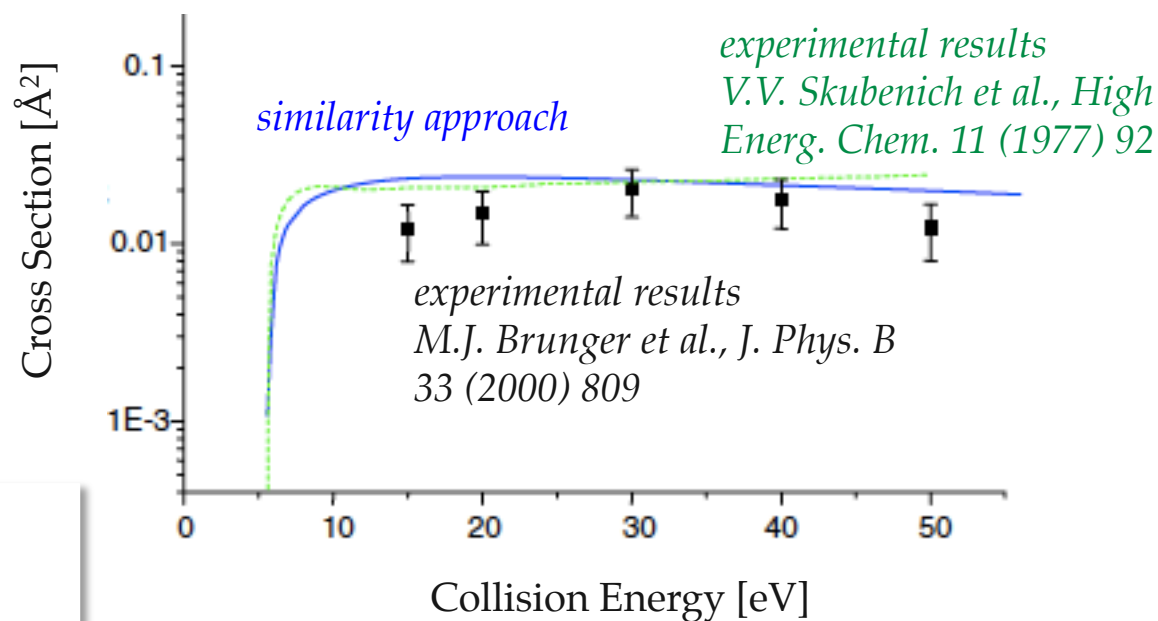
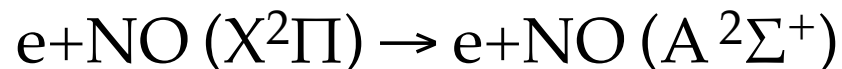
- ✓ Gryzinskii approach
- ✓ *Bef-scaled approach*
- ✓ Similarity approach



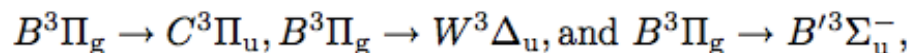
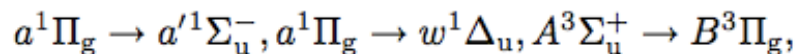
Y.K. Kim, J. Chem. Phys. 126 (2007) 064305

Simplified Theoretical Approaches

- ✓ Gryzinskii approach
- ✓ Bef-scaled approach
- ✓ *Similarity approach*



N_2 e-impact induced allowed excitation

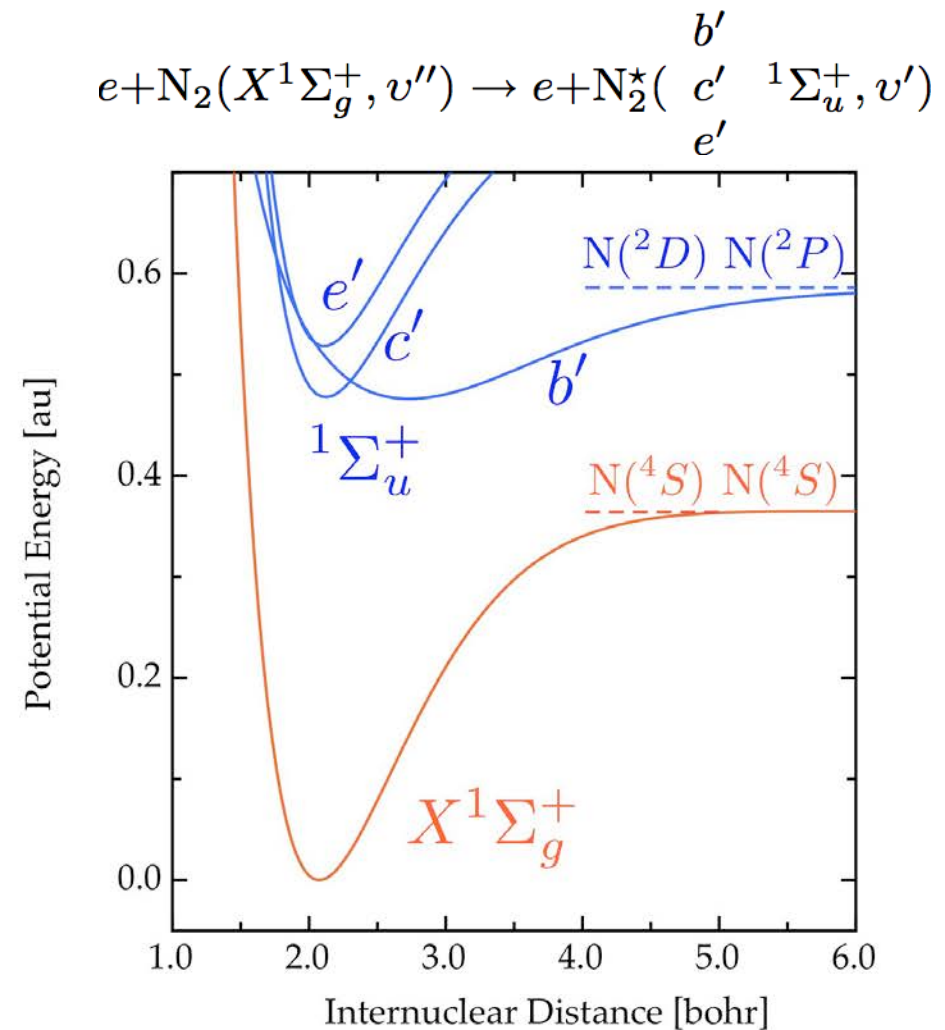
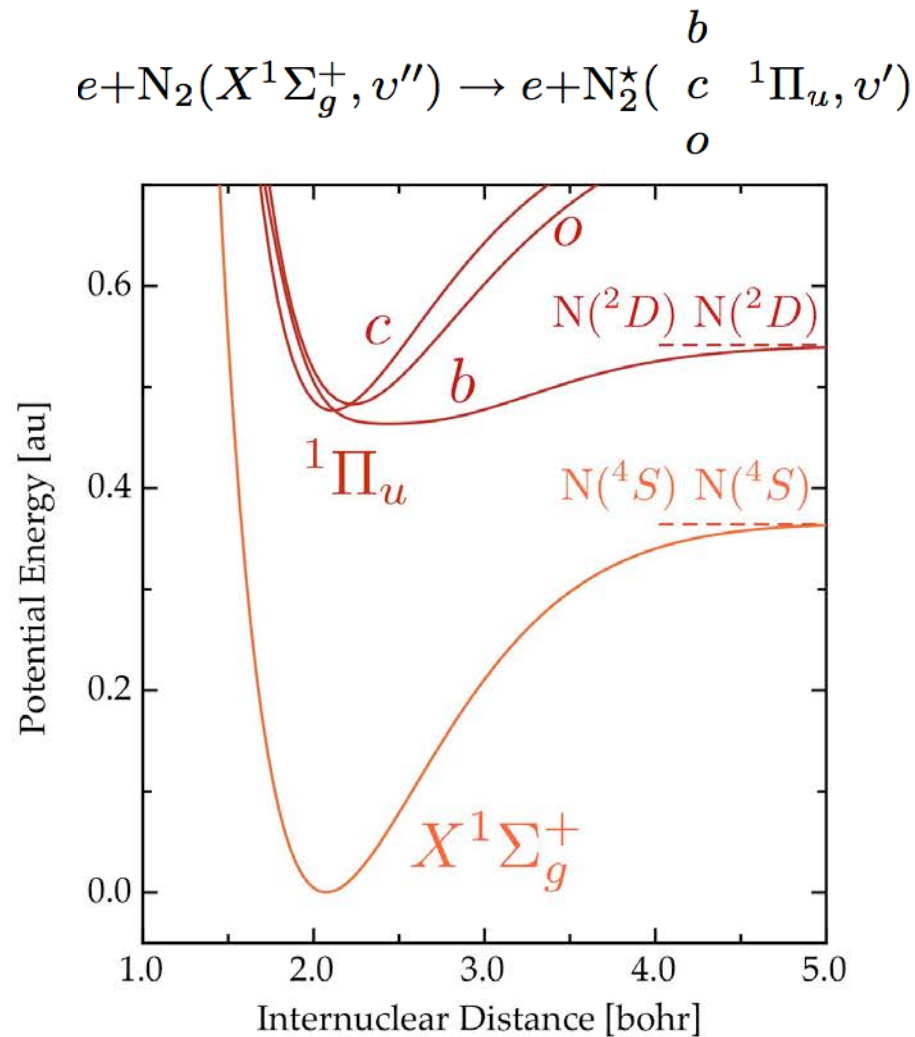


Xin&Ding, Plasma Science and Tech. 16 (2014) 104

S. Adamson, Chem.Phys. Letters 436 (2007) 308

A.V. Eletsii & B.M. Smirnov, Zh. Eksp. Teor. Fiz. 84 (1983) 1639

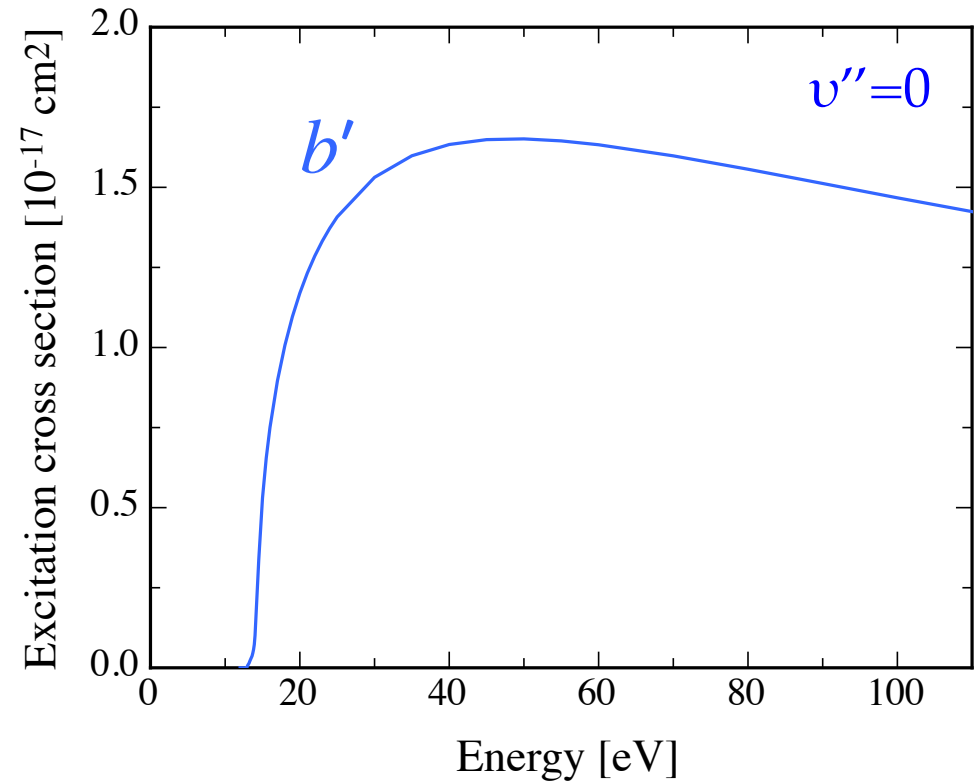
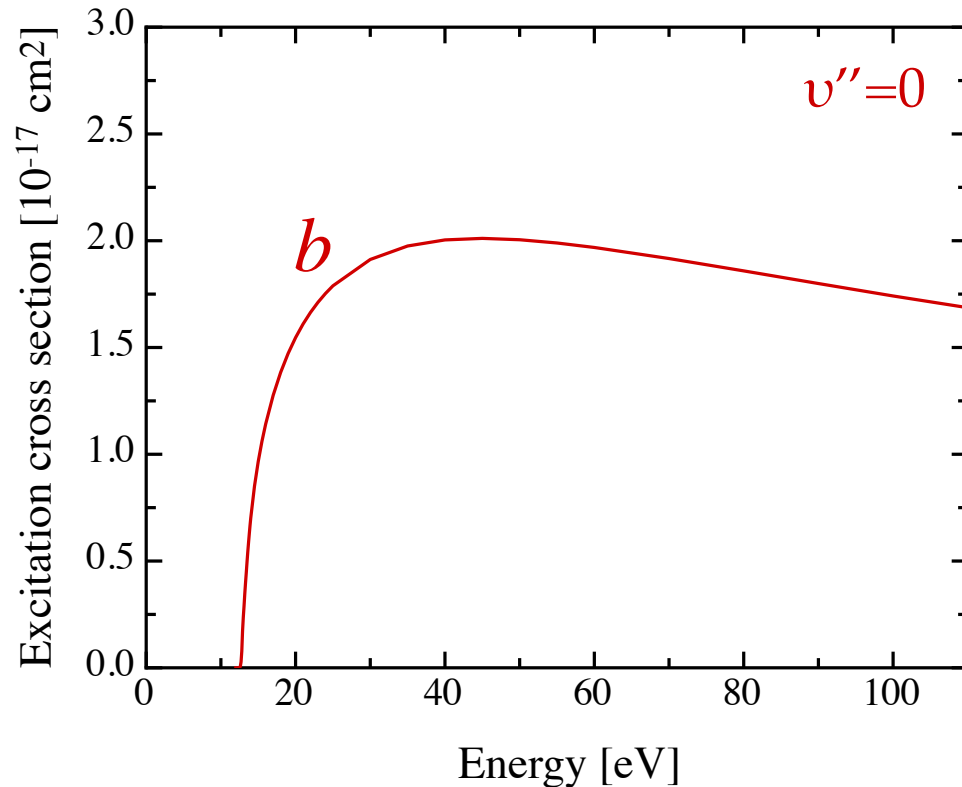
Electron-Impact Vibronic Excitation of Singlet Terms in N₂ Spectrum



- ✓ *perturbed vibrational progression*
- ✓ *responsible of emission in VUV region*
- ✓ *efficiently predissociated*

D. Spelsberg & W. Meyer, J.Chem.Phys 115 (2001) 6438
Z. Fan et al., Molecular Physics (2014)

Vibronic Excitation Cross Sections: comparison with experiments

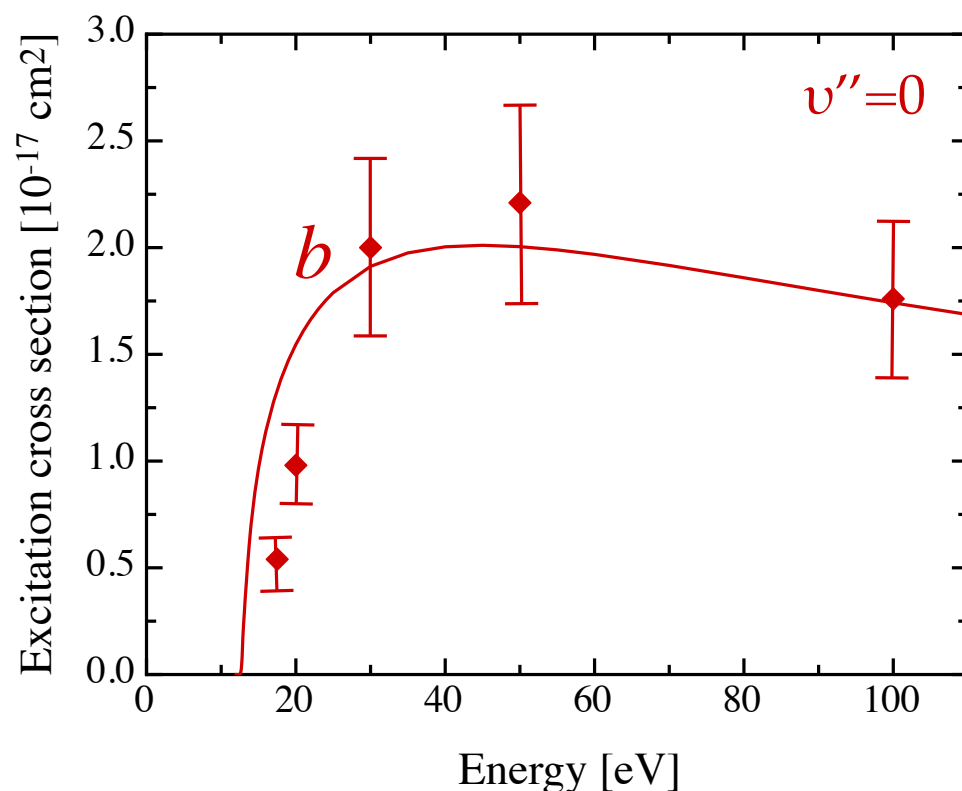


excitation from theoretical state-to-state cross sections

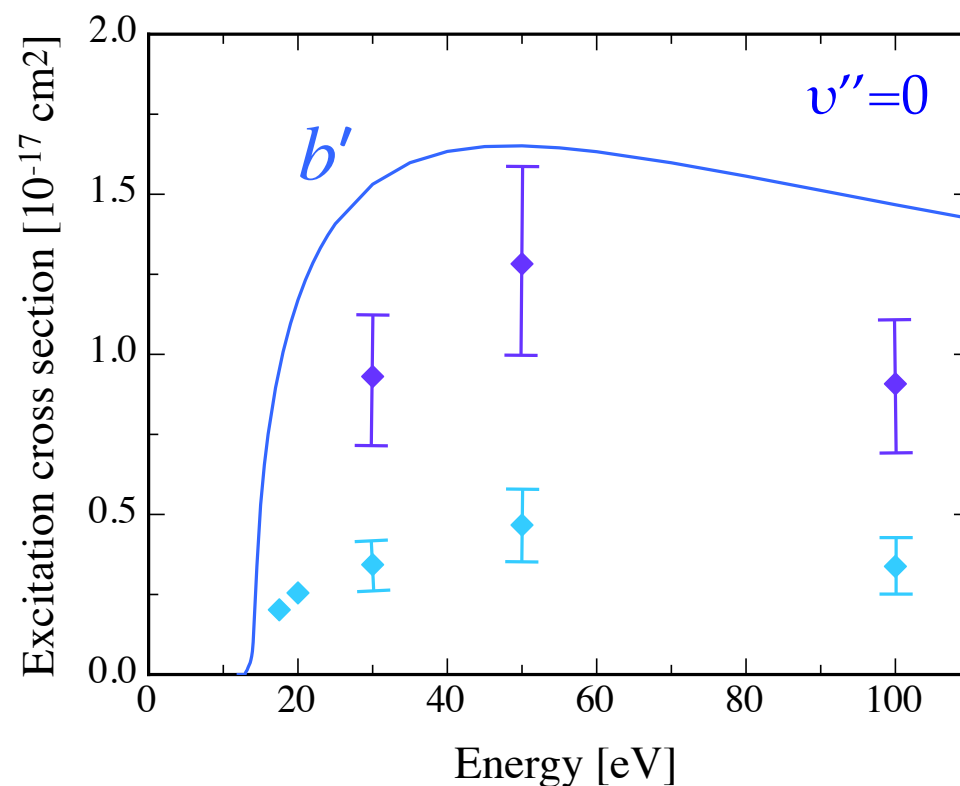
$$\sigma_{v''=0}^b(E) = \sum_{v'} \sigma_{v'v''}(E) P_{v'}(b) \quad \text{state character}$$

R. Celiberto et al., The Open Plasma Physics Journal 7 (Special Issue) (2014)

Vibronic Excitation Cross Sections: comparison with experiments



spectral range [12.00-13.82 eV]
vibrational analysis $b[v' 0-14]$ $c[v' 0-3]$ $o[v' 0-3]$



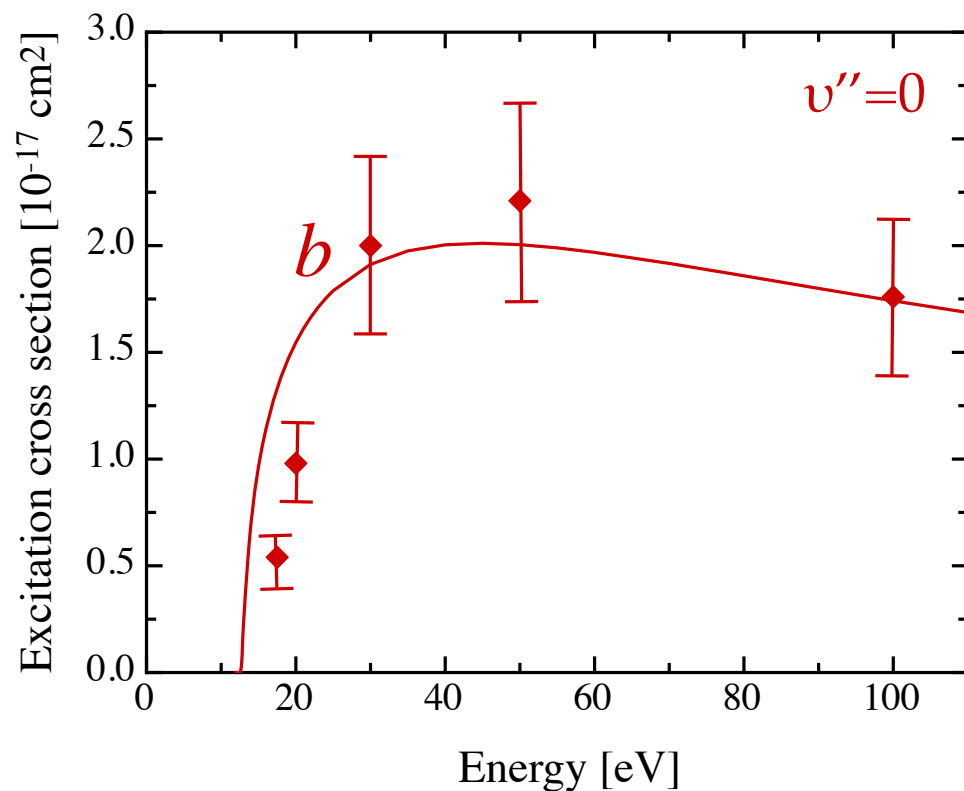
vibrational analysis $b'[v' 0-10]$ $c'[v' 0-3]$
or
 $b'[v' 0-16]$ $c'[v' 0-5]$

M.A. Khakoo et al., Phys.Rev.A 77 (2008) 012704

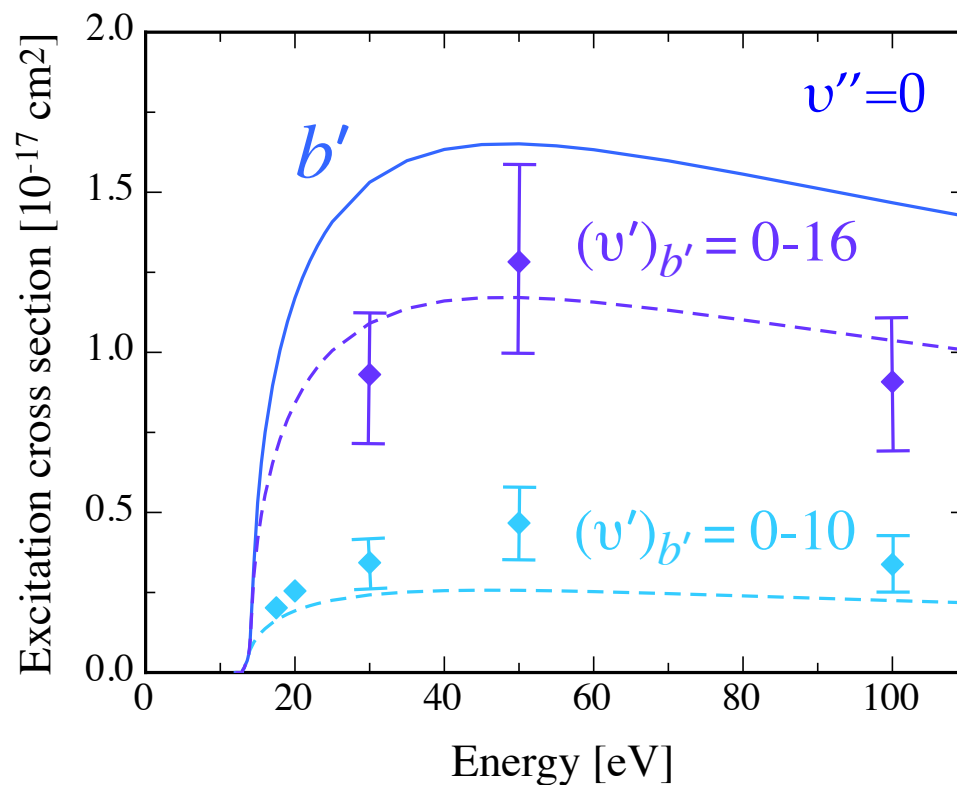
C.P. Malone et al., Phys.Rev.A 85 (2012) 062704

A. LARICCHIUTA The state-of-the-art description of vibrational-electronic excitations
MASTER CLASS 2014 Physikzentrum Bad Honnef (Germany), October 10th, 2014

Vibronic Excitation Cross Sections: comparison with experiments



spectral range [12.00-13.82 eV]
vibrational analysis $b[v' 0-14]$ $c[v' 0-3]$ $o[v' 0-3]$



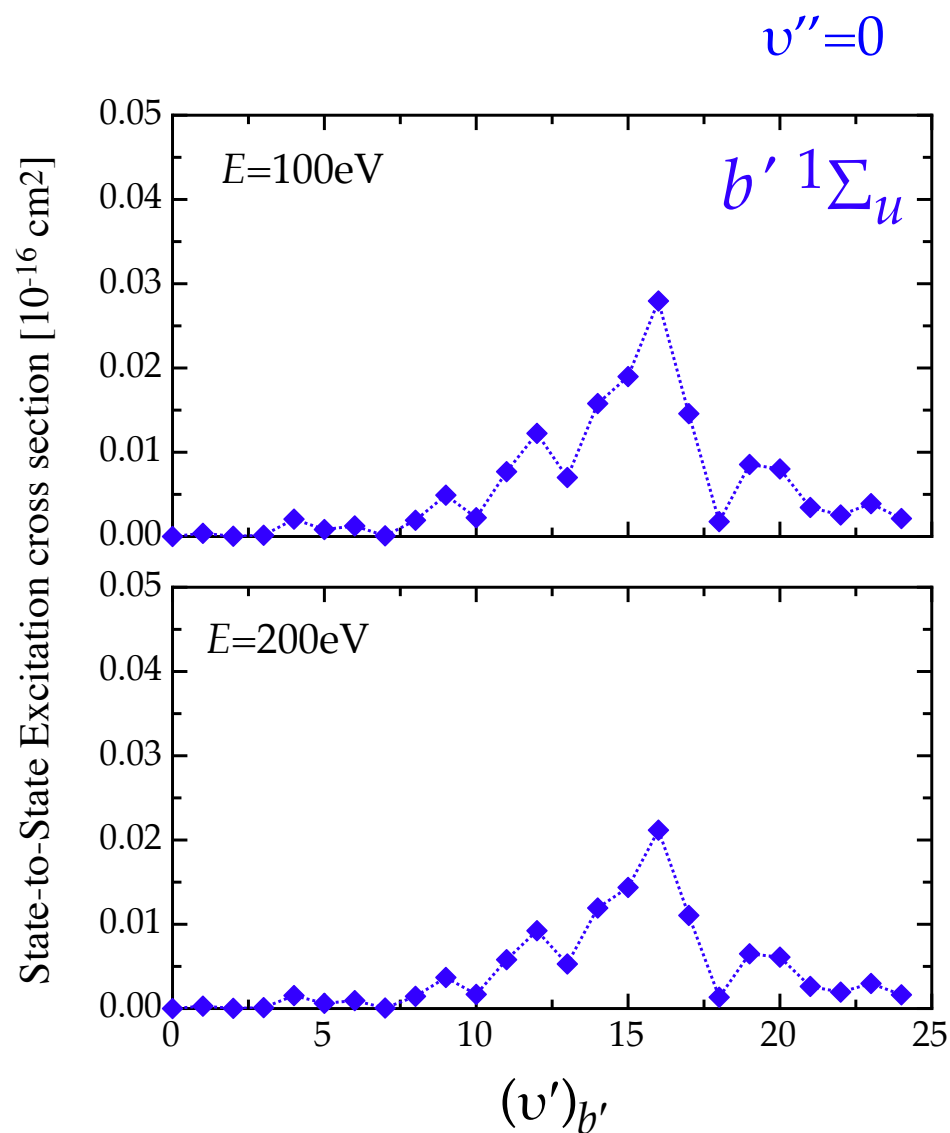
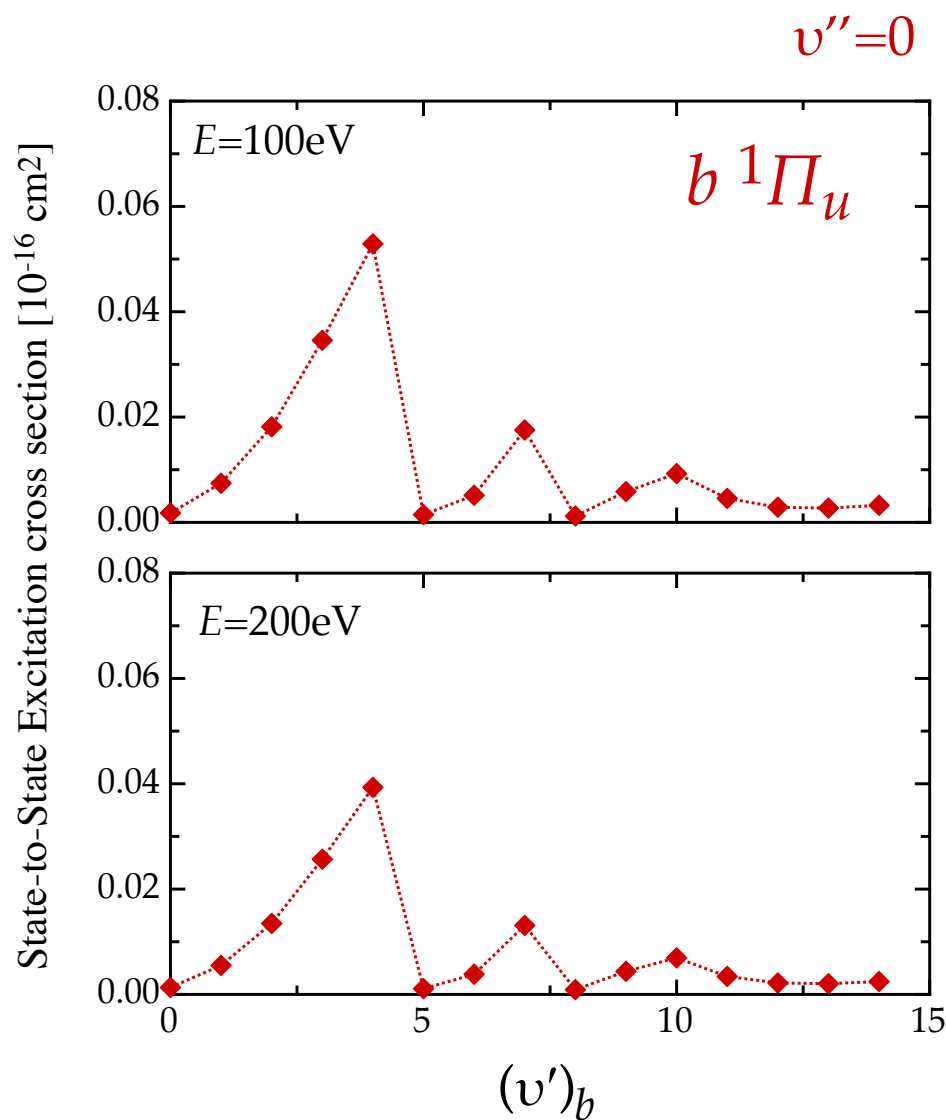
vibrational analysis $b'[v' 0-10]$ $c'[v' 0-3]$
or
 $b'[v' 0-16]$ $c'[v' 0-5]$

M.A. Khakoo et al., Phys.Rev.A 77 (2008) 012704

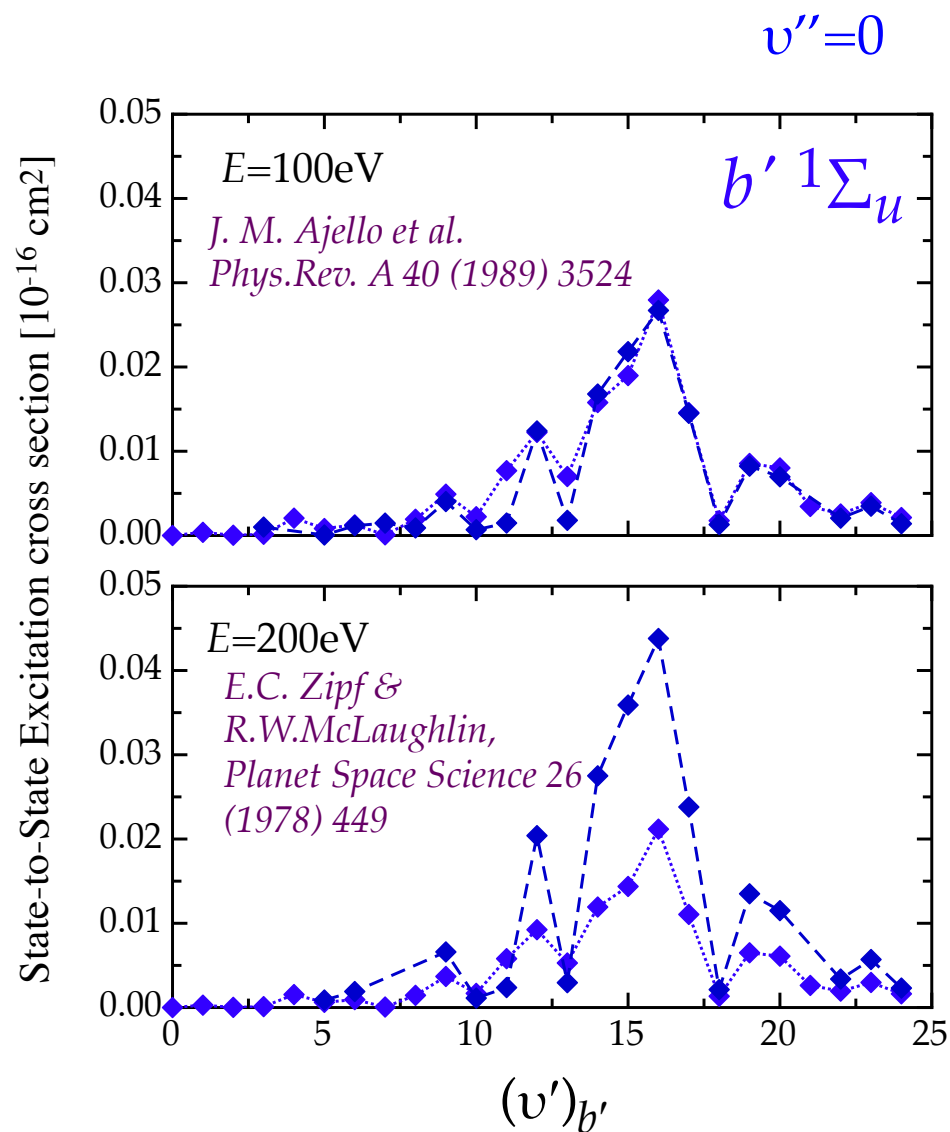
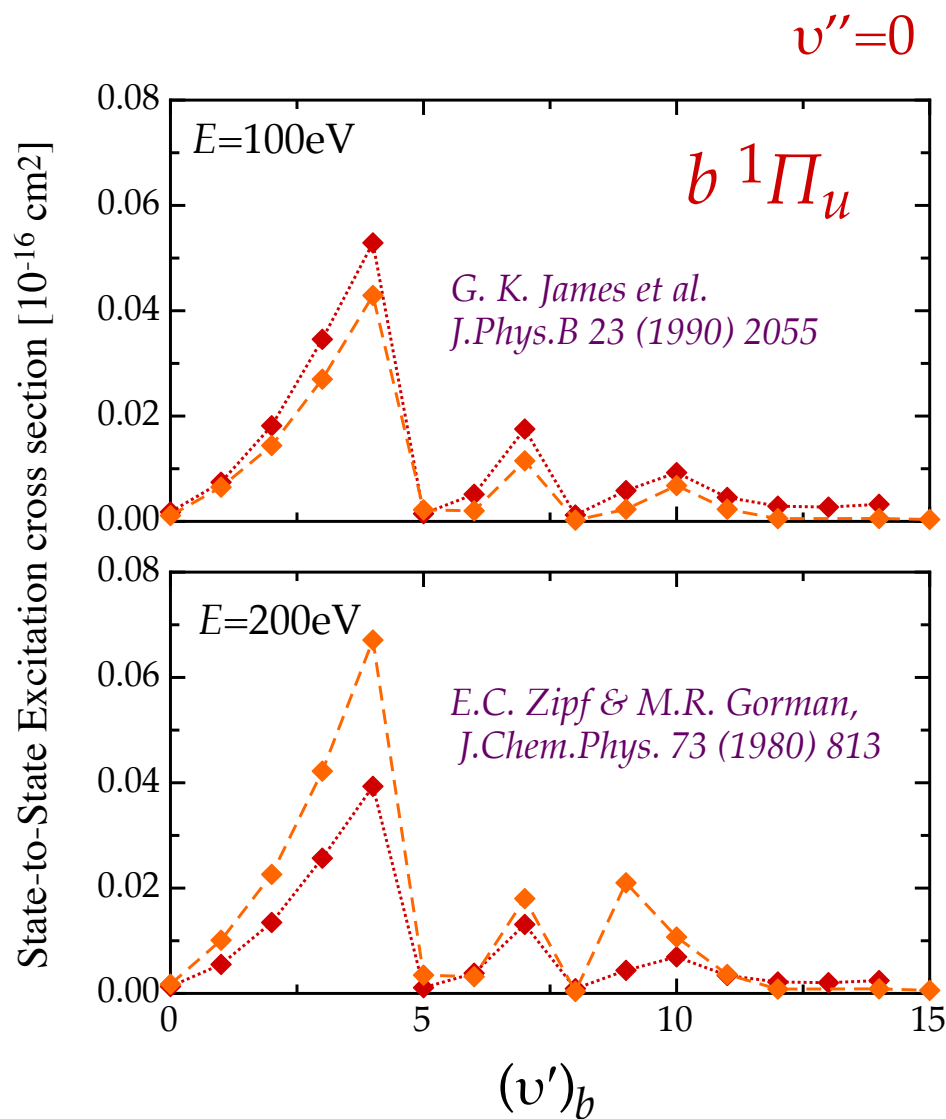
C.P. Malone et al., Phys.Rev.A 85 (2012) 062704

A. LARICCHIUTA The state-of-the-art description of vibrational-electronic excitations
MASTER CLASS 2014 Physikzentrum Bad Honnef (Germany), October 10th, 2014

State-to-State Vibronic Excitation Cross Sections: comparison with experiments

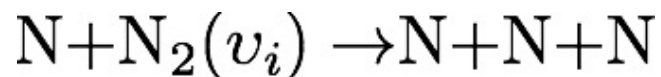


State-to-State Vibronic Excitation Cross Sections: comparison with experiments

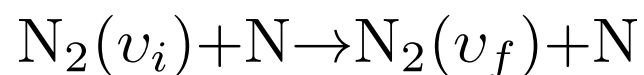


N-N₂ collisional-induced processes

PROCESSES



collisional-induced dissociation



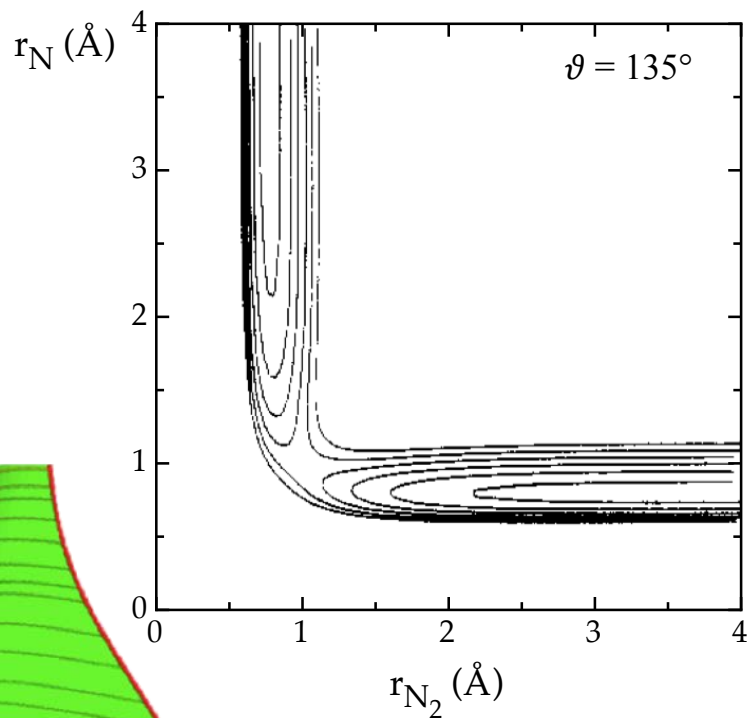
vibration-translation energy transfer

CALCULATION FEATURES

Theoretical approach	QCT method [Blais&Truhlar (1976)] [M. Pattengill "Rotational excitation III: Classical trajectory methods, in: Atom-molecule Collision Theory; a Guide for the Experimentalist" (1979) Plenum Press, NewYork and London]	translational energy range [0.001-10 eV] stratified sampling trajectory density per eV and per Angstrom of impact parameter: 96000 up to 0.5 eV 24000 up to 3.0 eV 6000 up to 10 eV
Potential energy surface of N ₃	LEPS PES [Laganà et al (1987)] and [Laganà & Garcia (1994)]	

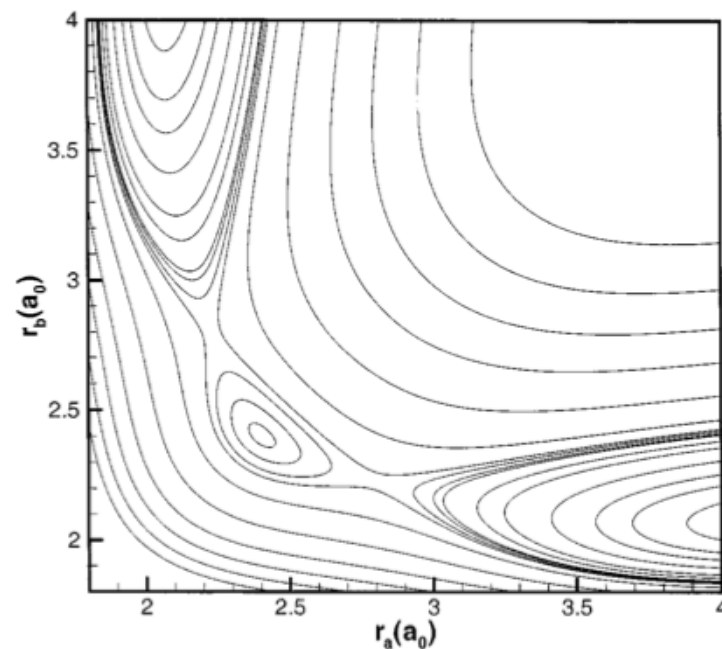
REFERENCES

- F. Esposito, M. Capitelli, Chemical Physics Letters 302 (1999) 49–54
F. Esposito, M. Capitelli, C. Gorse, Chemical Physics 257 (2000) 193–202
F. Esposito et al., Chem Physics (2006)
F. Esposito & M. Capitelli, Chem Phys Lett (2006)



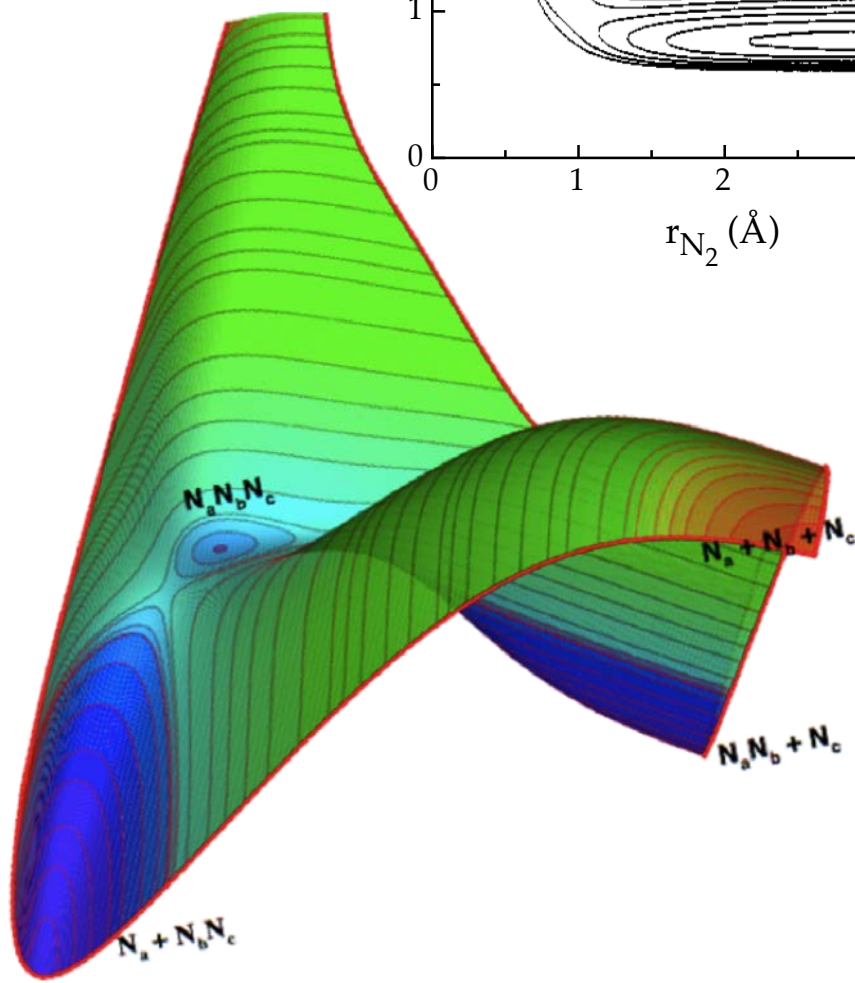
LEPS

A. Laganà, E. Garcia, L. Ciccarelli,
J. Phys. Chem. 91 (1987) 312-314



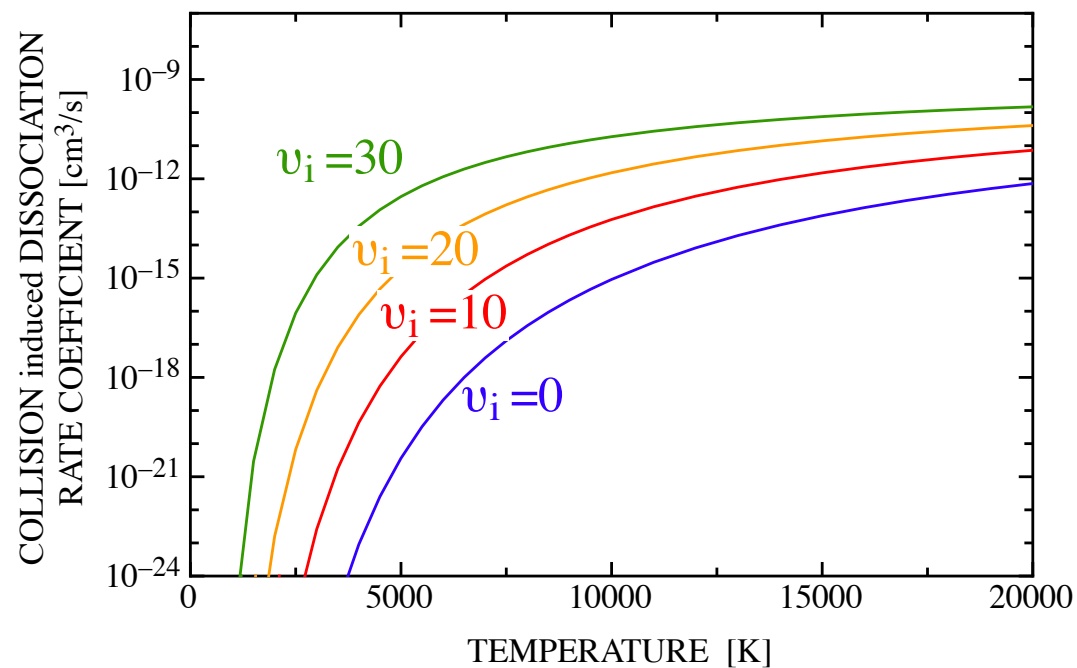
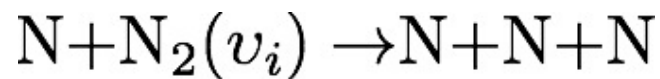
ab-initio

D. Wang, W.M. Huo, C.E. Dateo, D.W. Schwenke, J.R. Stallcop,
J. Chem. Phys. 120 (2004) 6041



P.J.S.B.Caridade, B.R.L. Galvao, A.J.C. Varandas
J. Phys. Chem. A 114 (2010) 6063–6070

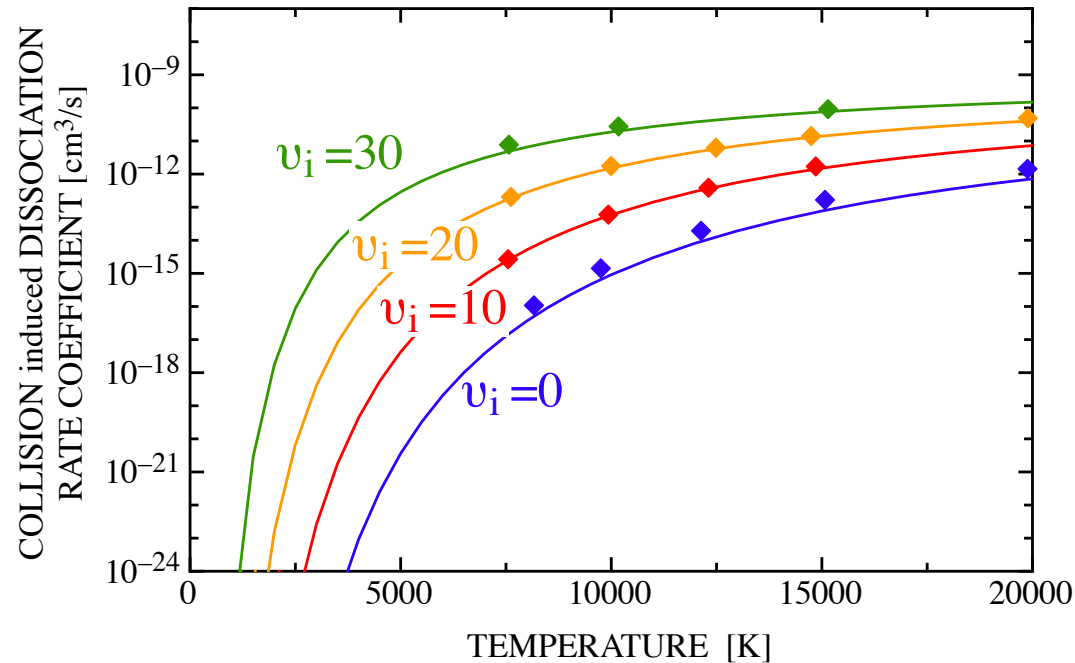
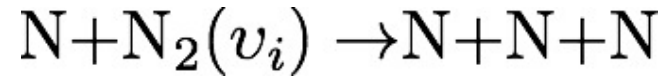
N-N₂ collisional-induced dissociation



F. Esposito & M. Capitelli, Chem Phys Lett (2006)

F. Esposito et al., Chem Physics (2006)

N-N₂ collisional-induced dissociation



R.L. Jaffe, D.W. Schwenke, and G. Chaban,
AIAA Paper 2010-4517 (2010)
Munafò et al, Eur. Phys. J. D (2012)

F. Esposito & M. Capitelli, Chem Phys Lett (2006)
F. Esposito et al., Chem Physics (2006)

N-N₂ collisional-induced dissociation: global dissociation rate coefficient

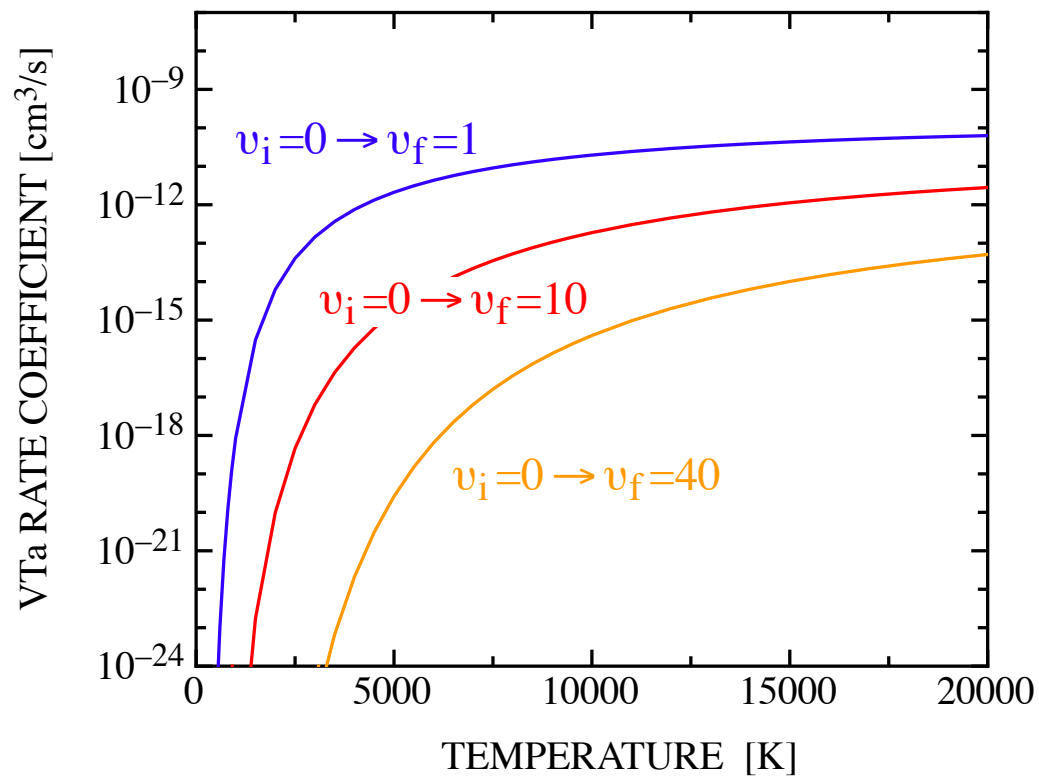
The screenshot shows the PHYS4ENTRY website interface. A modal window titled "Process Info" is open, displaying details for a specific chemical process. The background shows a search interface with filters for "Atmospheres" and "Class & Process".

Process Info

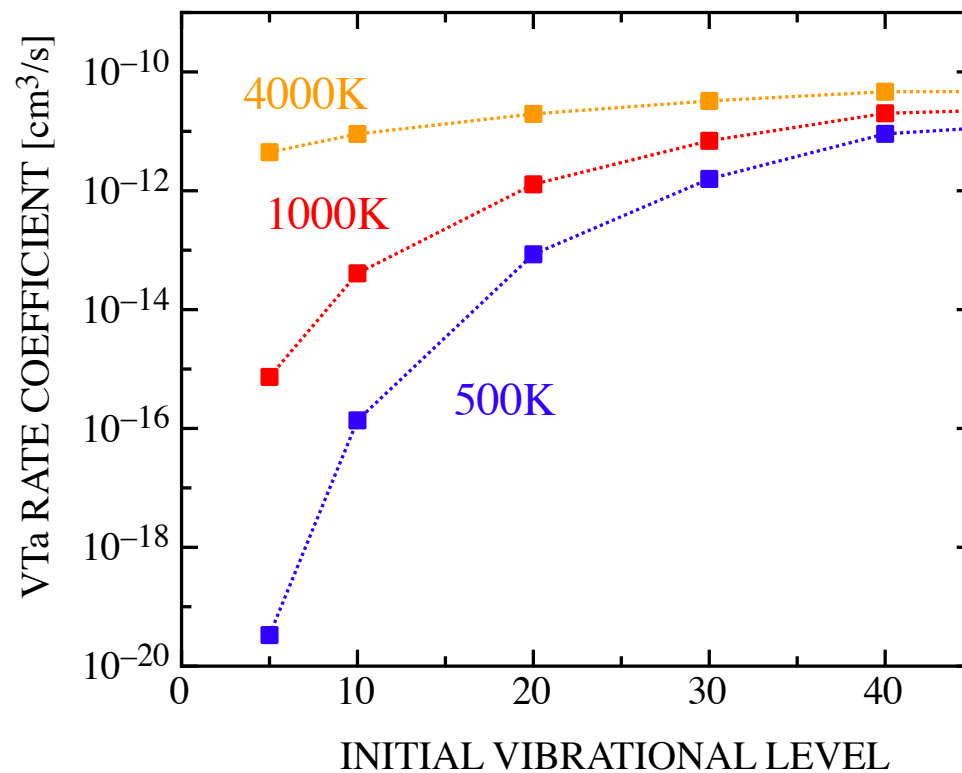
File type: **Rate**
Class: **heavy-particle processes**
Process Group: **atom-diatom**
Process: **dissociation**
Reagents and Products: **N(⁴S)+N₂(¹Σ_g⁺) → N(⁴S)+N(⁴S)+N(⁴S)**
Intermediate State: **N₃ (LEPS PES)**
Reference: **F. Esposito, M. Capitelli, Chemical Physics Letters 302 (1999) 49–54. F. Esposito, M. Capitelli, C. Gorse, Chemical Physics 257 (2000) 193–202. F. Esposito, I. Armenise, M. Capitelli, Chemical Physics 331 (2006) 1–8. I. Wysong, S. Gimelshein, N. Gimelshein, W. McKeon, F. Esposito, Physics of Fluids 24 (2012) 042002.**
Reference Link: <http://www.sciencedirect.com/science/article/pii/S0009261499000998>
Reference Link: <http://www.sciencedirect.com/science/article/pii/S0301010406005441>
Rate Constant: **cm³s⁻¹**
Temperature: **Kelvin**
Last Update: **2013-Jun-19 17:31:24 UTC**
More info: [click here](#)

VT energy transfer processes in N-N₂ collisions

$$v_i = 0 \rightarrow v_f$$

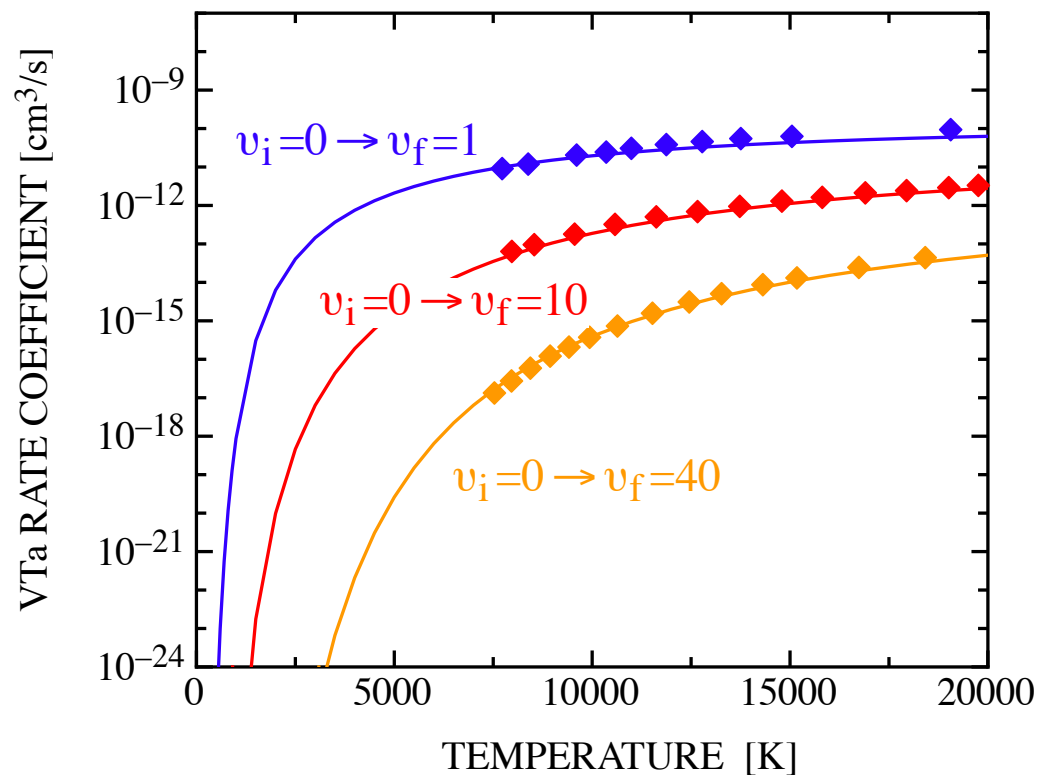


$$v_i \rightarrow v_f = v_i - 1$$



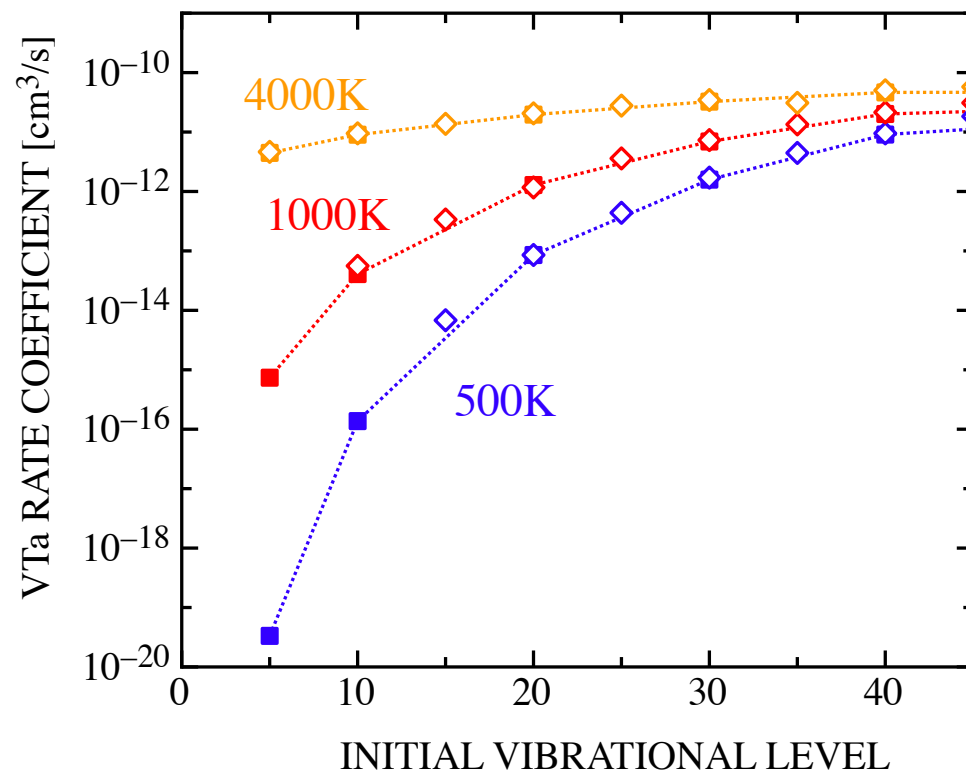
VT energy transfer processes in N-N₂ collisions

$$v_i = 0 \rightarrow v_f$$



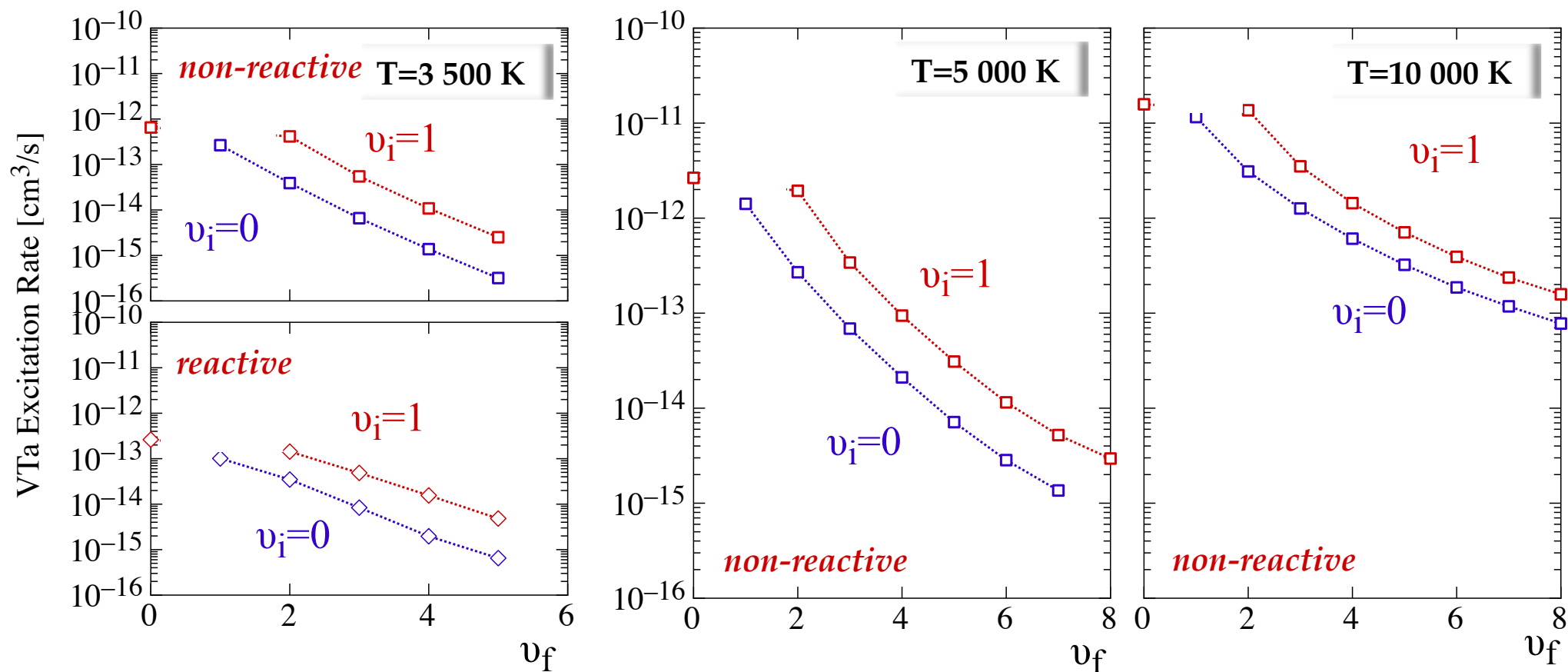
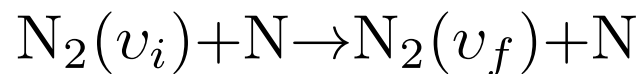
R.L. Jaffe, D.W. Schwenke, and G. Chaban,
AIAA Paper 2010-4517 (2010)
Munafò et al, Eur. Phys. J. D (2012)

$$v_i \rightarrow v_f = v_i - 1$$

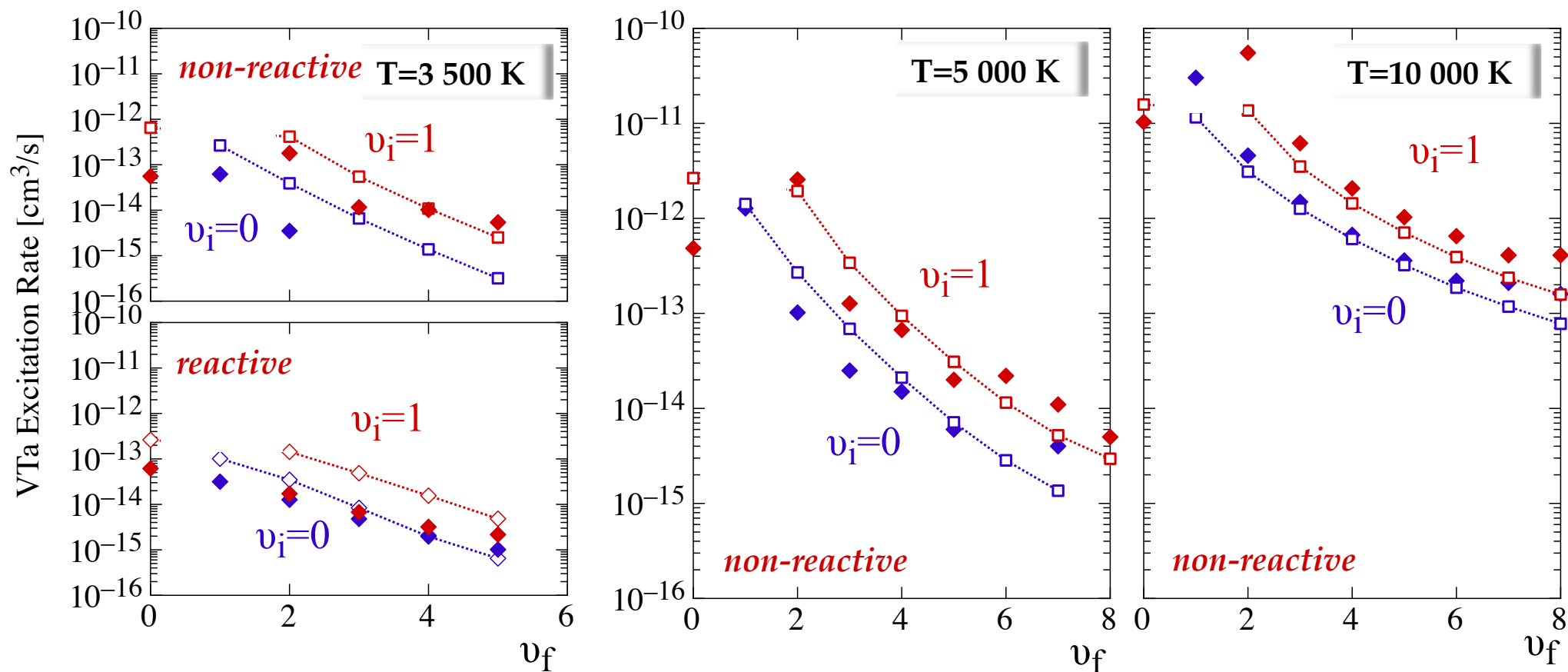
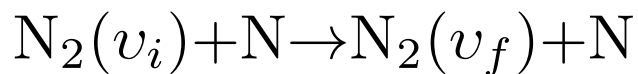


A. Laganà, G. Ochoa De Aspuru, E. Garcia,
Temperature dependence of quasiclassical and
quantum rate coefficients for N + N₂
Dip Chimica, Università di Perugia, Italy (1996)

VT energy transfer processes in N-N₂ collisions

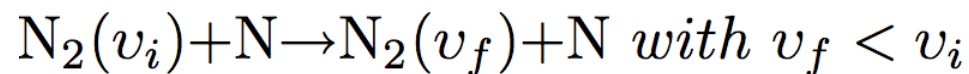


VT energy transfer processes in N-N₂ collisions

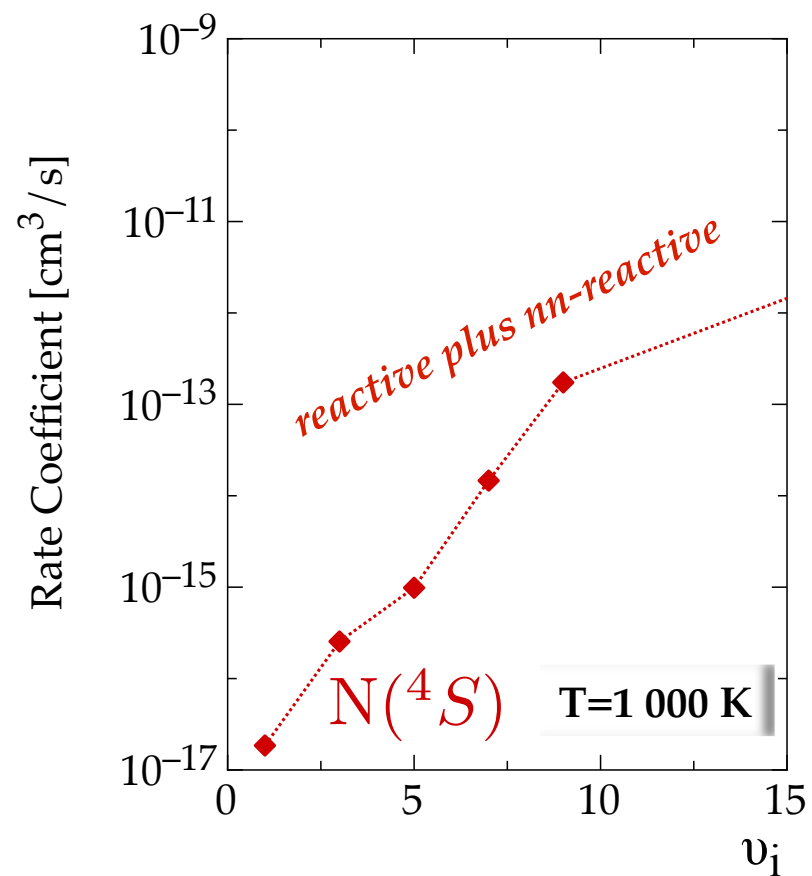
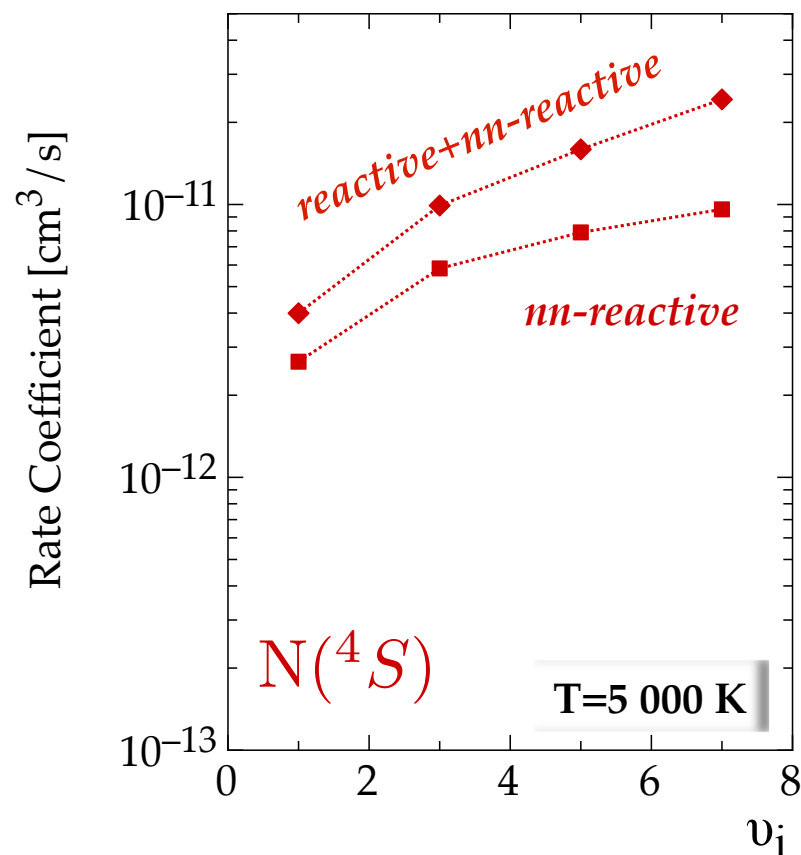


P.J.S.B.Caridade, B.R.L. Galvao, A.J.C. Varandas
J. Phys. Chem. A 114 (2010) 6063–6070

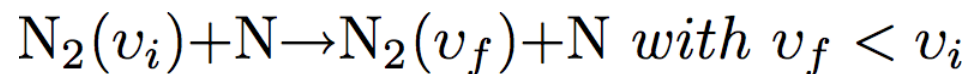
VT energy transfer processes in N-N₂ collisions



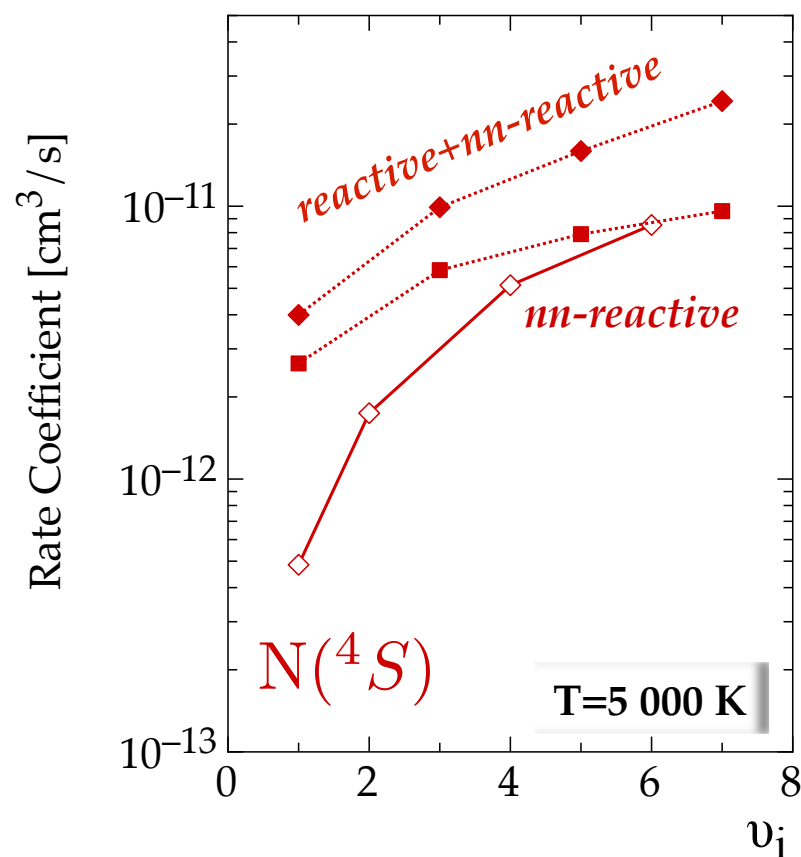
$$k^{\text{VT}}(v_i) = \sum_{v_f=0}^{v_i-1} k^{\text{VT}}(v_i, v_f)$$



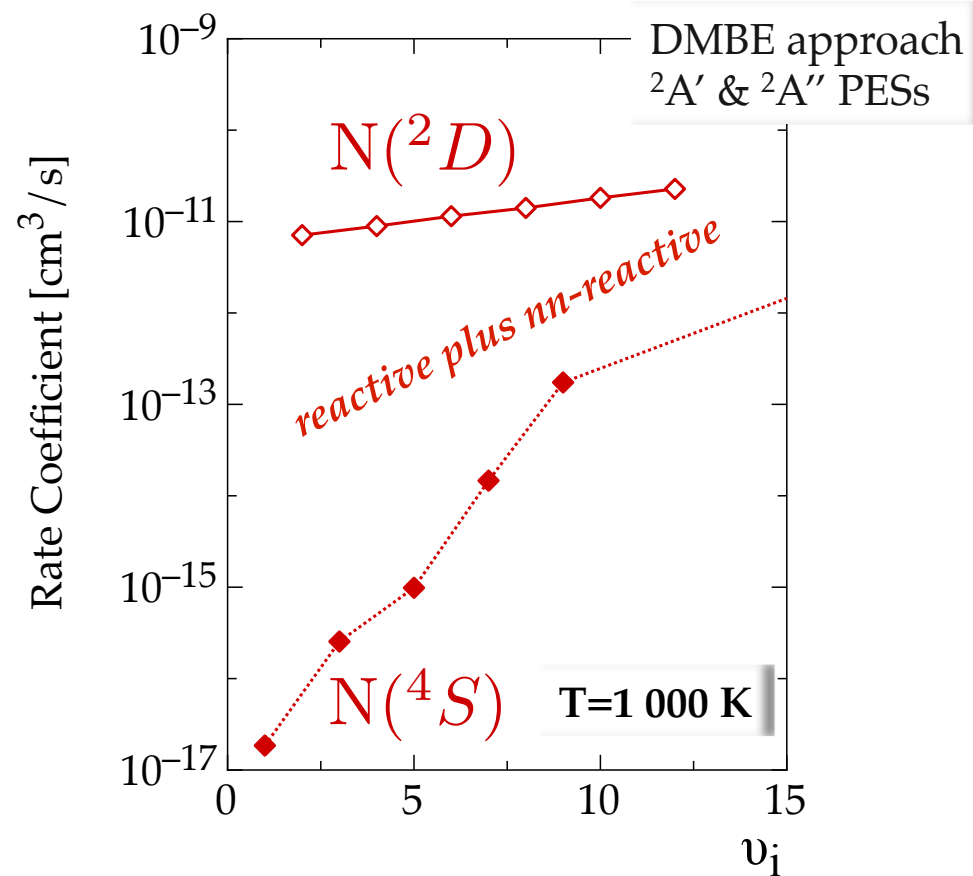
VT energy transfer processes in N-N₂ collisions



$$k^{\text{VT}}(v_i) = \sum_{v_f=0}^{v_i-1} k^{\text{VT}}(v_i, v_f)$$



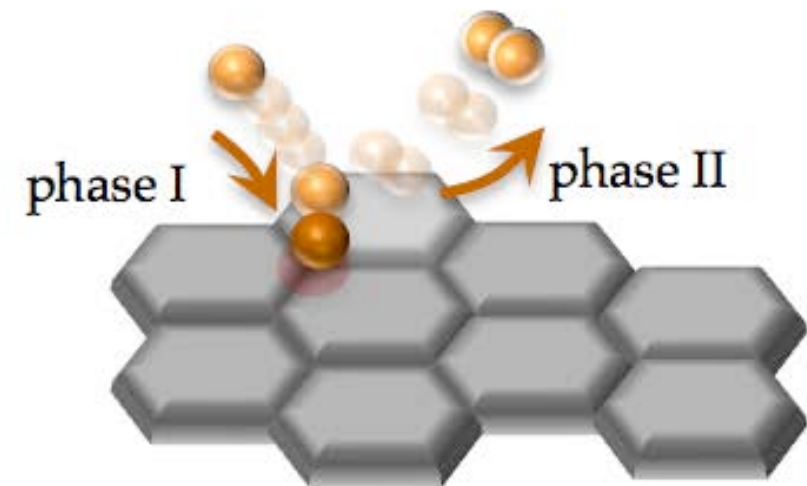
P.J.S.B.Caridade, B.R.L. Galvao, A.J.C. Varandas
J. Phys. Chem. A 114 (2010) 6063–6070



B.R.L. Galvão, A.J.C. Varandas, J.P. Braga, J.C. Belchior
Chemical Physics Letters 577 (2013) 27–31

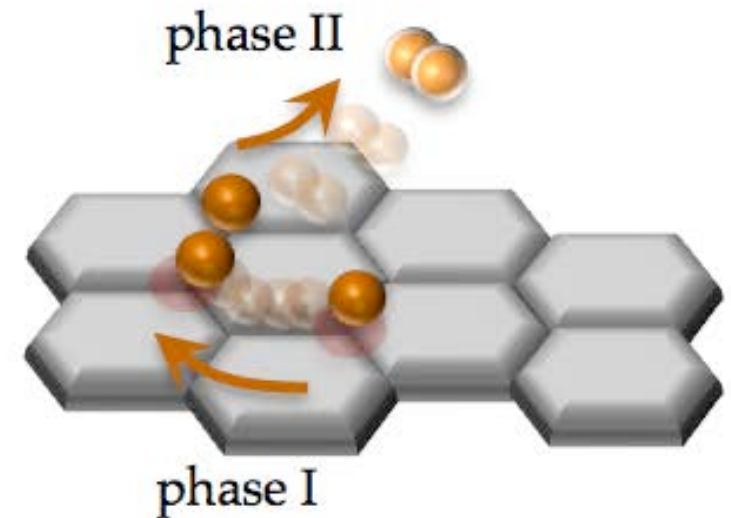
heterogeneous recombination at surface

Eley-Rideal mechanism



hot-atom
mechanism

Langmuir-Hinshelwood mechanism



heterogeneous recombination at surface

The screenshot shows the PHYS4ENTRY website interface. A pop-up window titled "Process Info" is displayed over the main content. The background shows a sidebar with "Query" and "Class & Process" sections, and a main area with a search bar and a table of results.

Process Info

File type: **Probabilities**
Class: **gas/surface processes**
Process Group: **diatom@surface-processes**
Process: **Eley-Rideal mechanism**
Reagents and Products: **O(³P)@beta-cristobalite (SiO₂)+O(³P) → beta-cristobalite (SiO₂)+O₂(X³Σ_g⁻)**
Intermediate State: **None**
Reference: **M. Rutigliano, C. Zazza, S.Orlandini, N. Sanna, V. Barone, and M. Cacciatore Oxygen Atoms and Molecules on Silica Surfaces: collisional data relevant to aerospace in : 'Catalytic Gas-Surface Interactions, NATO Applied Vehicle Technology Panel AVT-199', VKI, Rhode St Genèse (Belgique), 22-24 Oct. 2012, STO-MP-AVT199 5-1- 5-28 (2012)**
Probability: **ADM**
Energy : **eV**
Last Update: **2013-Apr-30 16:33:00 UTC**
More info: [click here](#)

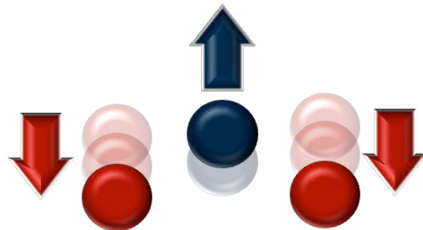
CO₂ vibrational kinetics

symmetric stretch



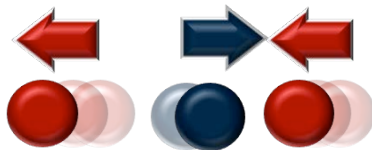
ν_1

bending



ν_2

asymmetric stretch

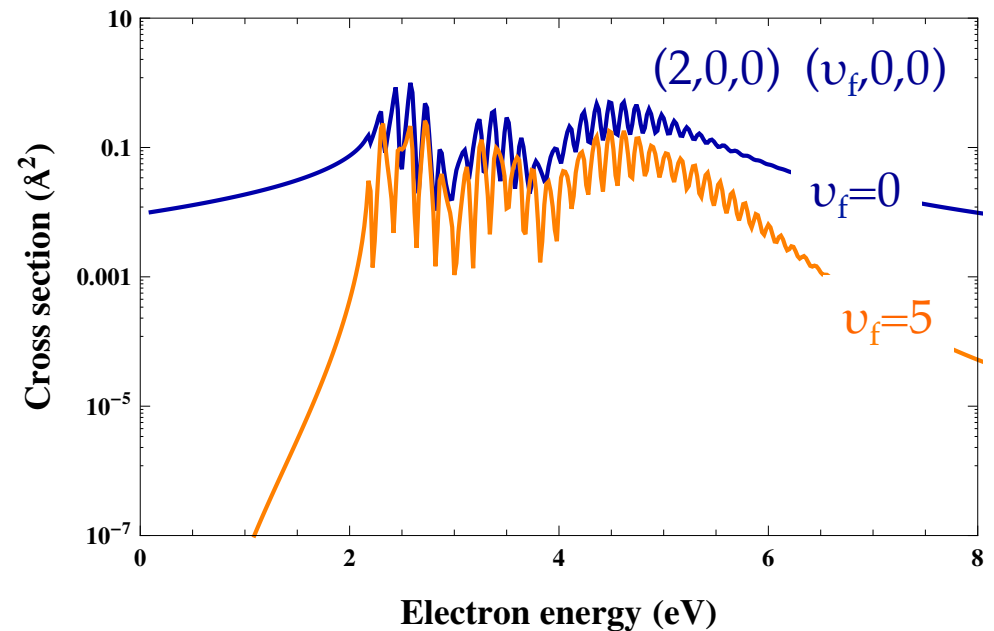
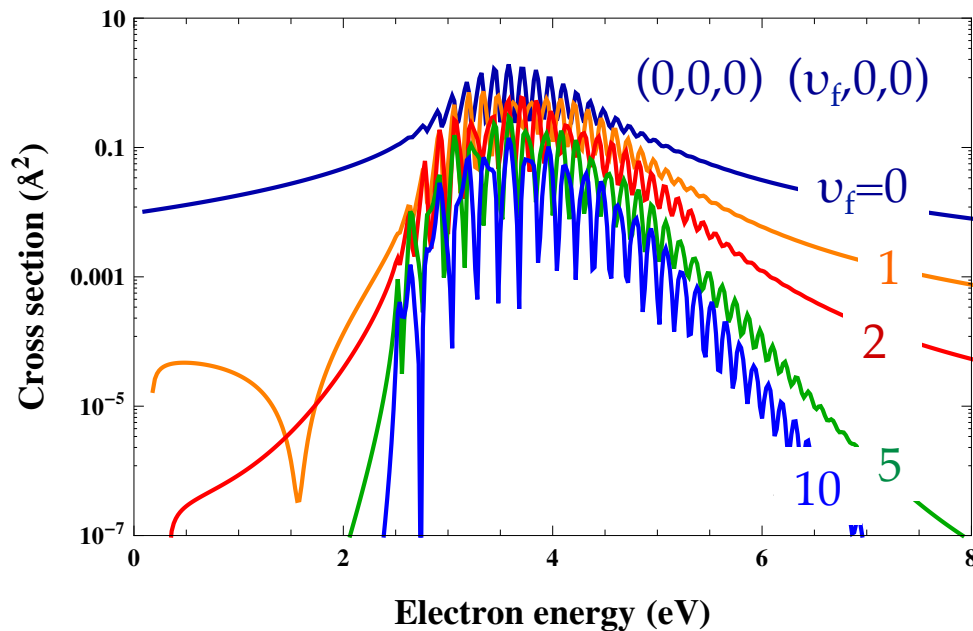
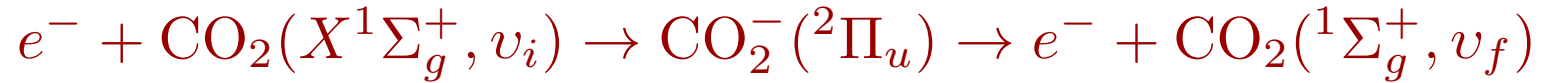


ν_3

Armenise & Kustova,
Chemical Physics (2013) and (2014)

Tomáš Kozák and Annemie Bogaerts,
Plasma Sources Science & Technology (2014)

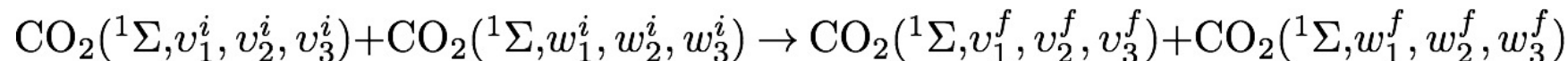
Resonant vibrational excitation in e -CO₂ collisions



R. Celiberto, V. Laporta, A. Laricchiuta, J. Tennyson and J.M. Wadehra,
The Open Plasma Physics Journal, 2014, 7, (Suppl 1: M2) 33-47

energy transfer processes in CO₂-CO₂ collisions

PROCESSES



CALCULATION FEATURES

Theoretical approach	QCT method [Blais&Truhlar (1976)] [M. Pattengill "Rotational excitation III: Classical trajectory methods, in: Atom-molecule Collision Theory; a Guide for the Experimentalist" (1979) Plenum Press, New York and London]
Potential energy surface	intra-molecule potential [J.S. Carter and N. Murrell, Croat. Chem. Acta 57 (1984) 355] inter-molecule potential [Bartolomei et al (1976)]
Rotational levels	a Boltzmann distribution is assumed for rotational levels at the translational temperature

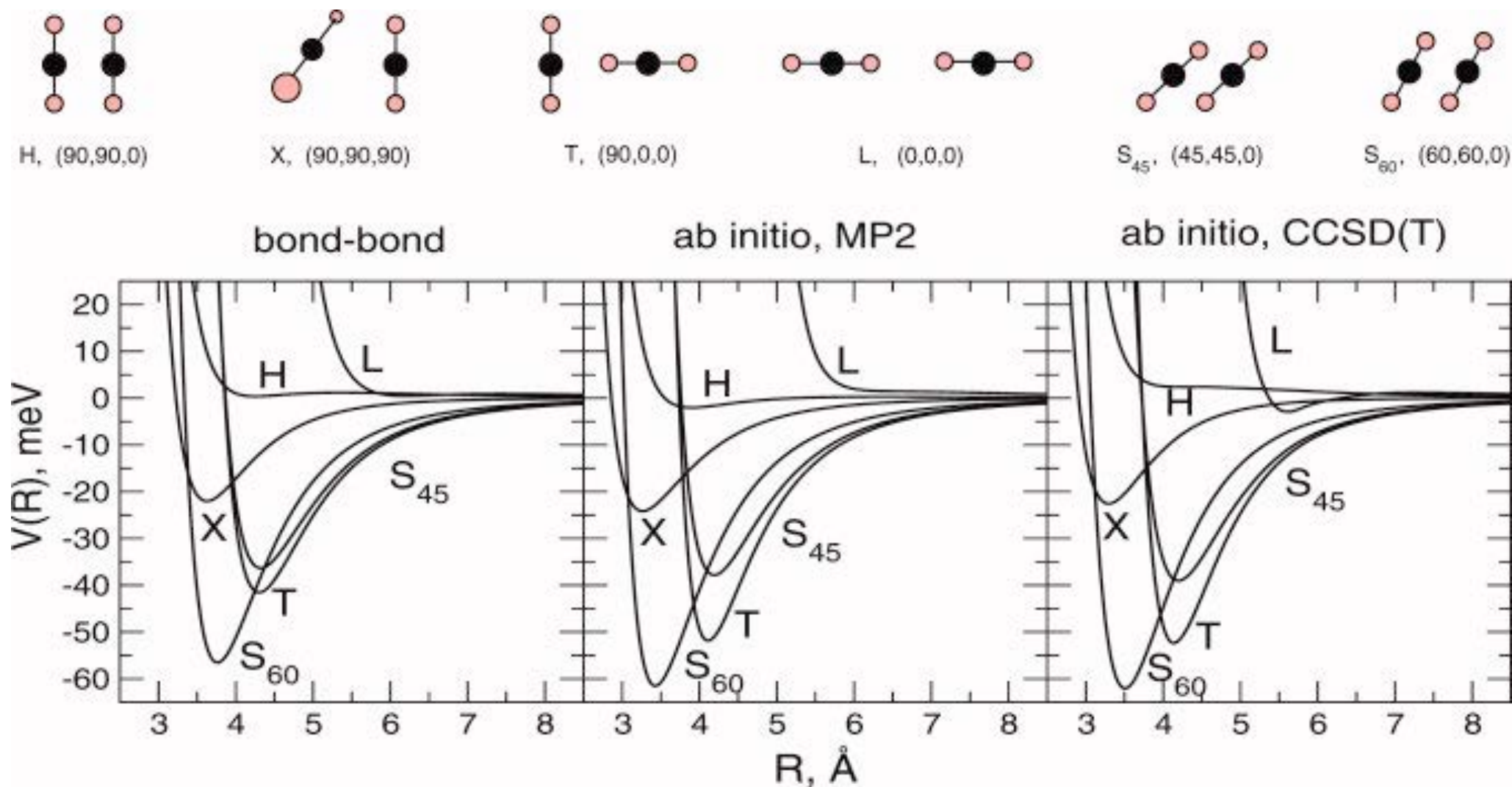
REFERENCES

M. Bartolomei, F. Pirani, A. Laganà, A. Lombardi, J. Comp. Chem. 33 (2012) 1806

A. Lombardi, N. Faginas Lago, A. Laganà, F. Pirani, S. Falcinelli, S.

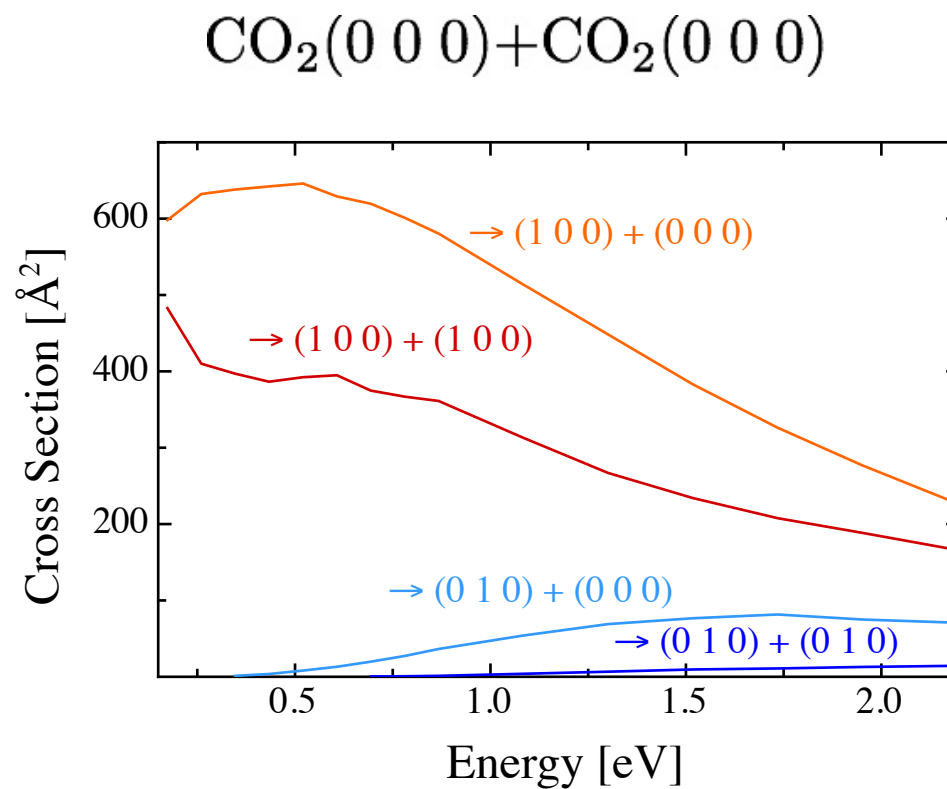
in: Murgante, B., Gervasi, O., Misra, S., Nedjah, N., Rocha, A.M.A.C., Tanir, D., Apduhan, B.O. (eds.) ICCSA 2012, Part I. LNCS, vol. 7333, pp. 387–400. Springer, Heidelberg (2012)

energy transfer processes in CO₂-CO₂ collisions

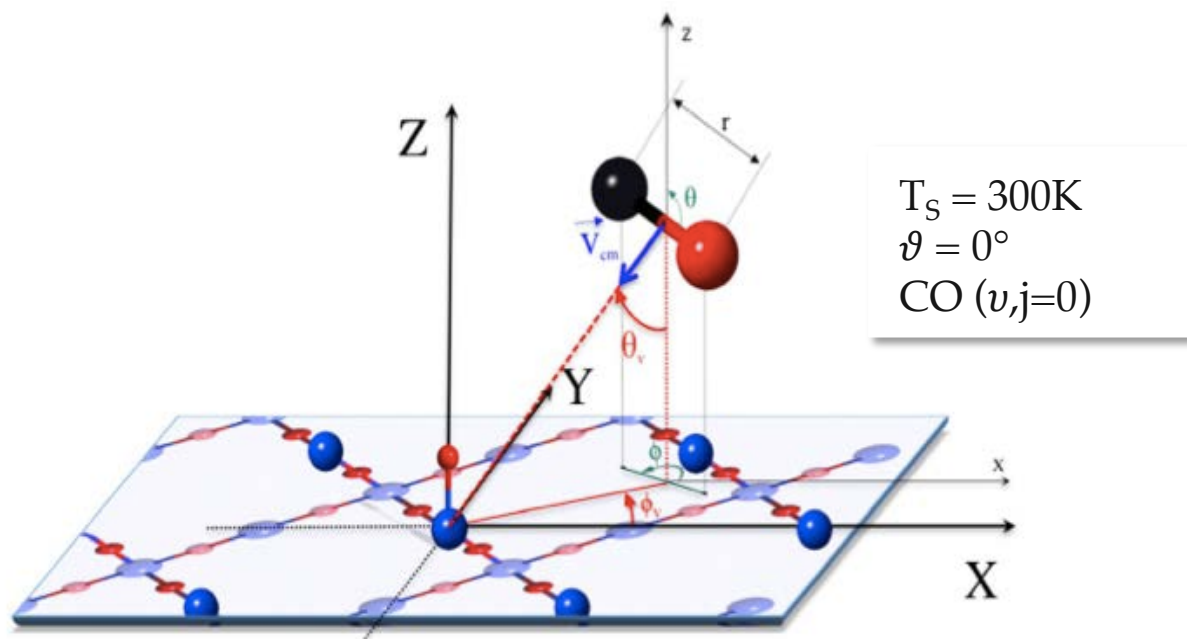


rigid-monomers ➡ stretched&bent monomers

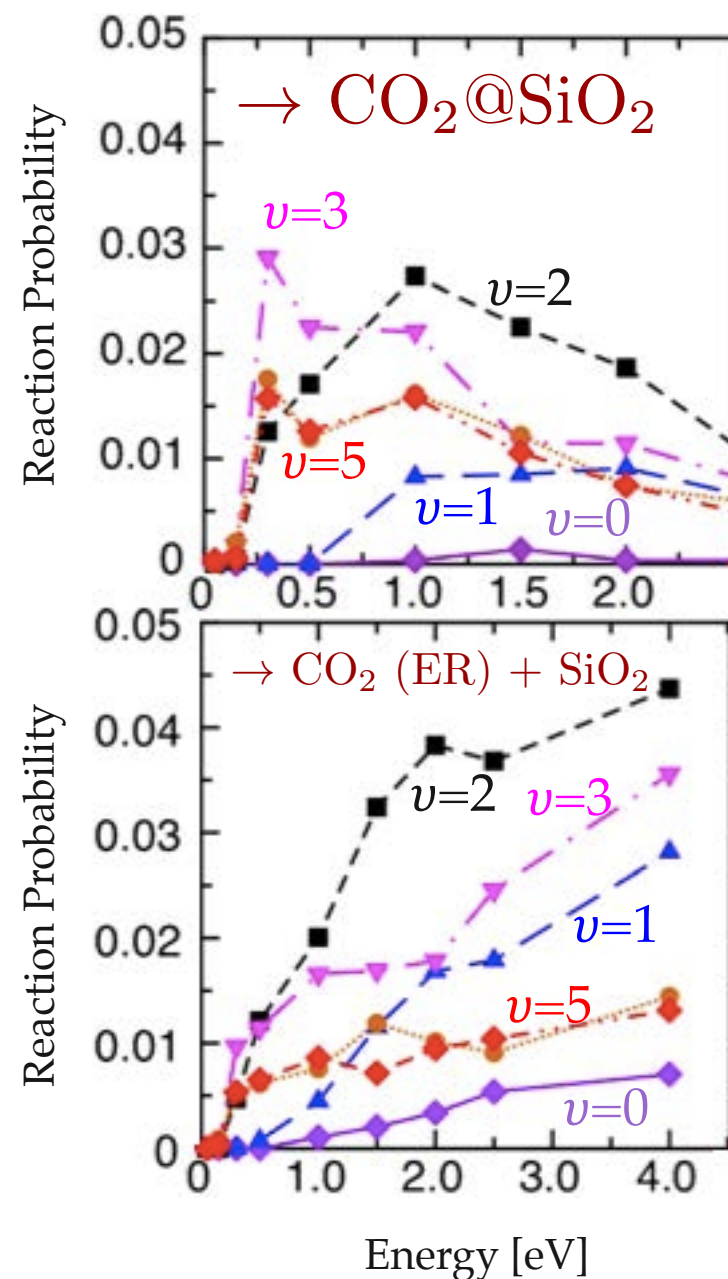
energy transfer processes in CO₂-CO₂ collisions



heterogeneous recombination of CO₂ at silica surface



REFLECTED CO



Pablo Gamallo, Hèctor Prats and Ramón Sayós
 Journal of Molecular Modeling 20 (2014) 2160

A. LARICCHIUTA *The state-of-the-art description of vibrational-electronic excitations*
 MASTER CLASS 2014 Physikzentrum Bad Honnef (Germany), October 10th, 2014

A clear research imperative in the next decade will therefore be to increase our knowledge of the chemical and physical interactions in such plasmas of electrons, ions and radicals with neutral species

N. Mason “The 2012 Plasma Roadmap” J. Phys. D: Appl. Phys. 45 (2012) 253001
citing the report of the Washington: National Research Council (1996)
Database Needs for Modeling and Simulation of Plasma Processing